



Complex distillation systems. Theory and models.

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INGAR
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Outline

1.- Introduction.

2.- Theory in simple columns design.

3.- Reversible distillation columns and sequences.

4.- Optimal synthesis distillation columns sequences.

1.- Introduction.

Mathematical models in Distillation \Rightarrow predict values:

Product composition for F (x_F)

Minimum energy demand R_{min}

Minimum number of stages N_{min}

Relationship R/R_{min} vs N/N_{min}

Mathematical models for:

Process optimization, design, synthesis.

Process control.

Process fault diagnosis.

Interest in Distillation grows because of:

Energy intensive process: Petrochemical, Biofuels.

New processes.

Complex sequences: energy intensive.

Reactive distillation: equipment intensive.

Reactive-extractive distillation.

High improvement potentials in distillation processes.

1.- Introduction.

Different mathematical models according to their sizes:

Aggregated (reduced): minimal information. Simple analytical formulae.

Short cuts: few equations but require numerical solution.

Rigorous: conservation laws, thermodynamics and constitutive models.

Trade off: size and complexity vs. information.

Limitations:

Rigorous models \Rightarrow general purpose.

Simple and reduced models \Rightarrow especial cases.

1.- Introduction

Fundamental concepts in distillation

Models components:

Mass balances

V-L equilibrium

Energy balances

Very exact descriptions using rigorous models for distillations

Concepts derived from distillation theory:

Pinch points

Residue curves

Distillation points

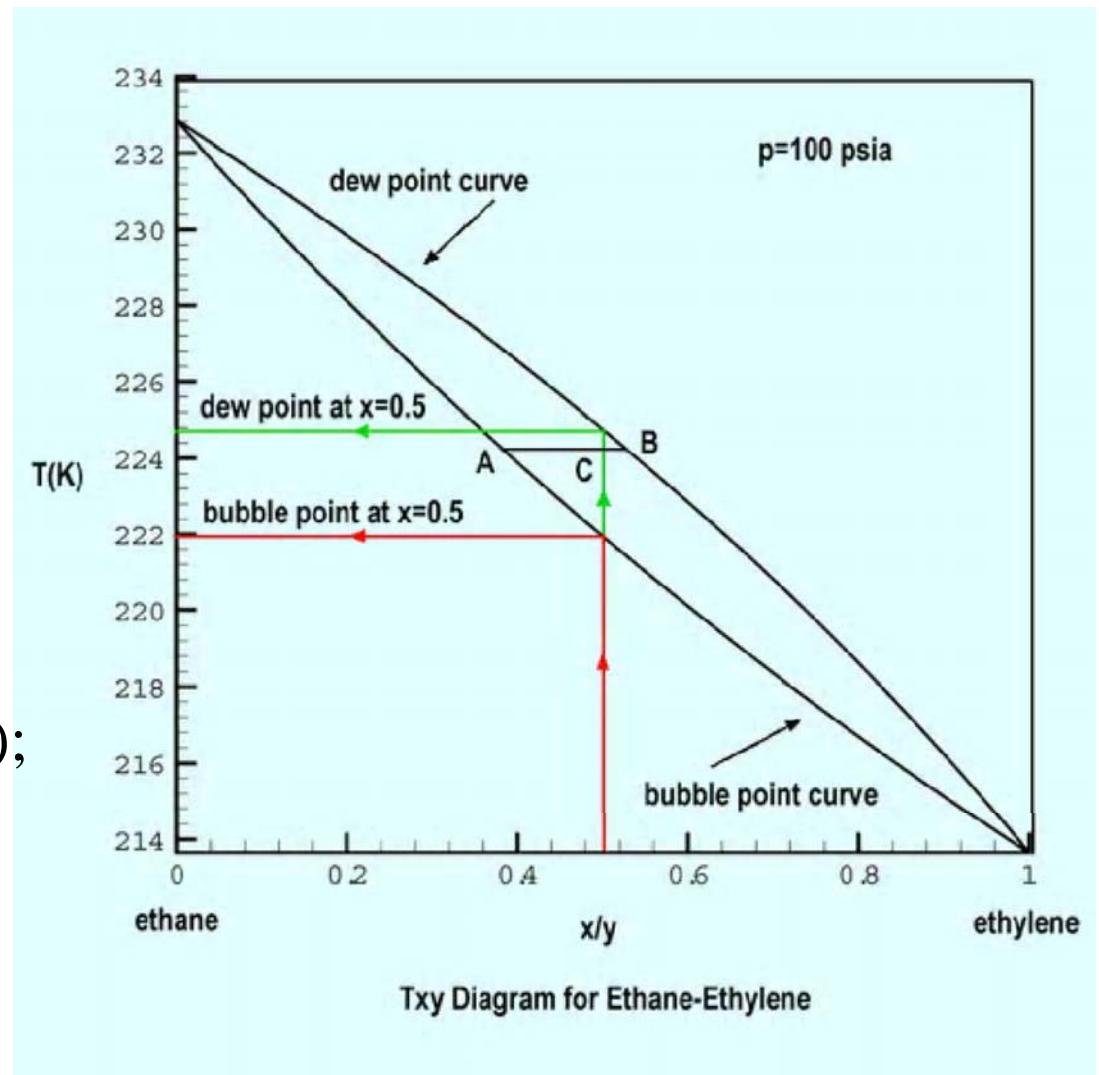
2.- Theory in simple columns design.

Mathematical models.
 Minimum energy demand.
 Minimum number of stages.

Thermodynamic aspects
 V-L Equilibrium
 Ideal Mixtures
 T vs $X(Y)$, $P=cte$. diagram

$$K_i = K_i(x, y, T, P) = P_i(T); \\ i = 1, \dots, NC$$

$$y_i = K_i(T) x_i$$



2.- Theory in simple columns design.

Simple columns. Mathematical models.

Minimum energy demand.

Minimum number of stages.

Assumptions:

Constant relative volatility and

Constant difference in vaporization enthalpy

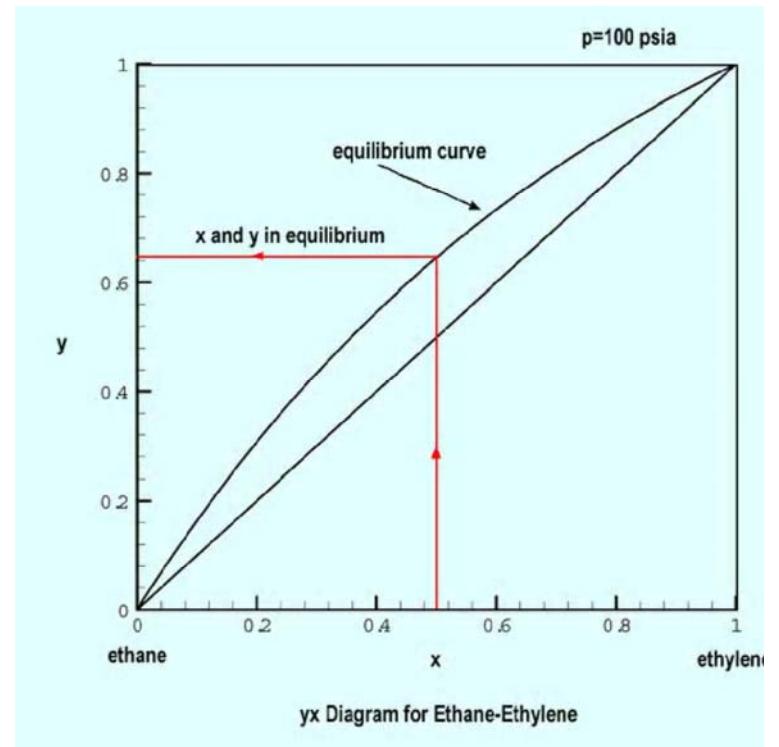


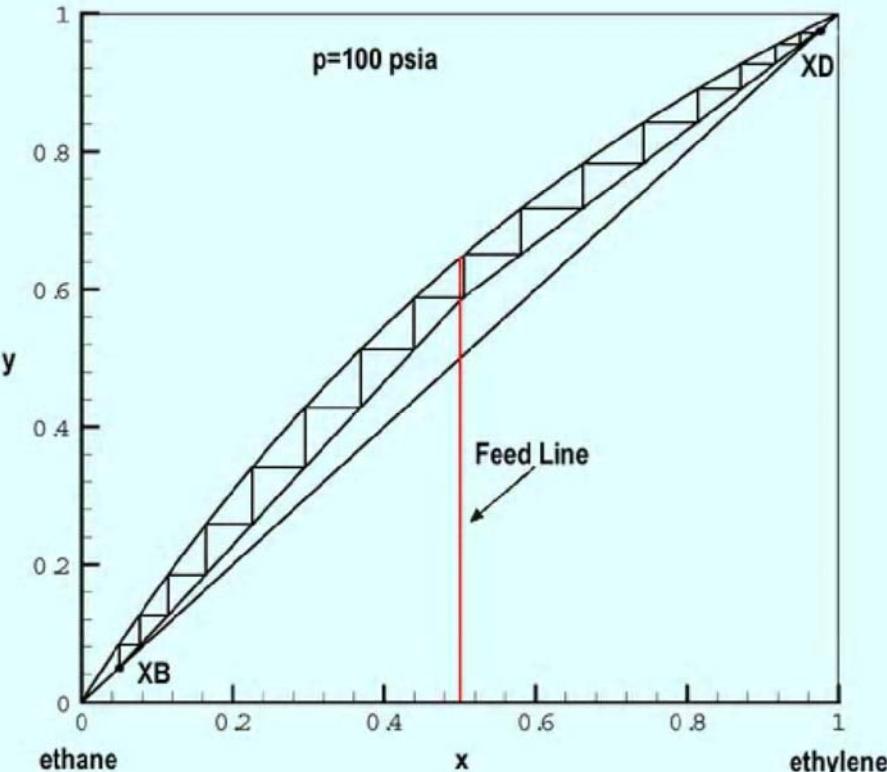
Constant Molar Overflow (CMO)



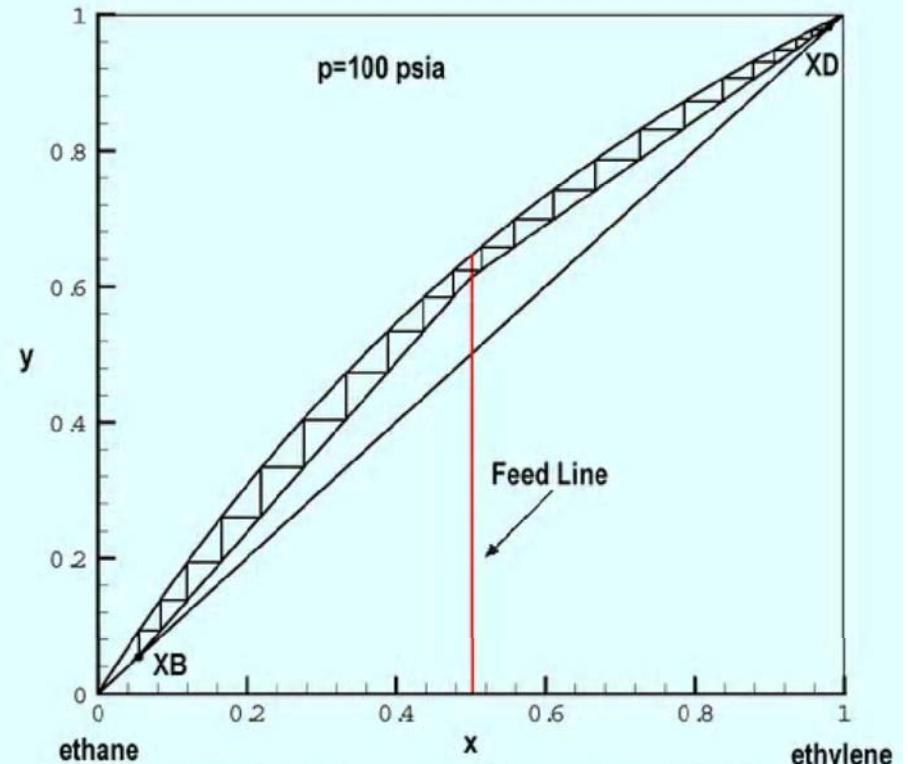
Straight operating line, decoupling mass and energy balances

Y vs X diagram





McCabe Thiele Diagram for Ethane-Ethylene ($r=3.9$)



McCabe-Thiele Diagram for Ethane-Ethylene ($r=3.0$)

Computing minimum energy demand.

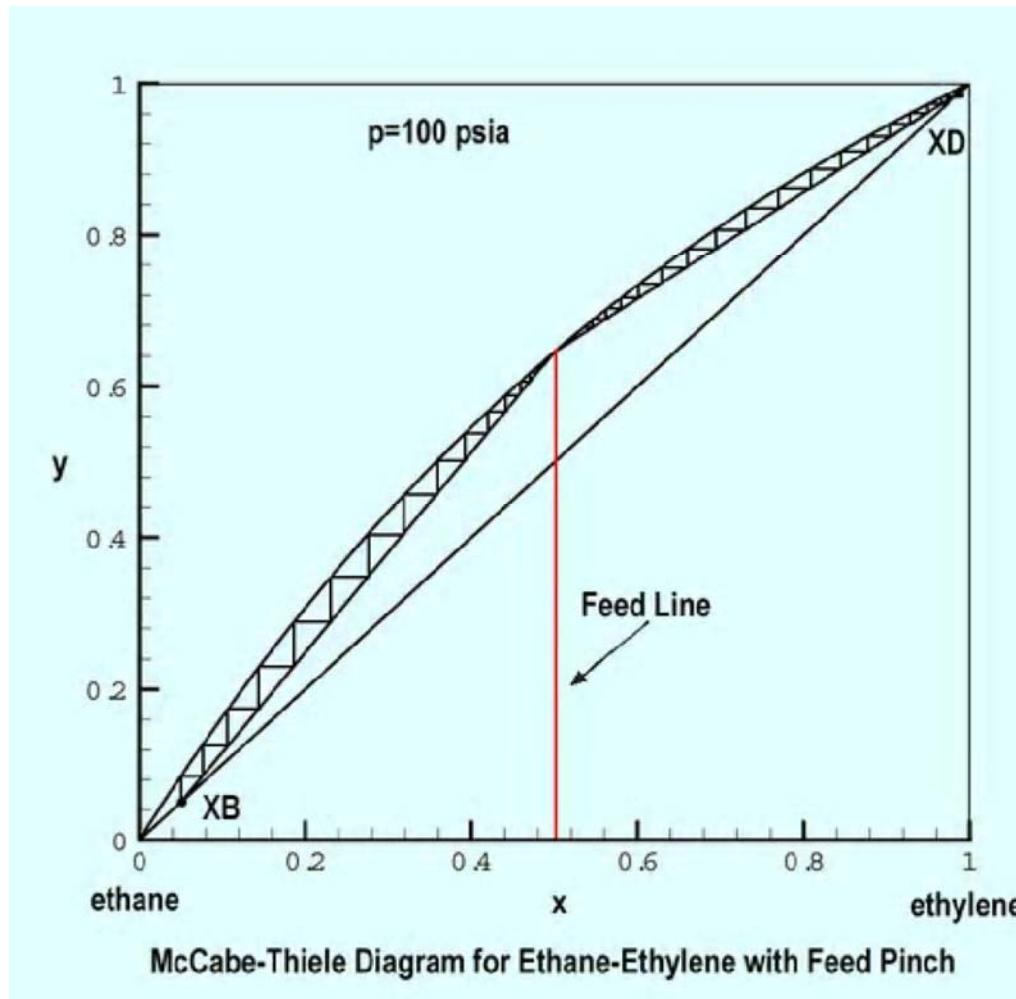
Constant molar overflow \Rightarrow Operating points belong
to Straight lines

Tray by tray calculations, equilibrium stages

Minimum energy demand.

Pinch at Feed tray.

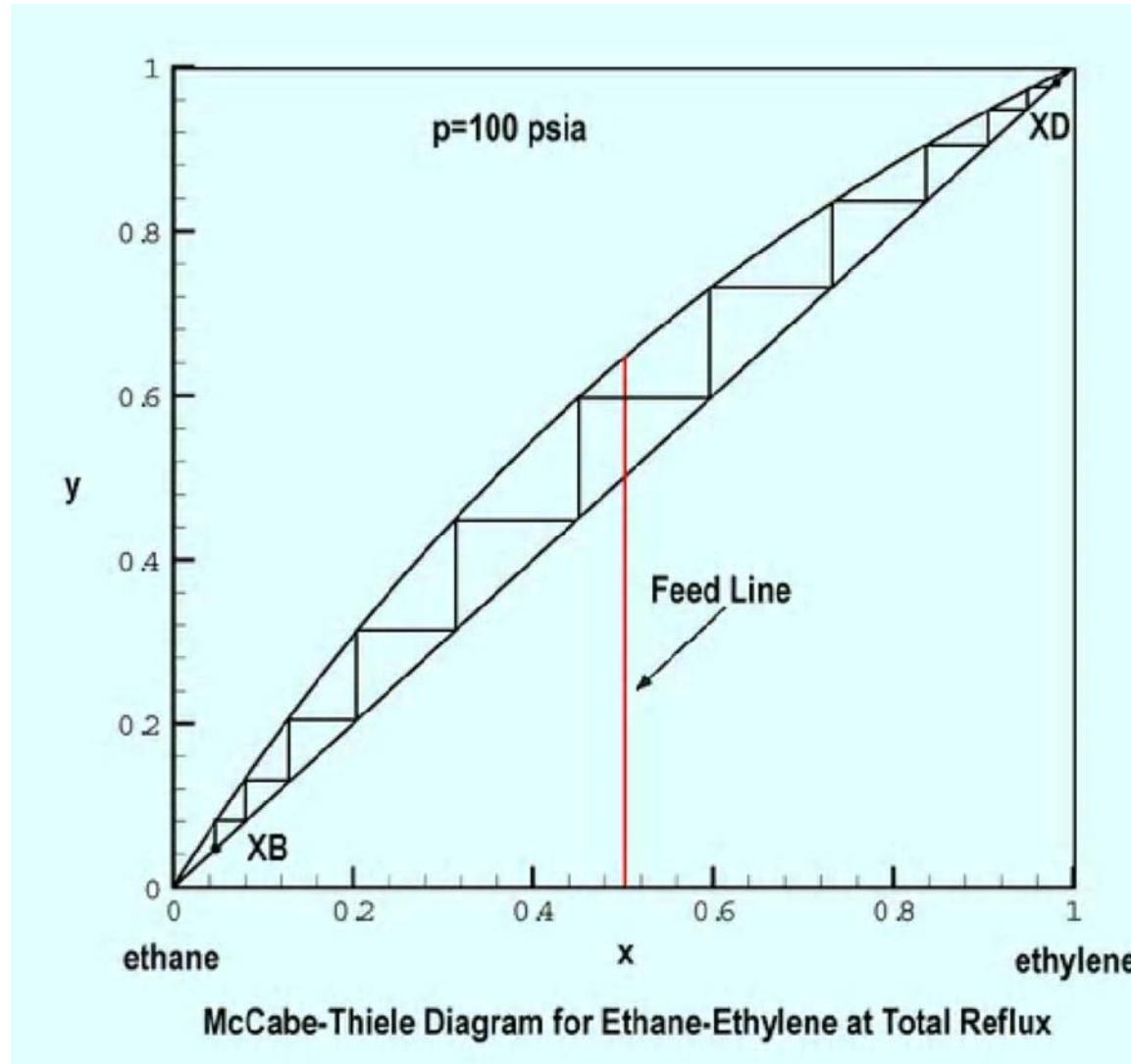
The number of stages increases
at the feed location.



Minimum number of stages.

Total reflux = Distillation points

$X_k \Rightarrow VLE \Rightarrow Y_k^* = X_{k+1} \Rightarrow VLE \Rightarrow Y_{k+1}^* = X_{k+2} \dots$



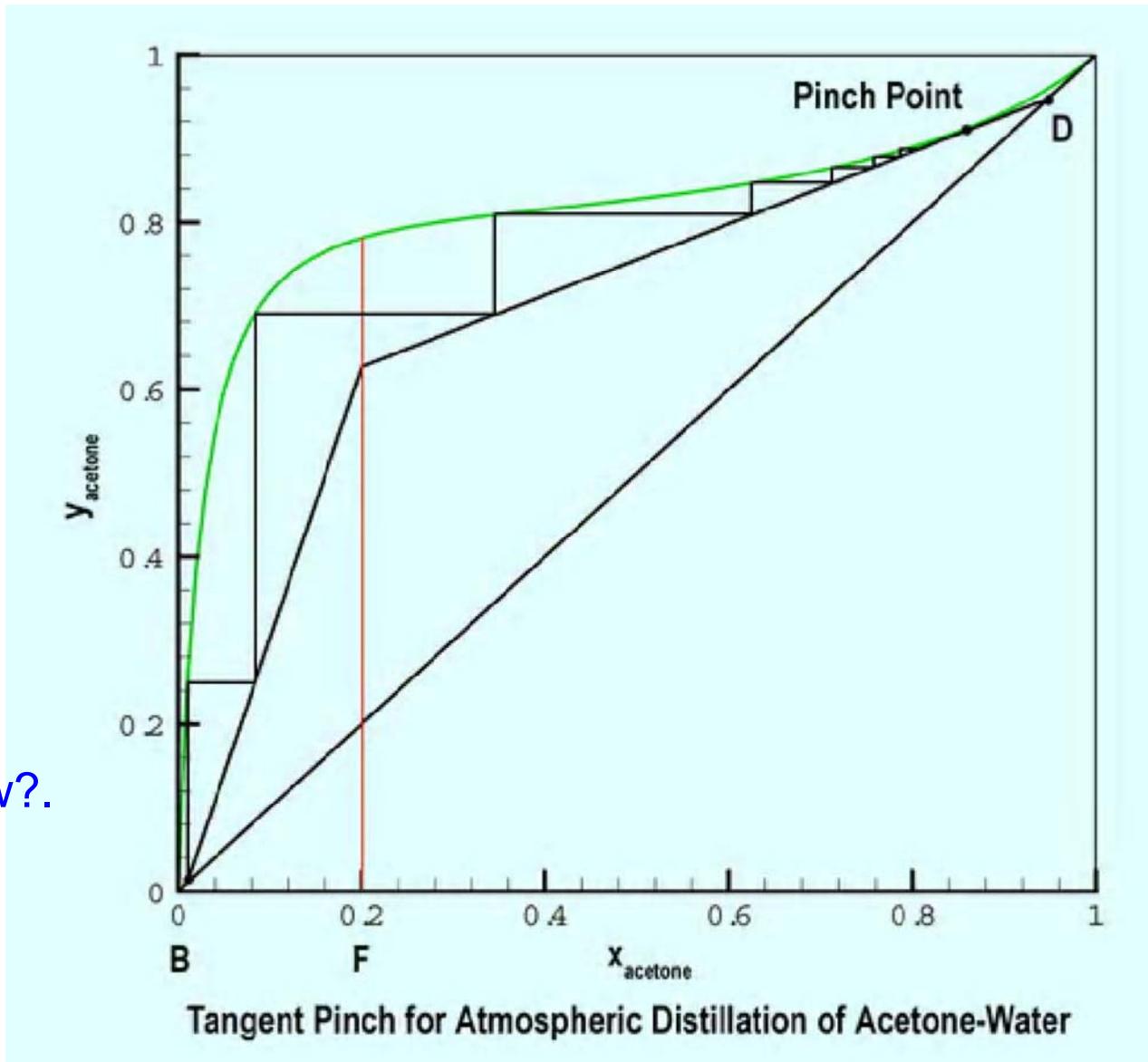
V-L Equilibrium

Non- Ideal Mixtures

X vs Y ; P=cte. diagram

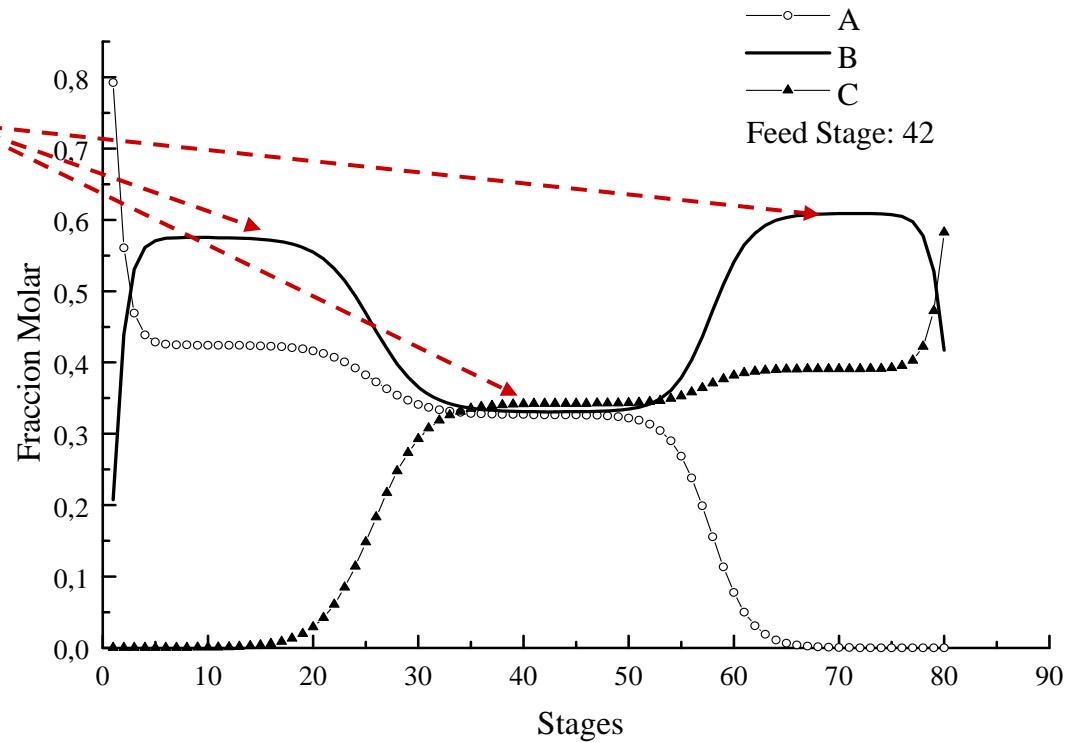
$$K_i = K_i(x, y, T, P) \\ = \gamma_i(x, T) P_i(T);$$

Constant molar overflow?



Multicomponent systems. Adiabatic Column. Constant pressure. Rigorous simulation.
Liquid composition profiles

Pinch points?



Composition profiles ($F, i = 0.33/ 0.33/ 0.34 - D, i = 0.7946/ 0.205/ 0 - B, i = 0/0.4184/0.5815 - p = 101.3 \text{ Kpa}$)
n-pentane, n-hexane and n-heptane

Ternary systems. Adiabatic Column. Constant pressure. Liquid composition profiles.

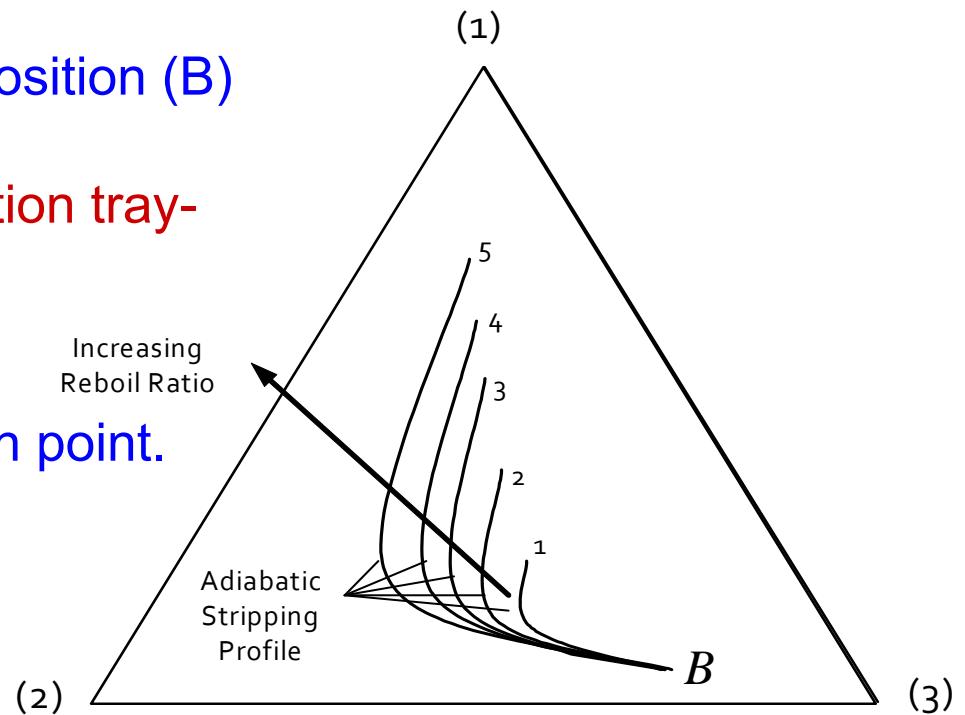
Rigorous simulation.

Specifying a liquid product composition (B)
and the reboil ratio,

the succession of liquid composition tray-tray profile can be computed.

The liquid profile approaches a pinch point.

The number of stages $\Rightarrow \infty$ at each
point: 1, 2, ...

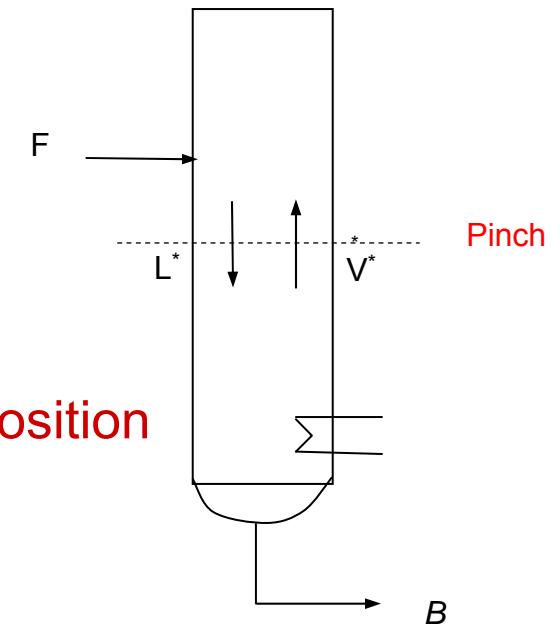
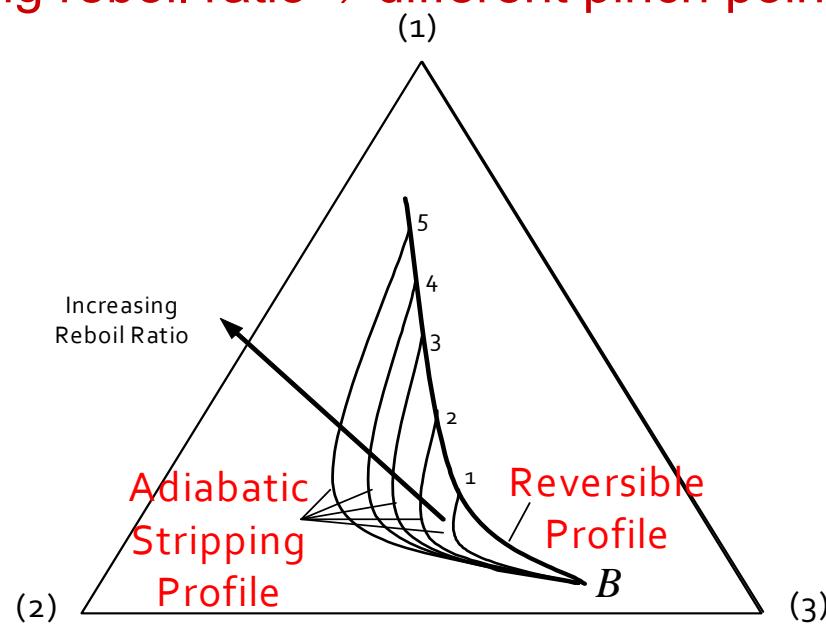


Ternary systems. Adiabatic Column. Constant pressure. Liquid composition profiles

Specifying:

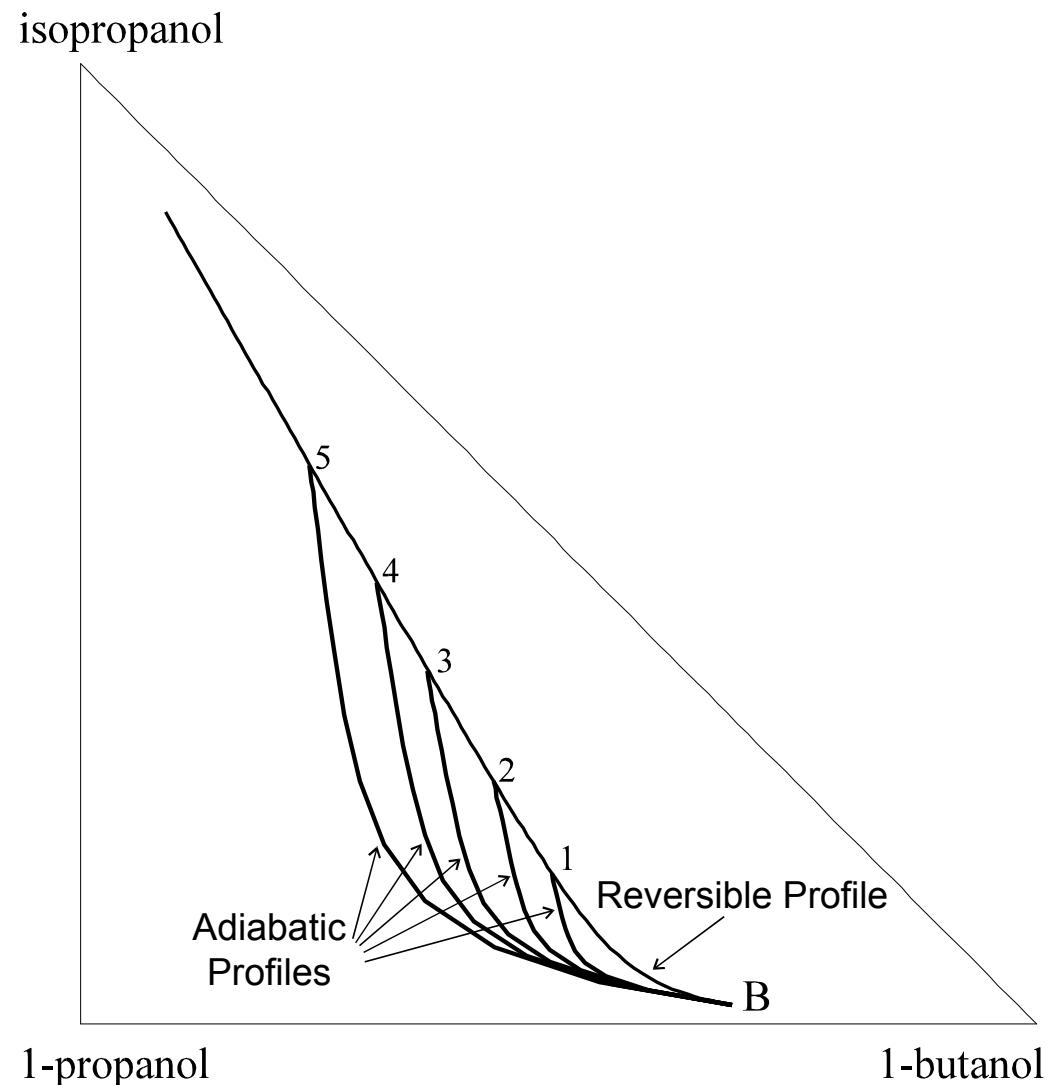
- 1) liquid product composition (B)
- 2) reboil ratio value
- 3) equilibrium between L^* and V^* ,
a liquid composition can be computed.

Changing reboil ratio \Rightarrow different pinch point composition



Reversible Profile = Collection of Pinch

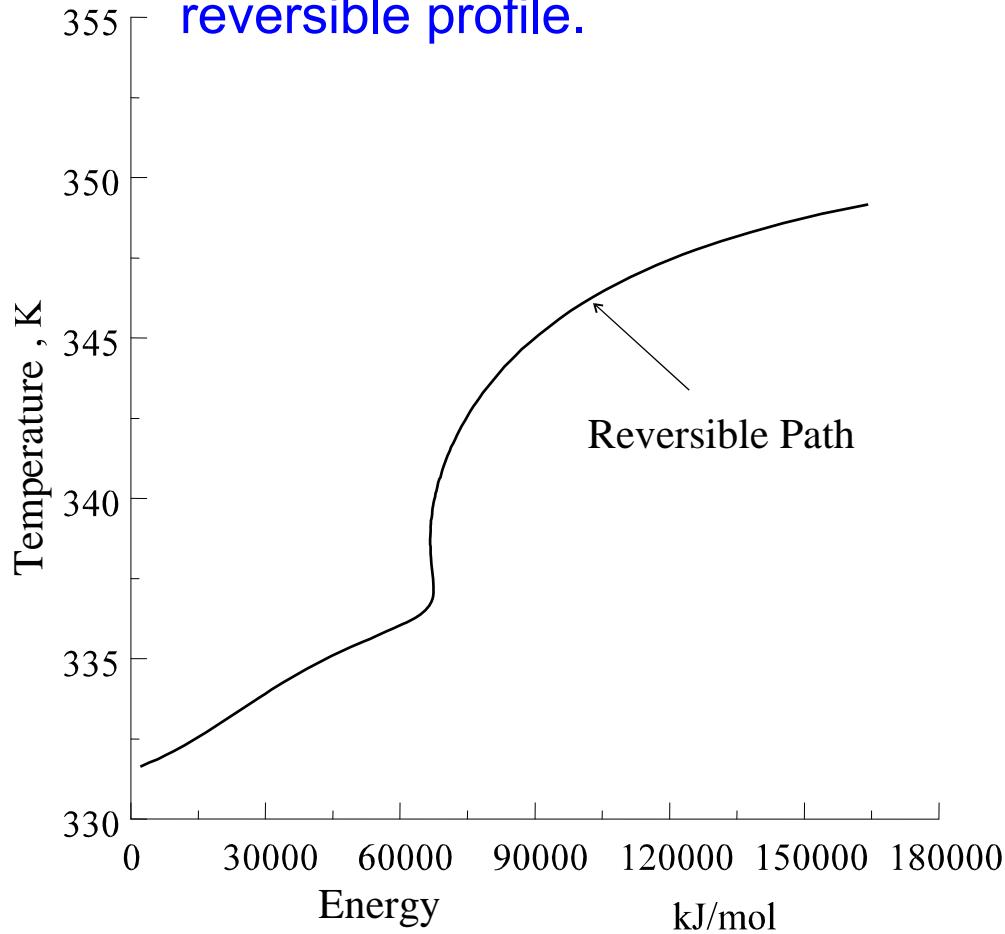
Reversible path for a bottom: distillative alcoholic mixture.

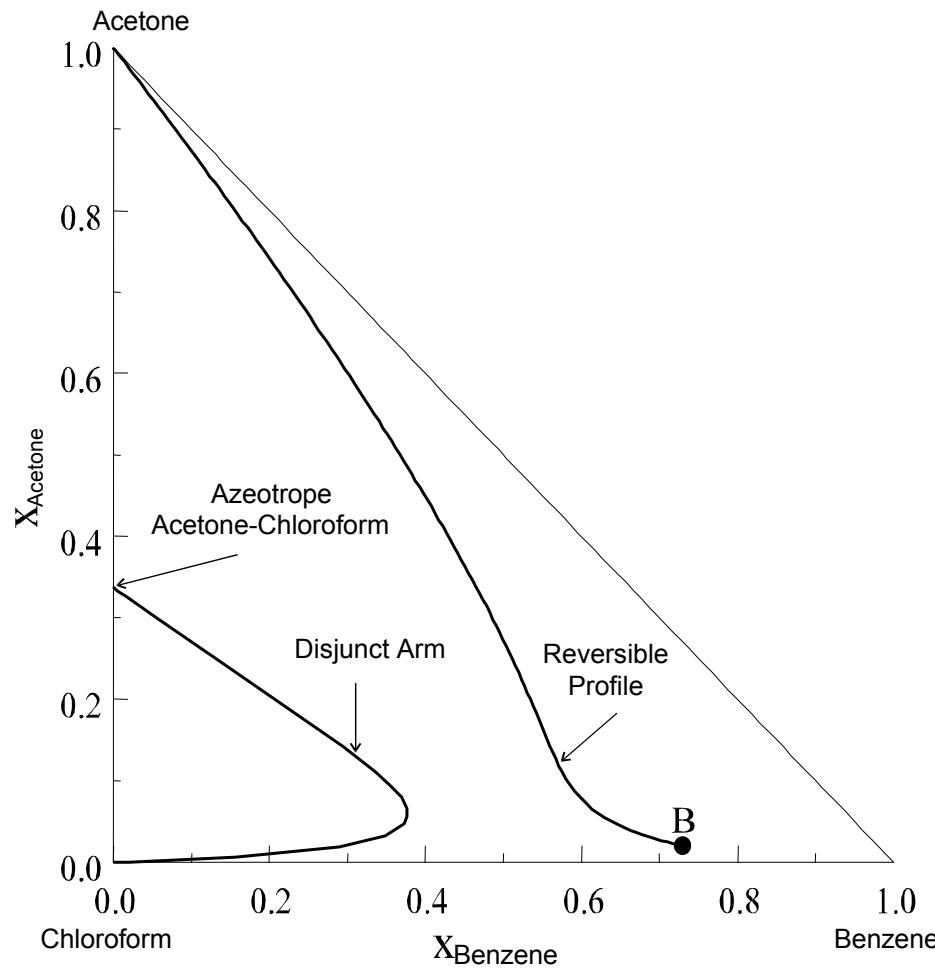


Reversible path for a bottom: distillative alcoholic mixture.

Starting from product D (T_D) and increasing reboil ratio, the profile is computed.

Temperature and Energy results from the reversible profile.





Reversible path for a bottom (B): azeotropic system.

Points computed with newton homotopy.

3.- Reversible distillation columns and sequences.

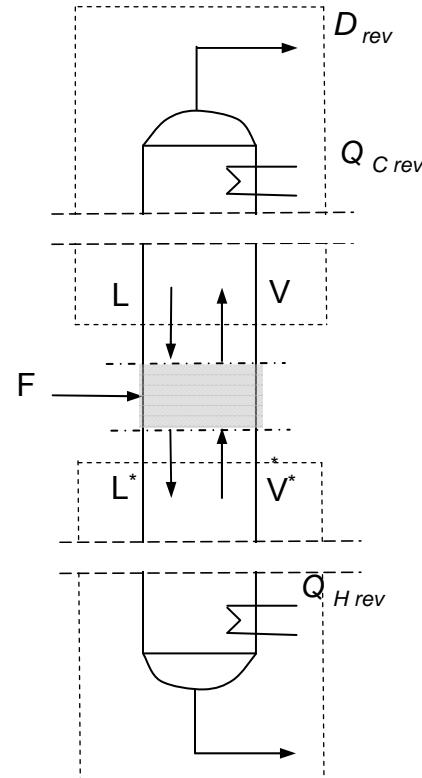
Specifying:

- 1) Feed composition,
- 2) product compositions B and D satisfying total mass balance,
- 2) reboil and reflux ratios satisfying total energy balance,
- 3) equilibrium between L^* and V^* and between L and V,

liquid composition profiles can be computed.

PINCH CURVES, PINCH PATHS,
REVERSIBLE PROFILES, REVERSIBLE
PATHS, ETC.

NO ASSUMPTION ON V-L EQUILIBRIUM
AND ON ENTHALPY MODELS



3.- Reversible distillation columns and sequences.

PINCH EQUATIONS

$$y_{i,s} = K_{i,s} x_{i,s} \quad \text{Equilibrium}$$

Mass balances

$$L_s x_{i,s} + D x_{i,D} = V_s y_{i,s}$$

Given:

X_D, Y_D, P, T_D and Q_D

at least one liquid composition

can be computed: $\underline{X}_{s,s}$

$s \in S$: set of different kinds of pinches.

$$L_s + D = V_s$$

Energy balance

$$L_s h_{L,s} + D h_D + Q_D = V_s h_{V,s}$$

NO ASSUMPTION ON V-L
EQUILIBRIUM AND ON
ENTHALPY MODELS

Thermodynamic Properties

$$K_i = K_i(x_{i,s}, y_{i,s}, T_s, p_s)$$

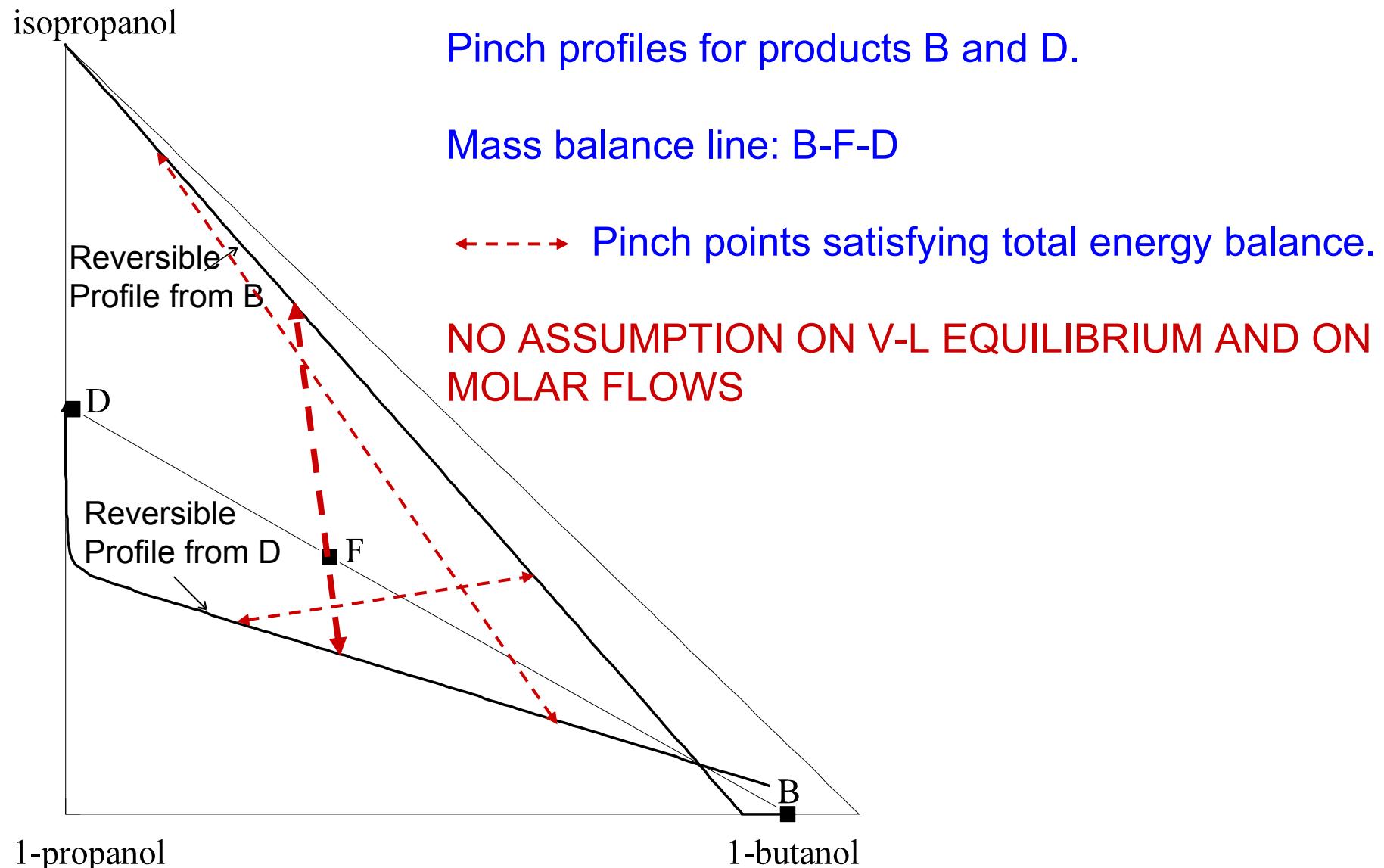
$$h_{L,s} = h_{L,s}(x_{i,s}, T_s, p_s)$$

$$h_V = h_V(y_{i,s}, T_s, p_s)$$

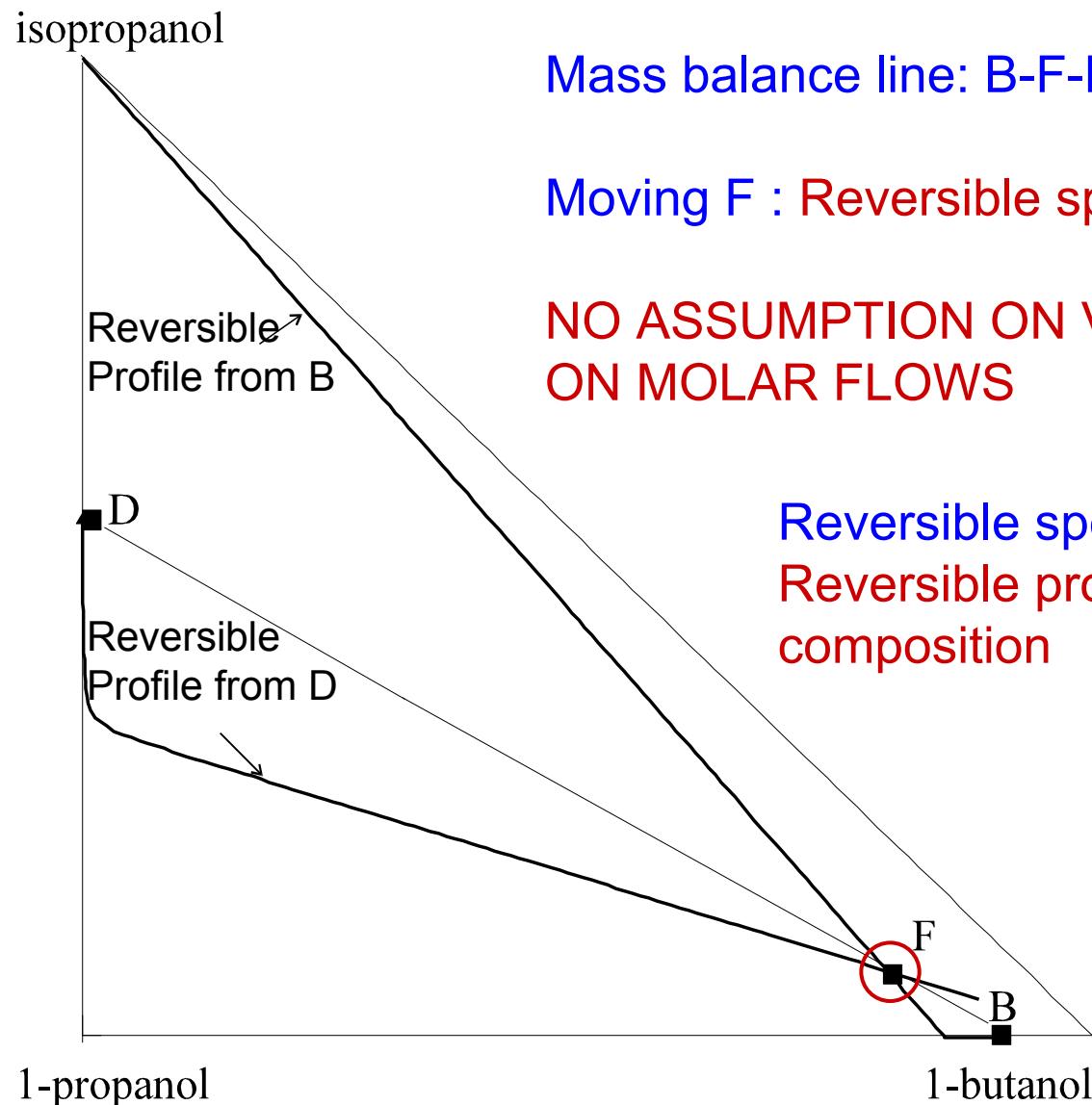
$$p_s = p_C$$

Constant pressure

3.- Reversible distillation columns and sequences.



Pinch profiles for products B and D.



Mass balance line: B-F-D

Moving F : Reversible specification

NO ASSUMPTION ON V-L EQUILIBRIUM AND
ON MOLAR FLOWS

Reversible specification
Reversible profiles intersect at feed
composition

isopropanol

Reversible Profile
from Bottom

P_2

Pinch profiles for products B and D.

Sharp splits.

Adiabatic profiles and pinch profiles.

Pinch points → critical points for the adiabatic profiles

**NO ASSUMPTION ON V-L EQUILIBRIUM AND ON
MOLAR FLOWS**

D

Reversible Profile
from Top

P_3

Adiabatic
Profiles

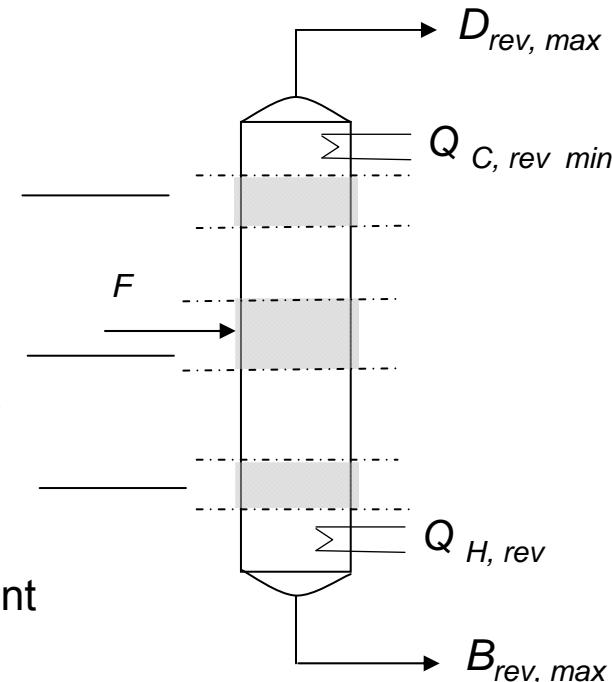
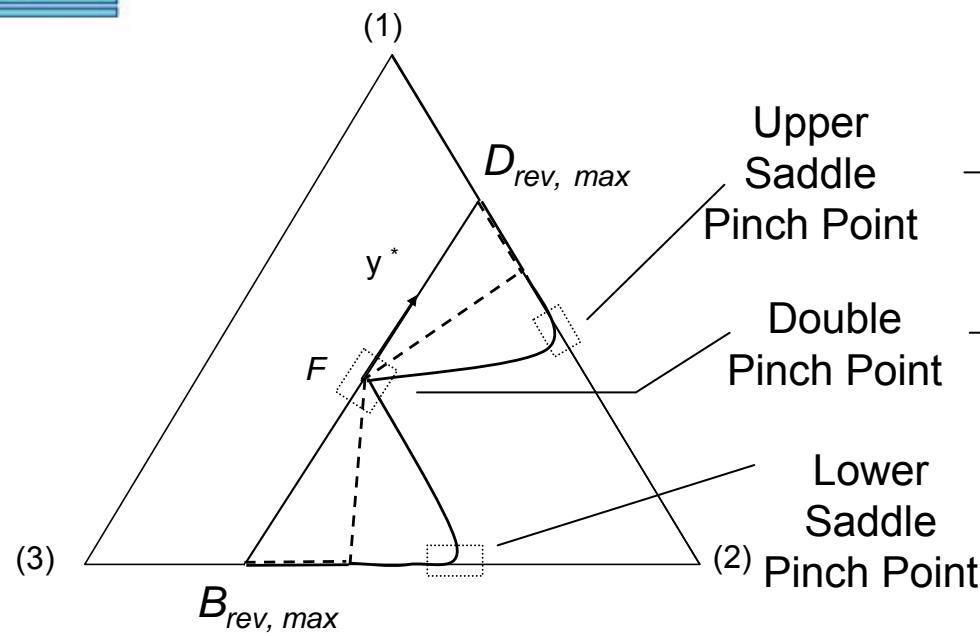
P_1

Stable pinch
Saddle pinch

1-propanol

Axis – Binary Reversible Profile

1-butanol



Reversible Distillation. (Sharp).

For each feed composition, there exists a special, “preferred” specification.

V-L equilibrium vector (Y^*-X^*) belongs to Mass balance line: D-F-B.

NO ASSUMPTION ON V-L EQUILIBRIUM AND ON MOLAR FLOWS

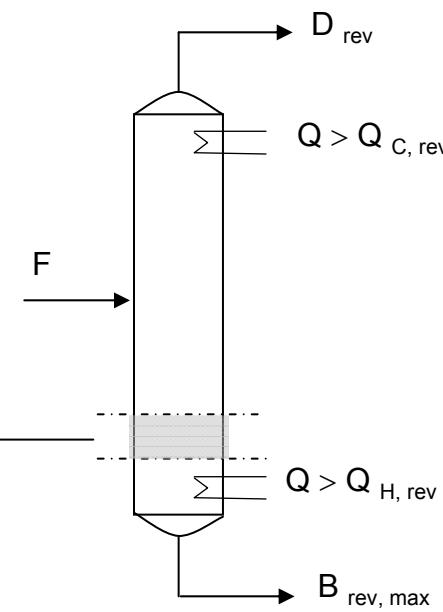
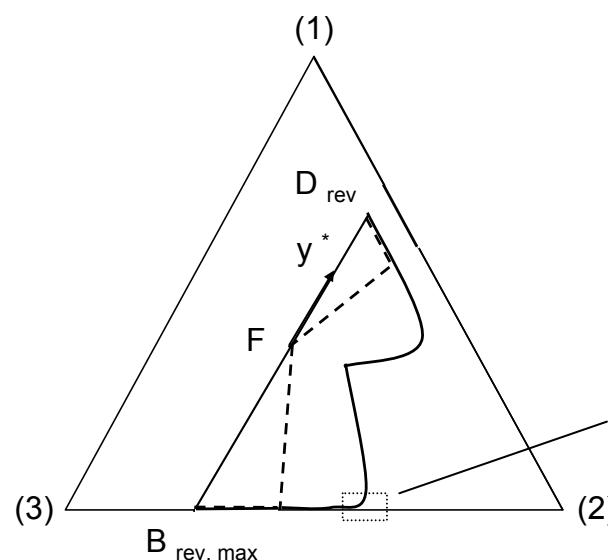
Reversible distillation.

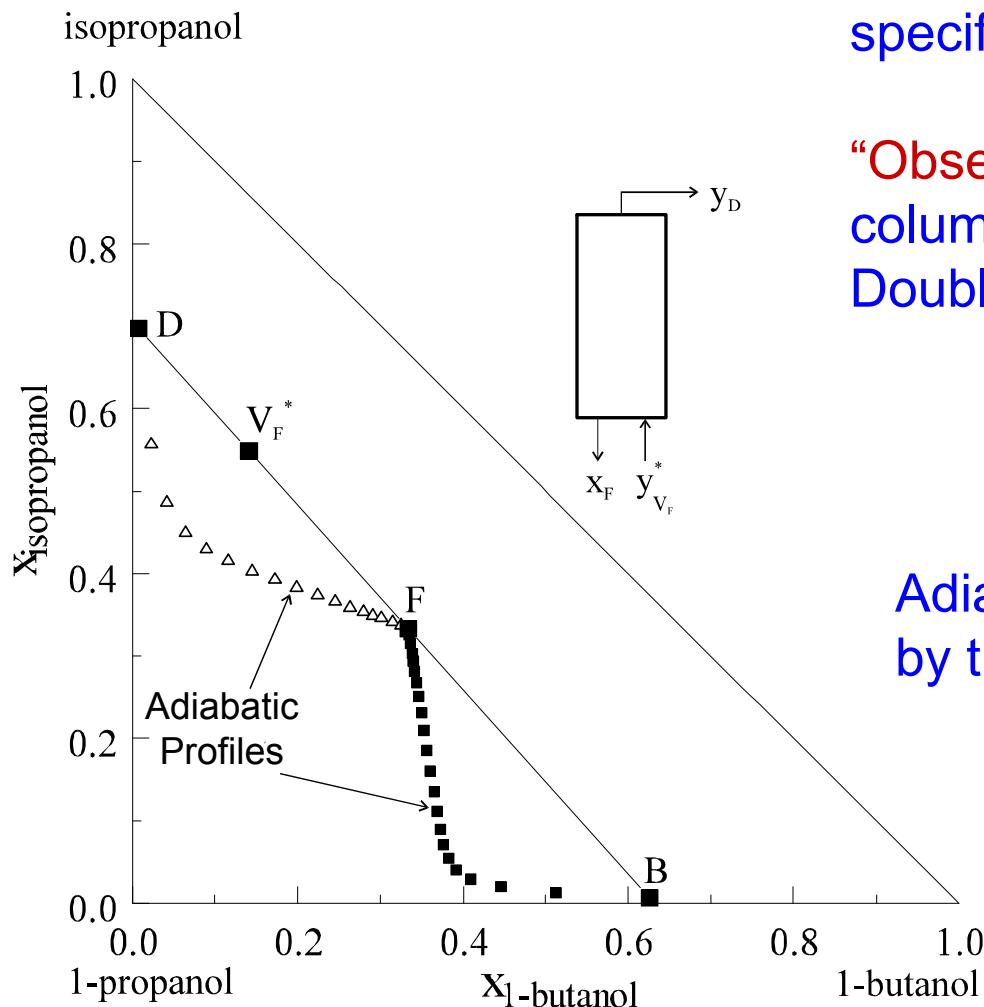
For each feed composition, there exists a special “preferred” product specification set.

Other product specifications \Rightarrow Non reversible distillation (1 col.)

Are pinch points always “observable” in adiabatic columns profiles?

It depends on feed and product specification and on RR. Non sharp in this figure.





Reversible distillation.

$(Y^* - X^*) \square B-F-D \Rightarrow$ Reversible specification

“Observable” pinch in adiabatic column.
Double feed pinch.

Adiabatic profile shape is determined by the pinch point.

Non Reversible distillation.

$(X^* - Y^*) \notin B-F-D$

Pinch topology.

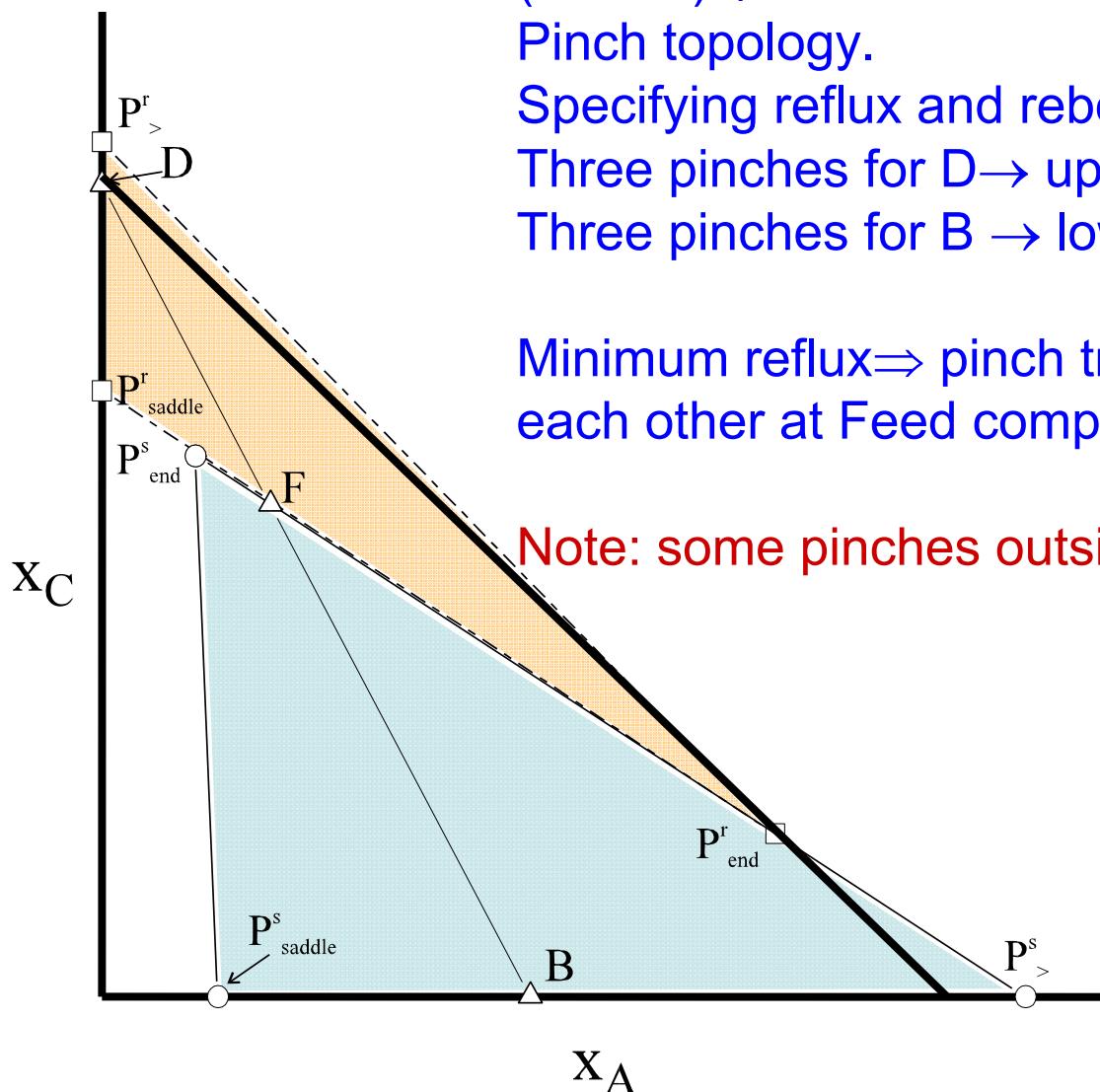
Specifying reflux and reboil ratios

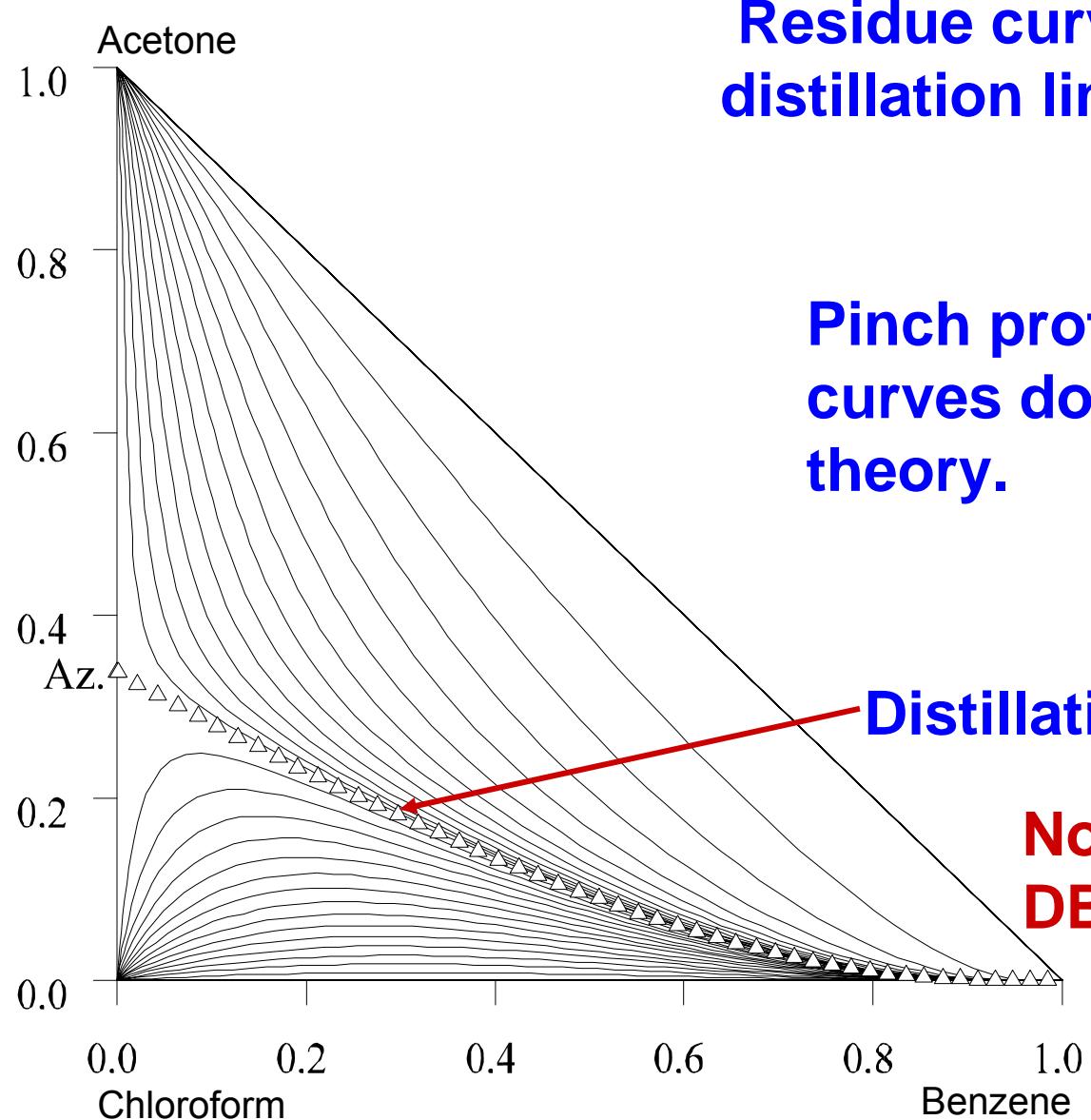
Three pinches for D \rightarrow upper triangle

Three pinches for B \rightarrow lower triangle

Minimum reflux \Rightarrow pinch triangles touch each other at Feed composition

Note: some pinches outside: $x_b < 0$





Residue curves distillation lines

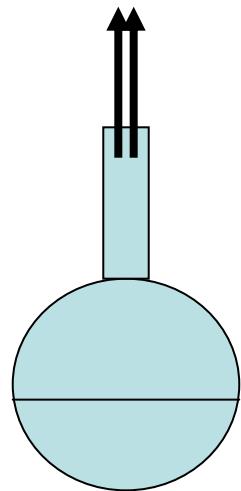
Pinch profiles and Residue
curves dominate distillation
theory.

Distillation boundary (DB)

Not possible to cross
DB at total reflux

Residue curves

F_{vap} ; \underline{Y} ; T ; P

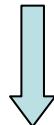


V ; \underline{X} ; T ; P

$$\frac{d(V x_i)}{dt} = -F_{Vap} y_i$$

$$\frac{dV}{dt} = -F_{Vap}$$

$$K_i x_i = y_i$$

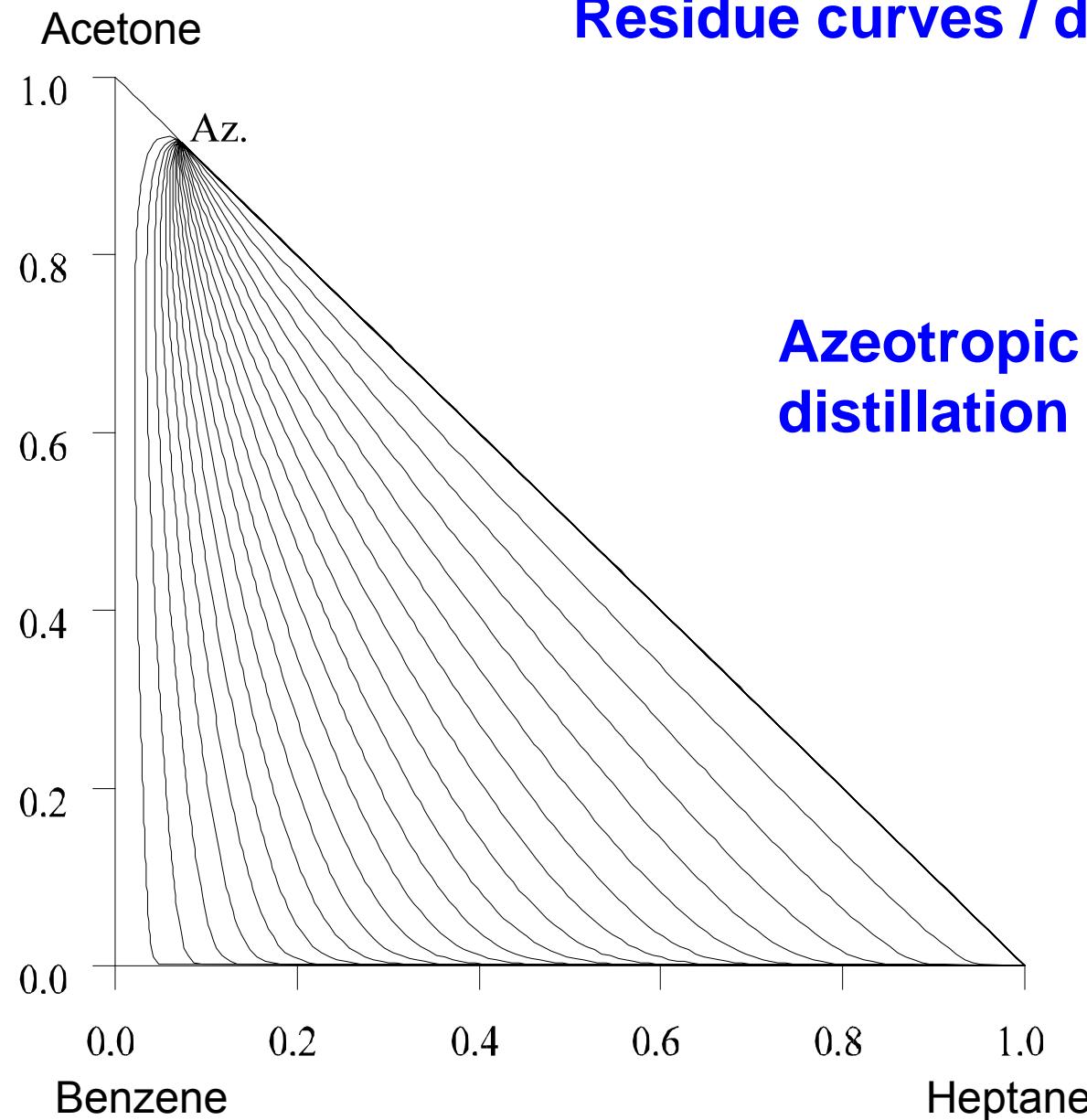


$$V \frac{d x_i}{dt} = -F_{Vap} (y_i - x_i)$$

$$\frac{dV}{dt} = -F_{Vap}$$

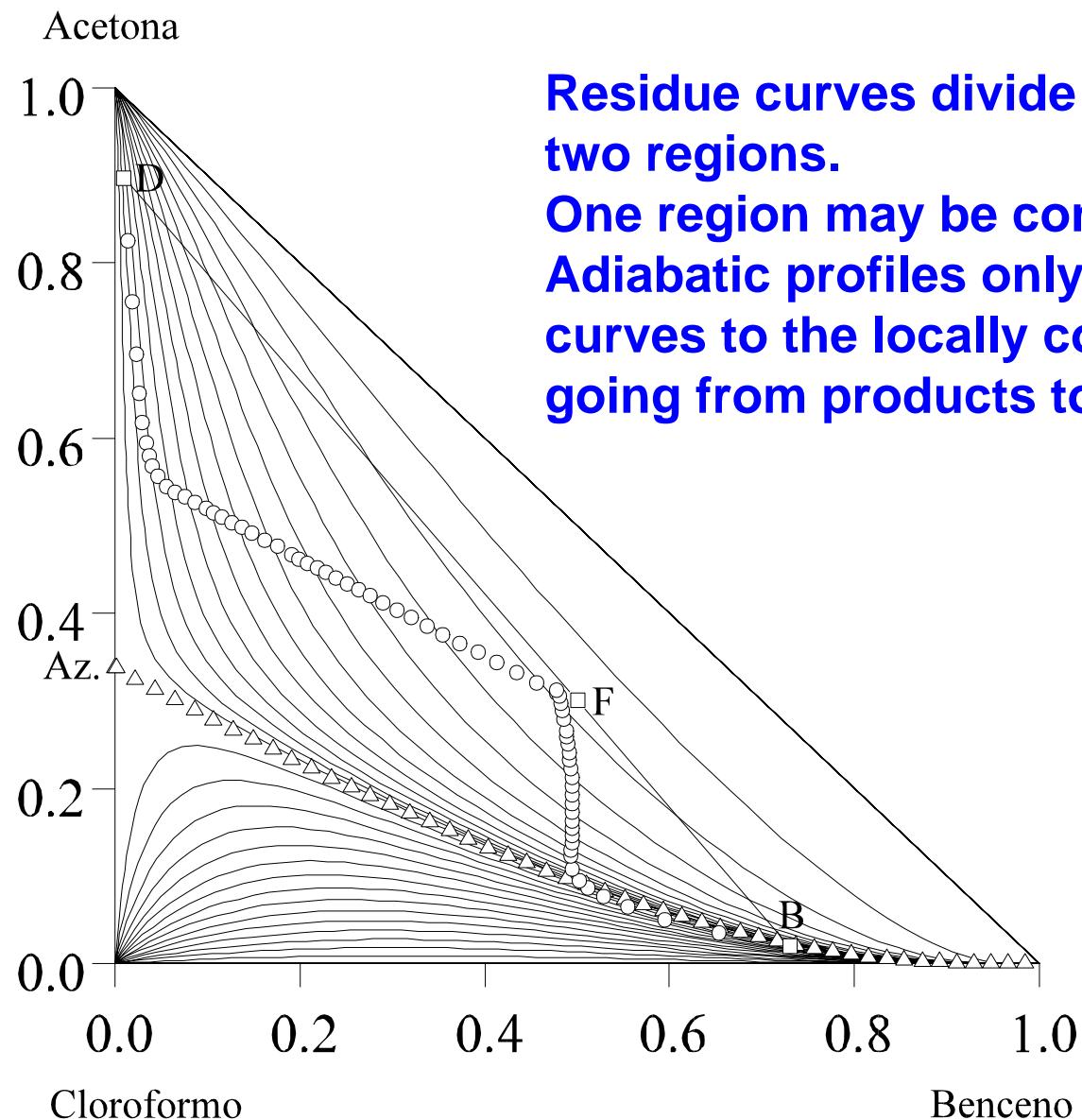
$$K_i x_i = y_i$$

Residue curves / distillation lines



Azeotropic system with no distillation boundary

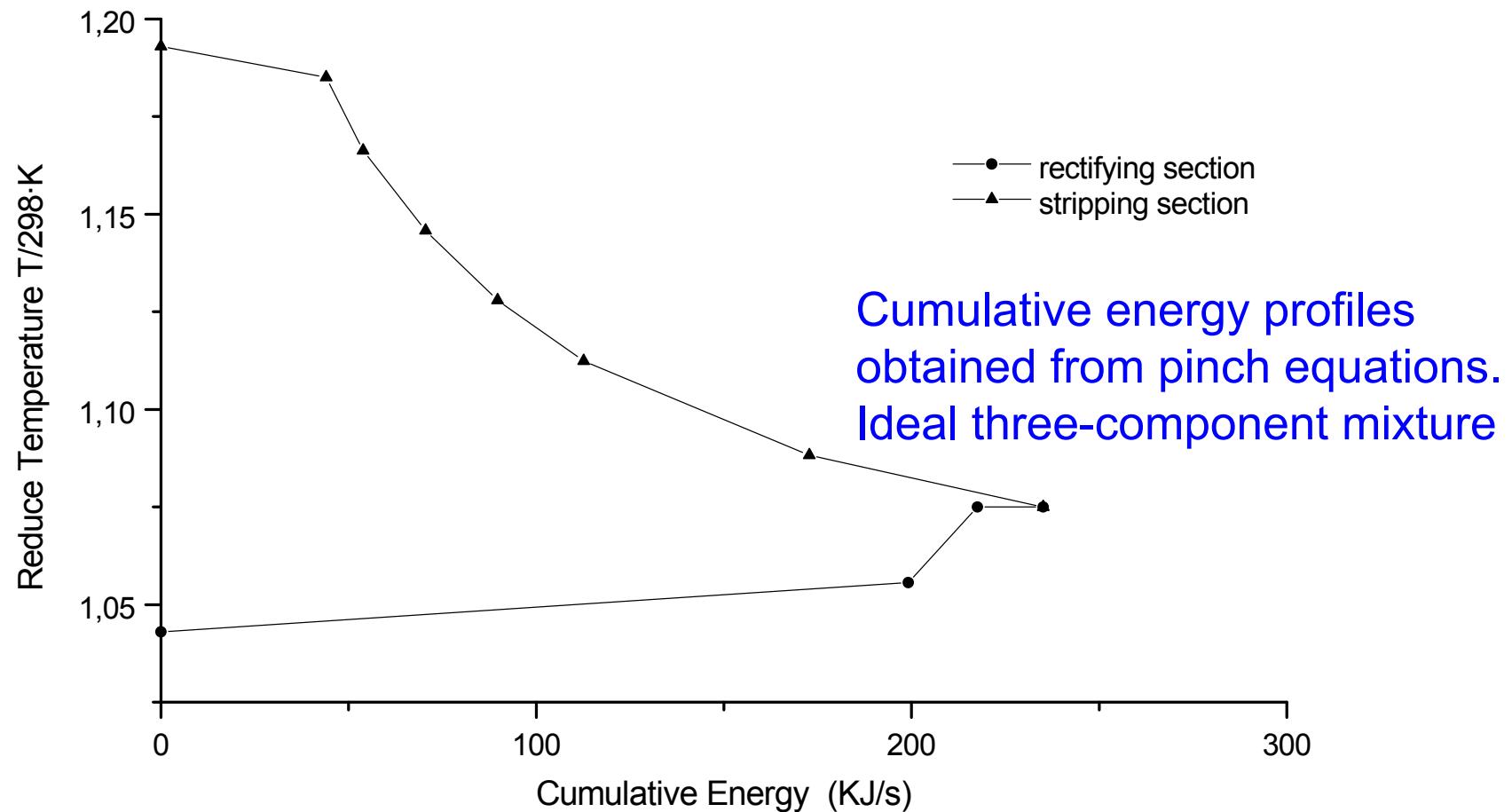
Ternary systems



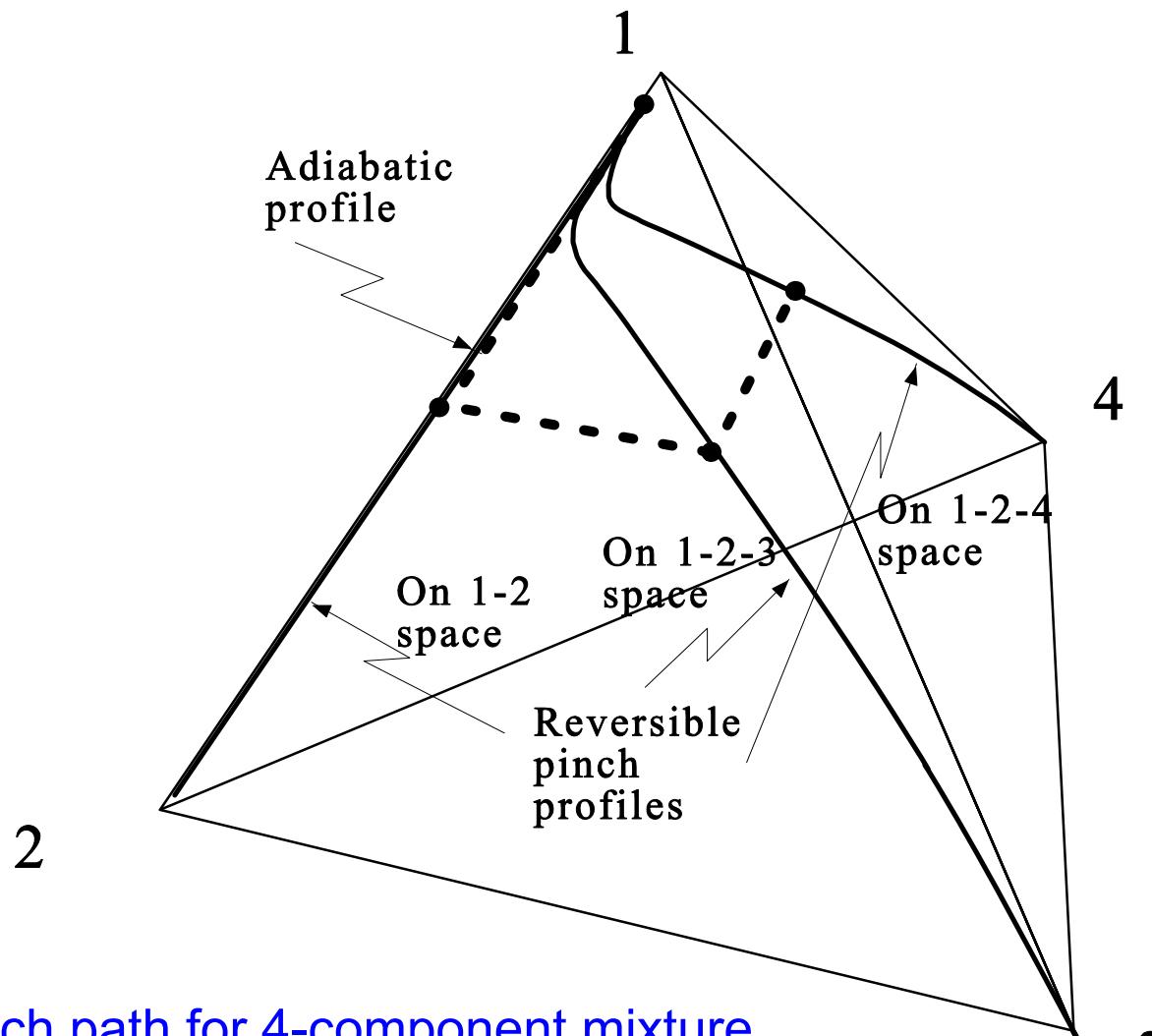
**Residue curves divide the simplex in two regions.
One region may be convex.
Adiabatic profiles only cross residue curves to the locally convex side when going from products to feed.**

Improving energy distribution in Simple columns.

Temperature – Energy Pinch profiles



Improving energy distribution in simple columns.



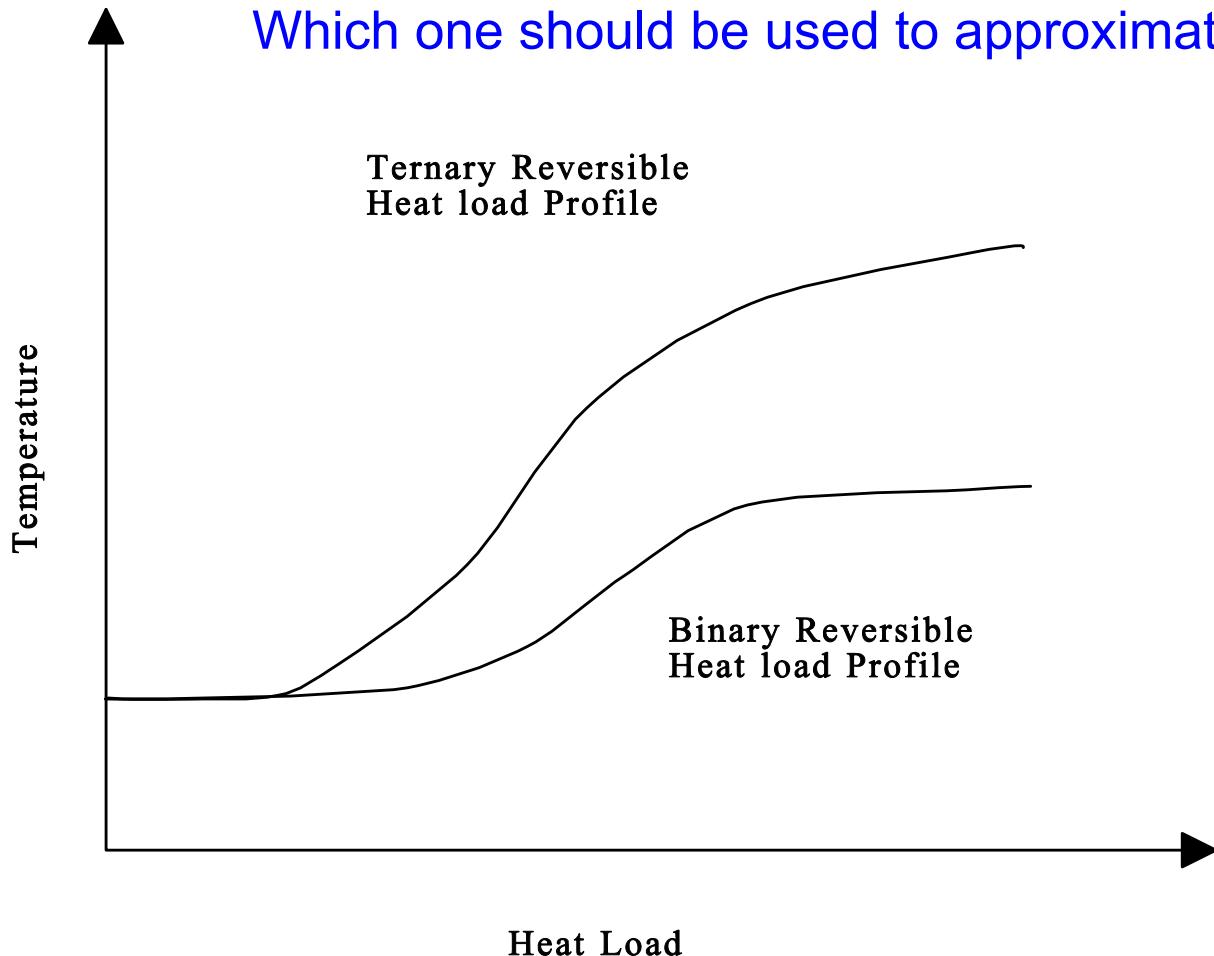
Pinch path for 4-component mixture.
of paths into the simplex: 3

Improving energy distribution in simple columns.

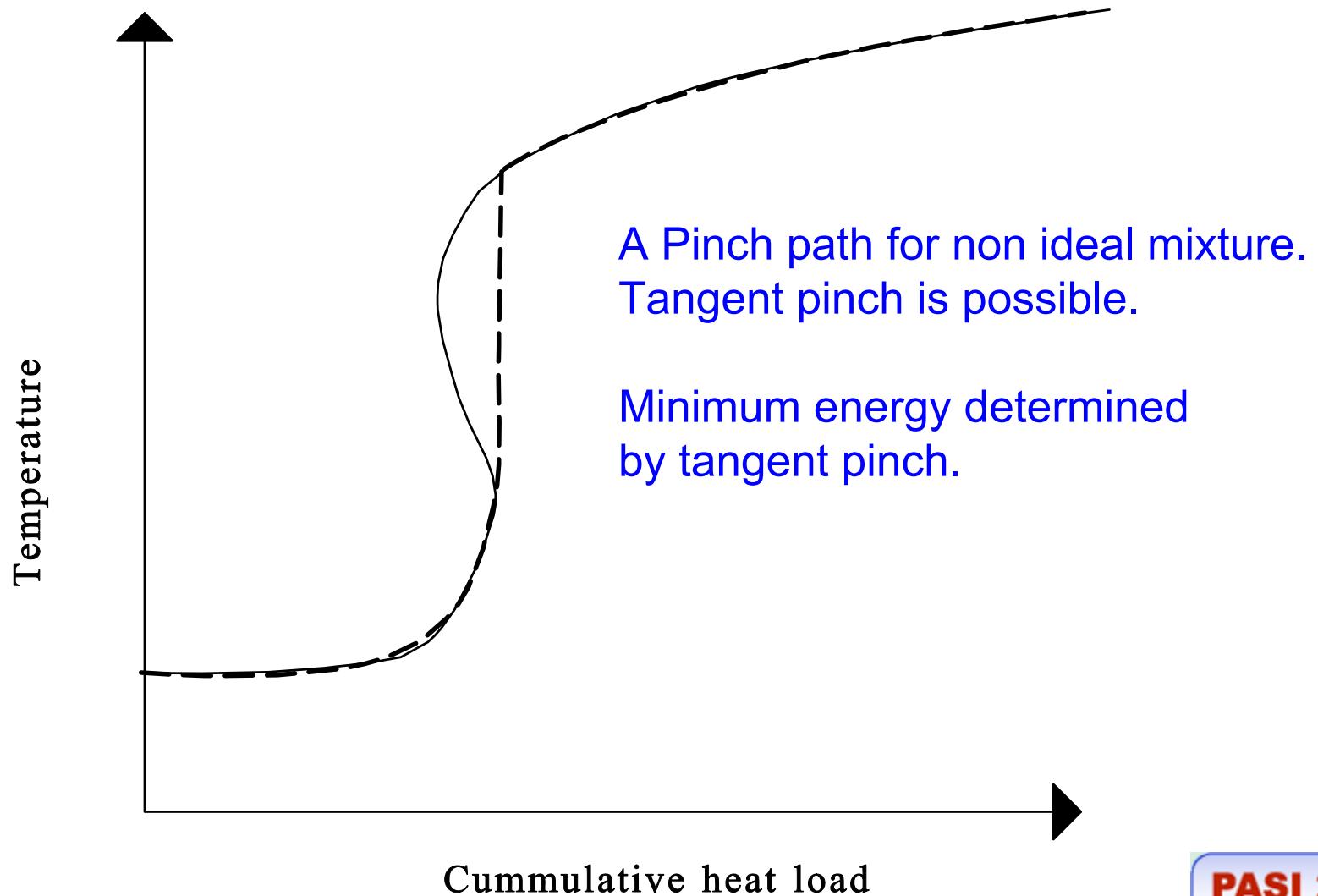
Pinch path for 3-component mixture.

of paths into the simplex: 2

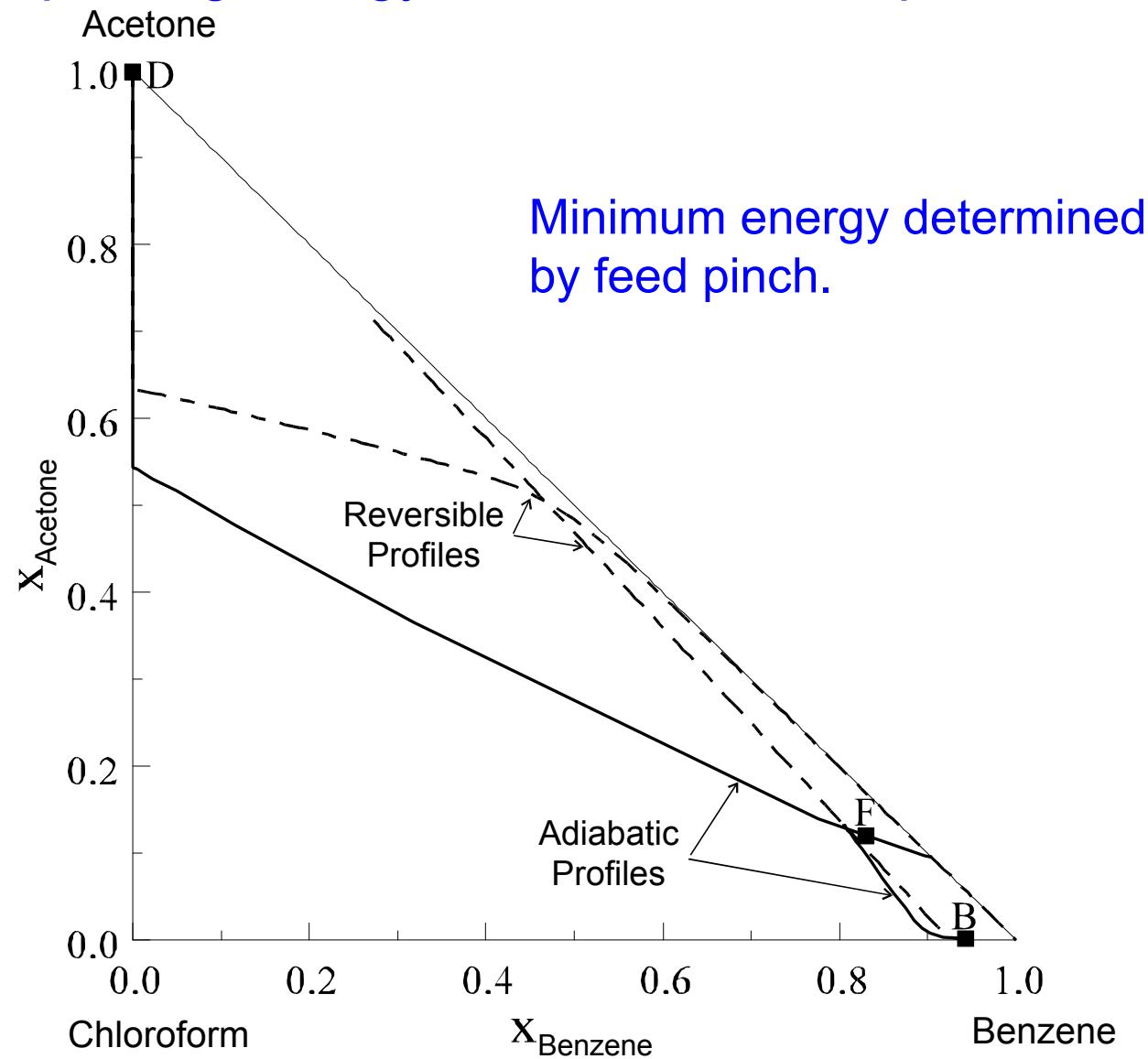
Which one should be used to approximate reversible profiles?



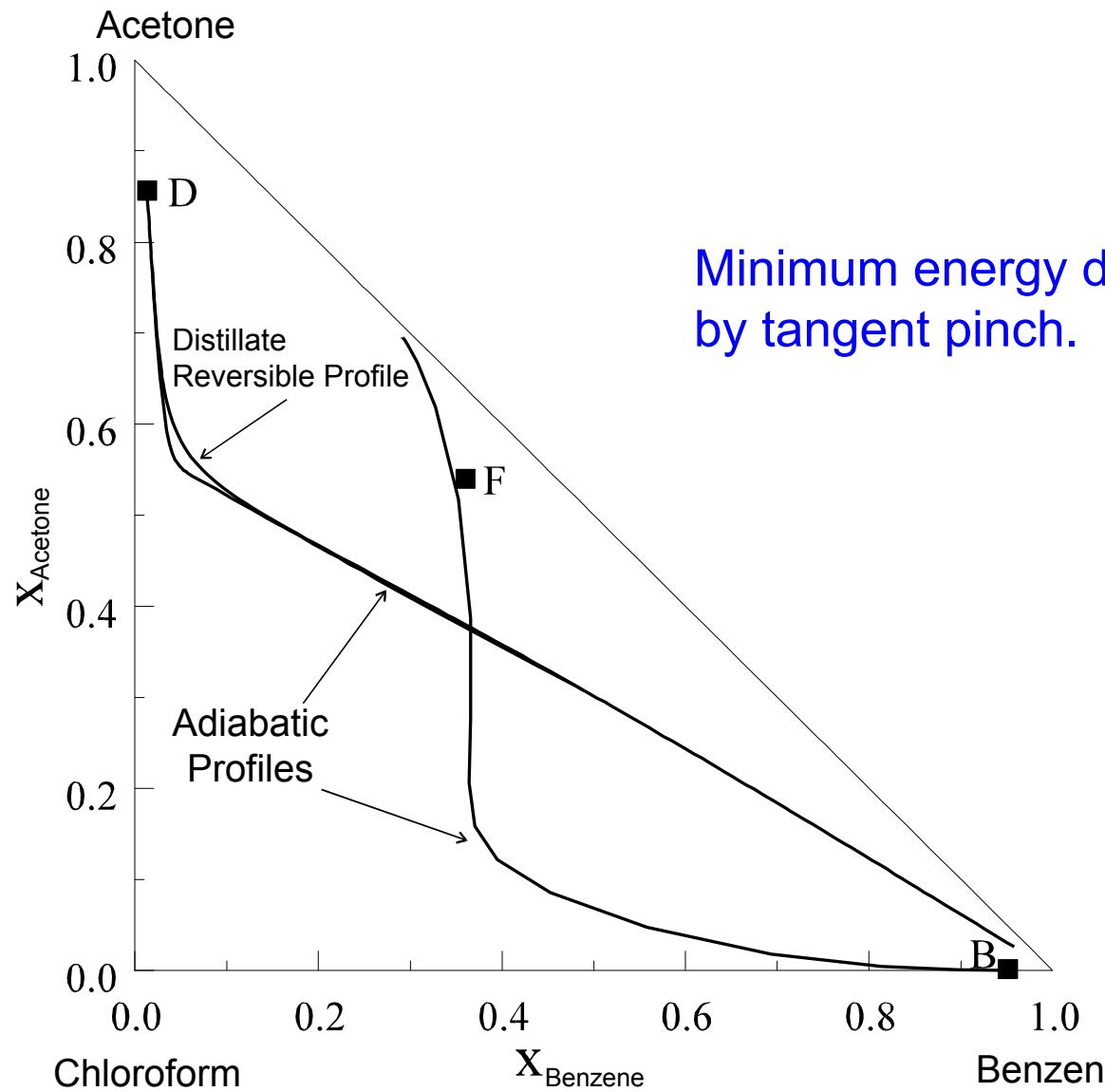
Improving energy distribution in simple columns.



Improving energy distribution in simple columns.

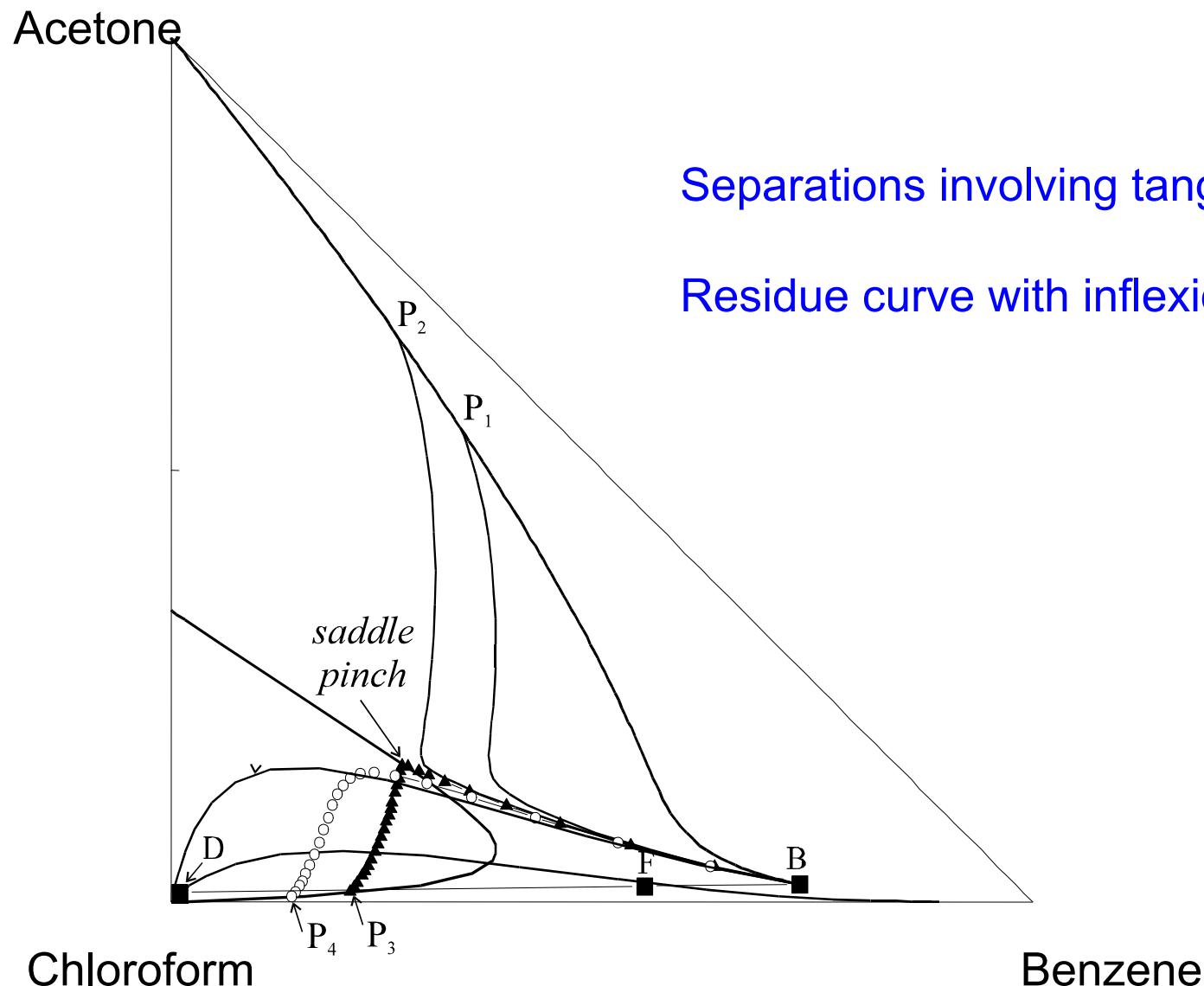


Improving energy distribution in simple columns.



Minimum energy determined
by tangent pinch.

Improving energy distribution in simple columns.



Separations involving tangent pinch.

Residue curve with inflexion points.

3.- Reversible distillation columns and sequences.

Reversible Distillation Sequence Model (RDSM)

What is the **superstructure** that most closely approximates the Reversible Distillation of Multicomponent Mixtures like?

Which is the optimal **energy distribution** in the RDSM-based sequence?

A Reversible Column

Theoretical model.

Infinite number of stages.

Continuous heat distribution along column length.

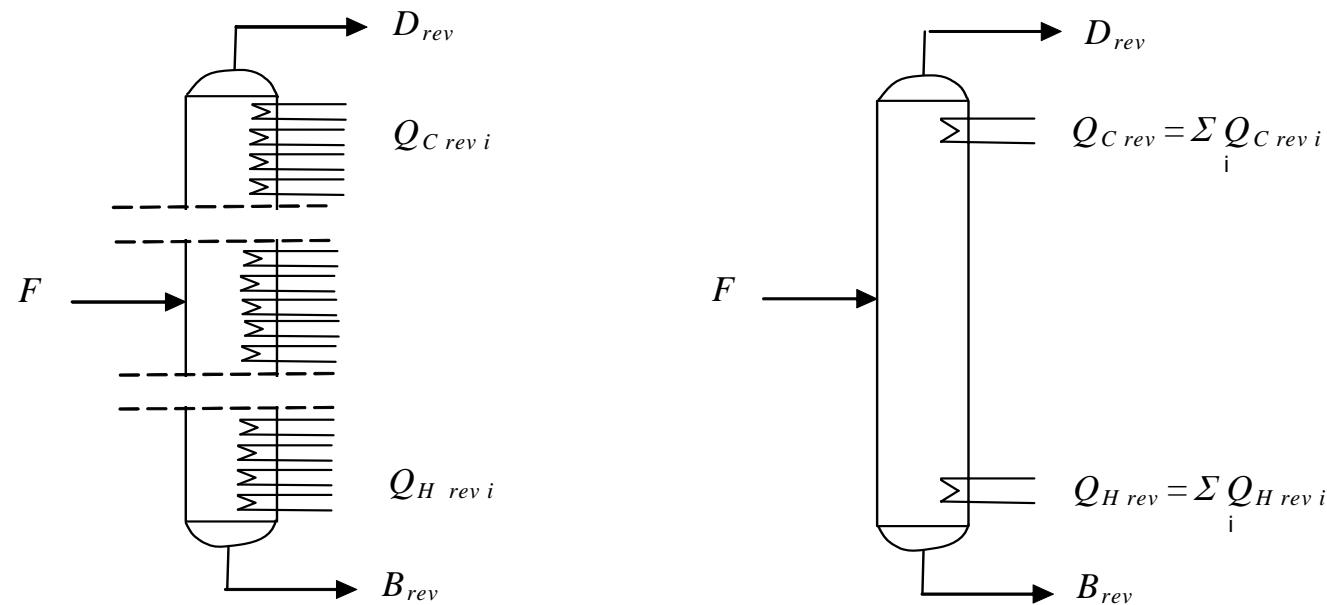
Heat to and from the column is transferred at zero temperature difference.

No contact of non-equilibrium streams take place in any point of the column.

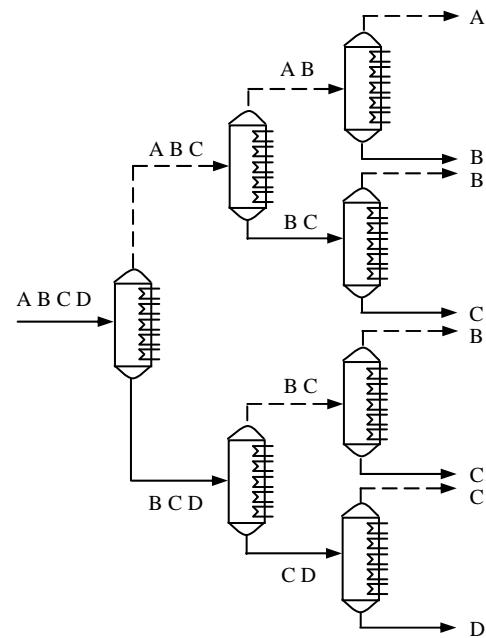
Only one separation task can be performed: **the reversible separation.**

RDC Adiabatic Approach

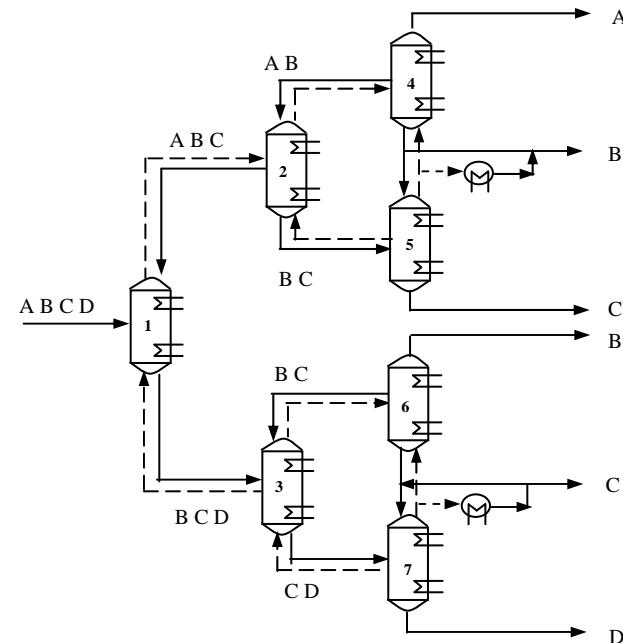
A sufficiently high number of stages must be employed.
The same energy is involved in the separation with a different distribution.
The same products are achieved but different composition profiles develop within each unit.



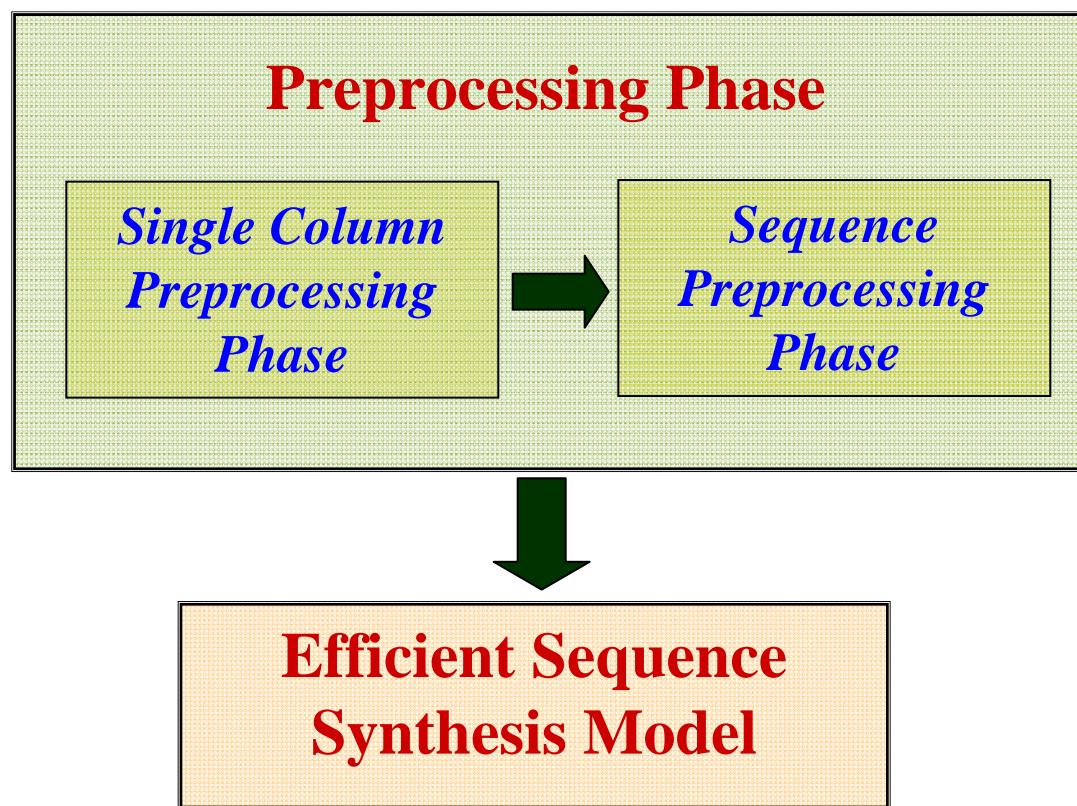
RDSM Sequence



RDSM-based Sequence



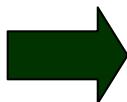
RDSM-based Synthesis Model



Preprocessing Phase

Single Column Preprocessing

Adiabatic Approximation to the Reversible Separation



- ✓ Values to **initialize** variables.
- ✓ Values to **bound** critical variables related to unit interconnection and heat loads.
- ✓ Parameters to **formulate** objective functions.

Sequence Preprocessing

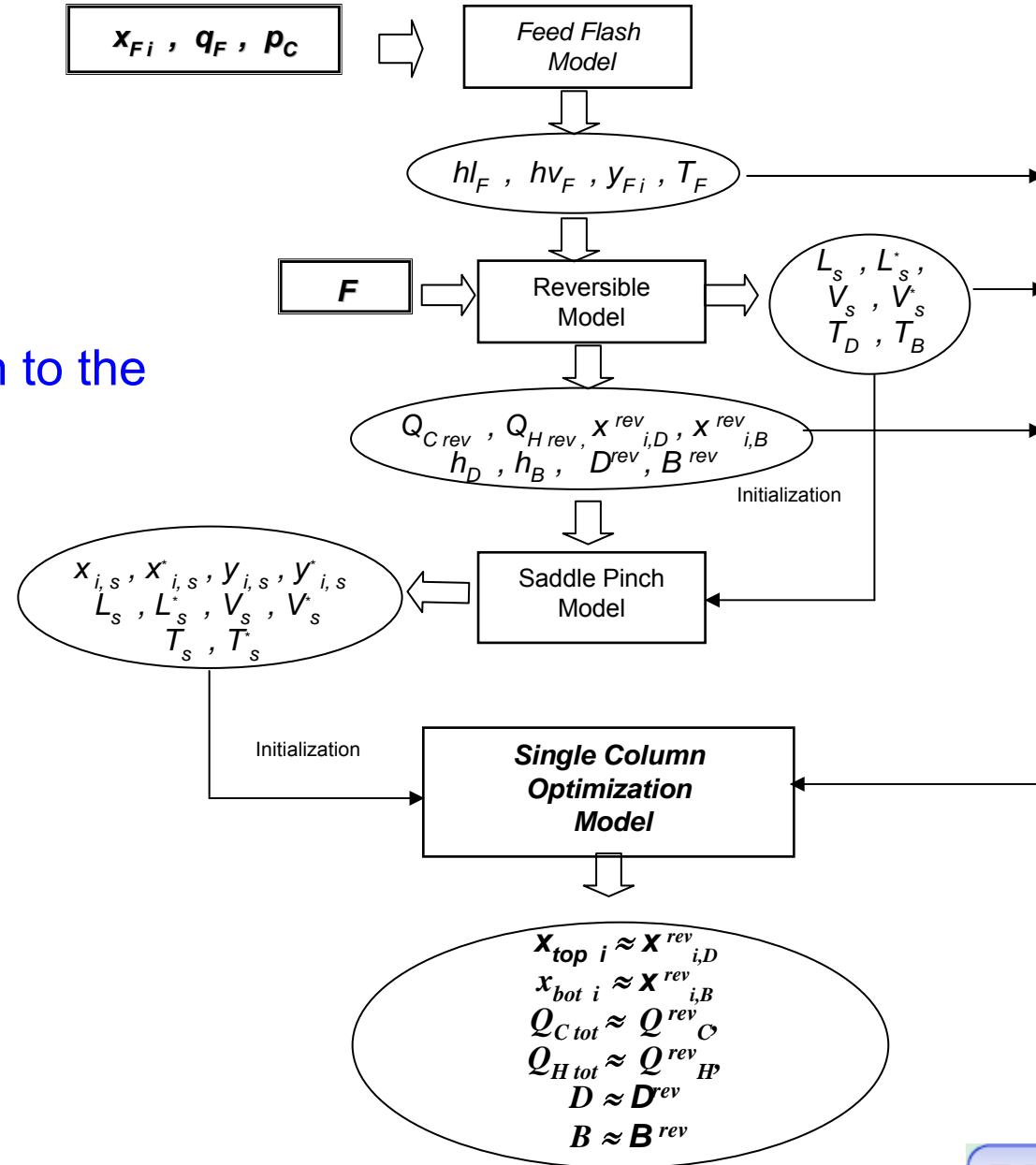
Reversible Products,
Saddle Pinch Points and
Reversible Exhausting Pinch
Points Calculations



- ✓ Optimal Energy Distribution for each single unit.
- ✓ **Number of stages** for each single column of the sequence.
- ✓ Objective functions and constraints to approach the reversible separation task.

Single Column Preprocessing

Adiabatic Approximation to the Reversible Separation



(RM):

$$\min \quad z = x_{c,D}^{rev} + x_{a,B}^{rev}$$

$$L + (1 - q_F) F = L^*$$

$$V + q_F F = V^*$$

Known values from
flash calculations at Feed

$$L + D^{rev} = V$$

$$L^* = V^* + B^{rev}$$

$$L x_i + D^{rev} x_{i,D}^{rev} = V y_i$$

$$L^* x_i = V^* y_i + B^{rev} x_{i,B}^{rev}$$

$$L h_L + D^{rev} h_D + Q_C^{rev} = V h_V$$

$$L^* h_L + Q_H^{rev} = V^* h_V + B^{rev} h_B$$

Single Column Preprocessing

Adiabatic Approximation to the
Reversible Separation

$$K_i = K_i(x_i, y_i, T, p) \quad i=1, \dots, NC$$

$$h_L = h_L'(x_i, T, p)$$

$$h_V = h_V(y_i, T, p)$$

(SPM):

$$x_{NC,s} = 0 \quad y_{NC,s} = 0$$

$$L_s + D^{rev} = V_s$$

$$L_s x_{i,s} + D^{rev} x_{i,D}^{rev} = V_s y_{i,s}$$

$$L_s h_{L,s} + D^{rev} h_D + Q_C^{rev} = V_s h_{i,s}$$

Known values from
flash calculations at Feed

$$x_{1,s}^* = 0 \quad y_{1,s}^* = 0$$

$$L_s^* = V_s^* + B^{rev}$$

$$L_s^* x_{i,s} = V_s^* y_{i,s} + B^{rev} x_{i,B}^{rev}$$

$$L_s^* h_{L,s}^* + Q_H^{rev} = V_s^* h_{V,s}^* + B^{rev} h_B$$

Single Column Preprocessing

Adiabatic Approximation to the
Reversible Separation

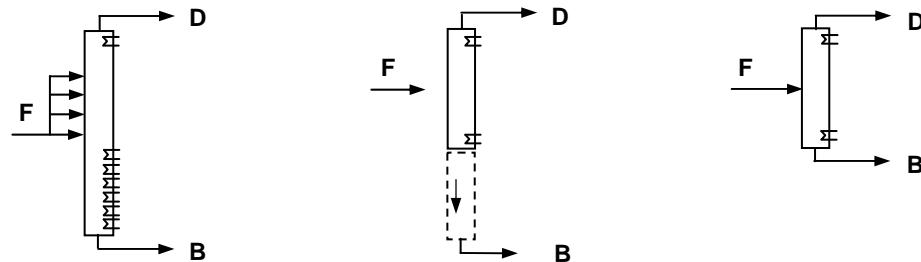
$$K_i = K_i(x_i, y_i, T, p) \quad i=1, \dots, NC$$

$$h_L = h_L'(x_i, T, p)$$

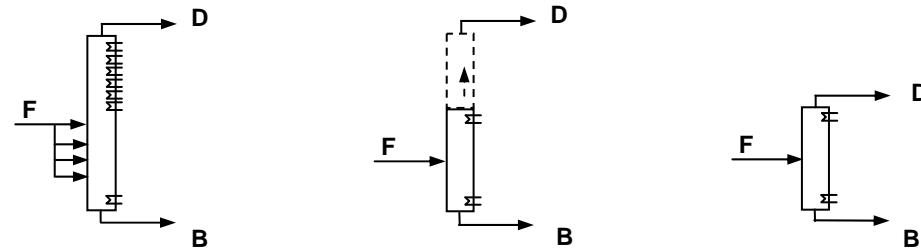
$$h_V = h_V(y_i, T, p)$$

The number of trays is selected by optimizing the condenser, reboiler and/or feed stream locations.

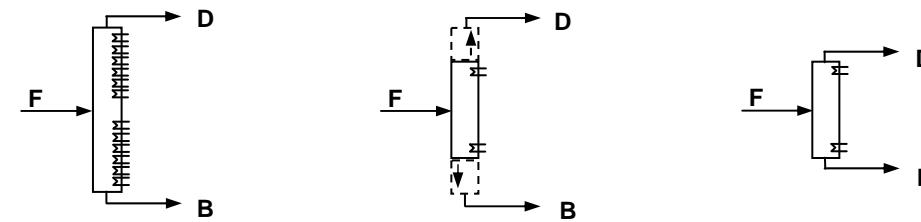
Variable feed and
reboiler location



Variable feed and
condenser location

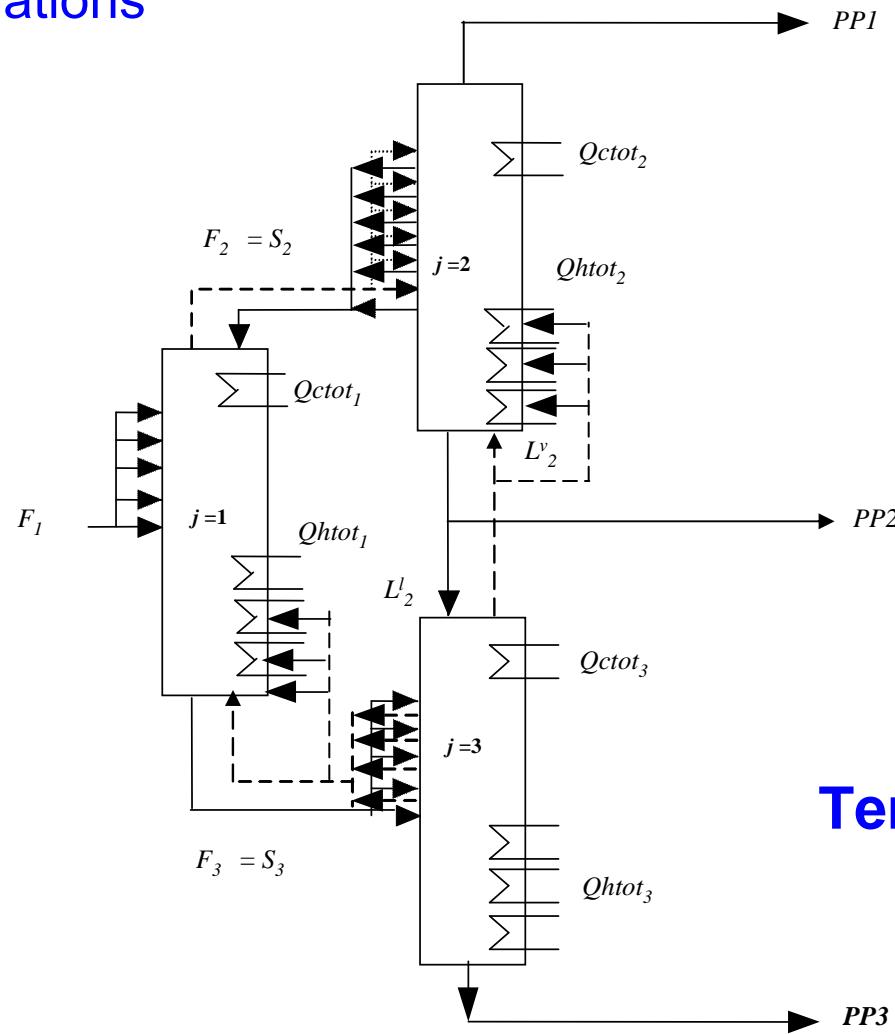


Variable condenser and
reboiler location

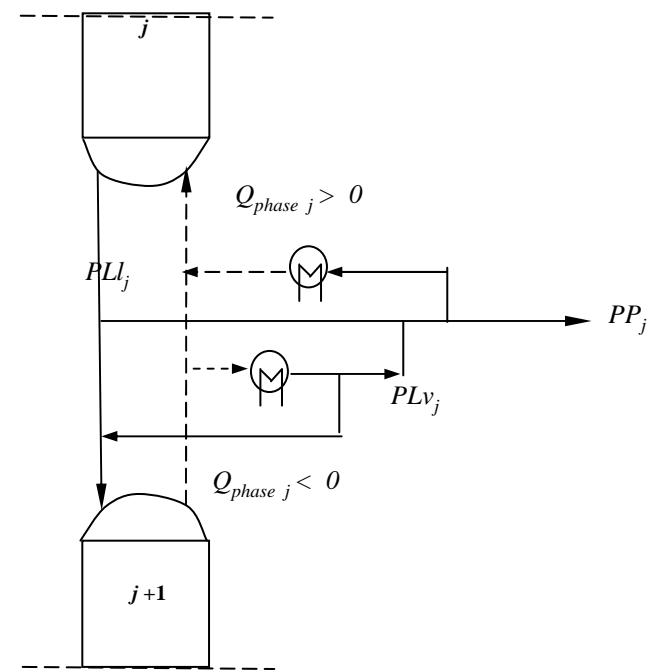


Sequence Preprocessing

Reversible Products, Saddle Pinch Points
and Reversible Exhausting Pinch Points
Calculations

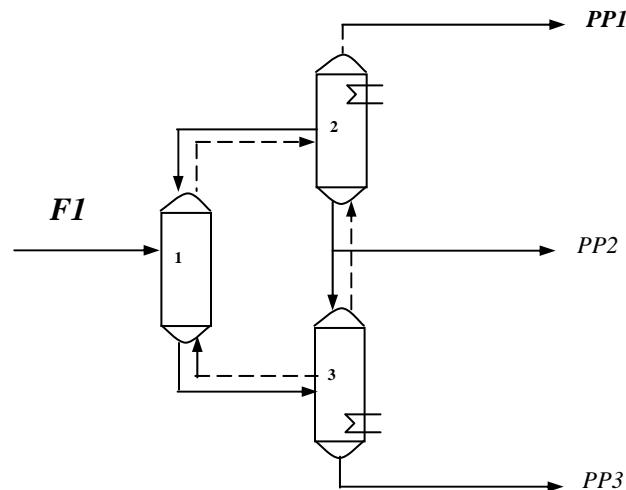


Heat Integration Superstructure

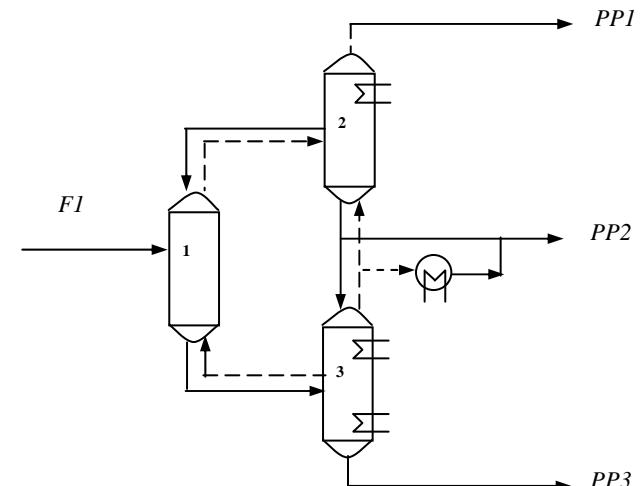


Ternary mixture.

Less efficient RDSM-based superstructures:

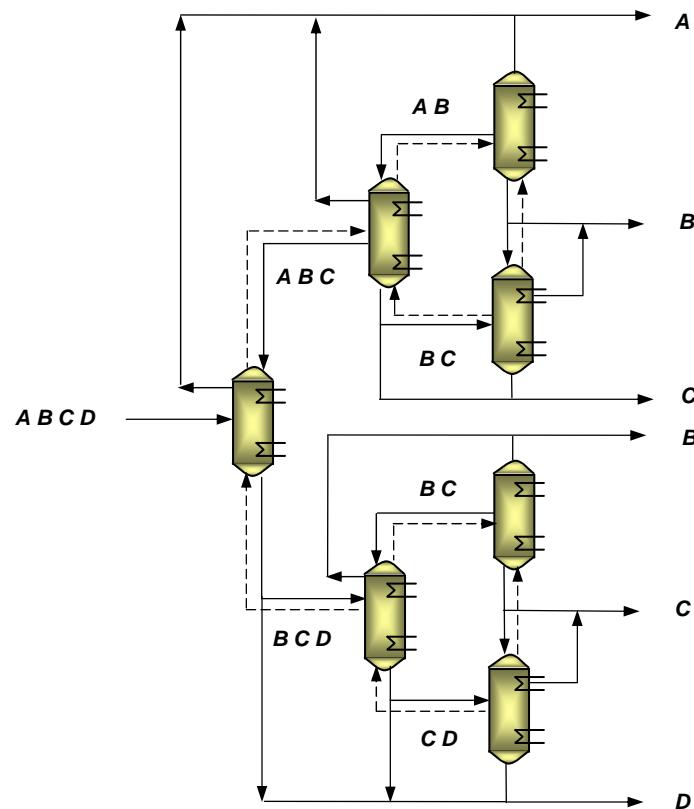


Fully Thermally Coupled
Scheme (Petlyuk Column)

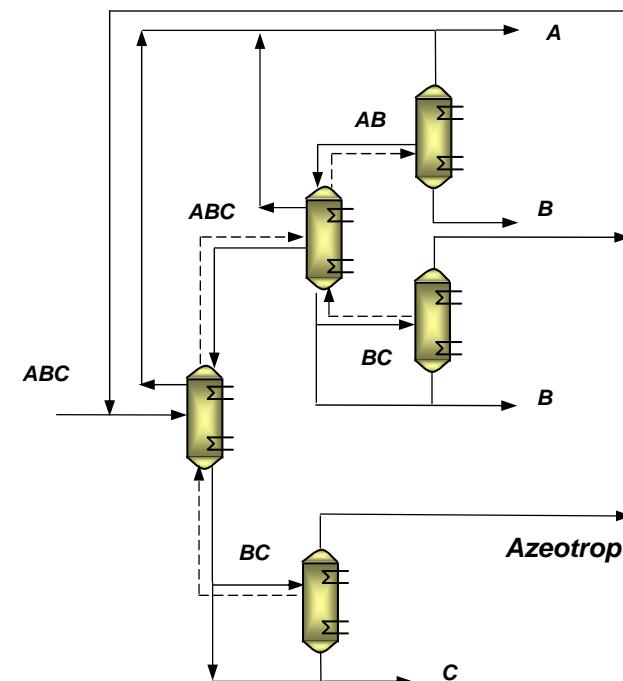


Fully Thermally Coupled Scheme
with Heat Integration

3.- Reversible distillation columns and sequences.



Zeotropic Mixture

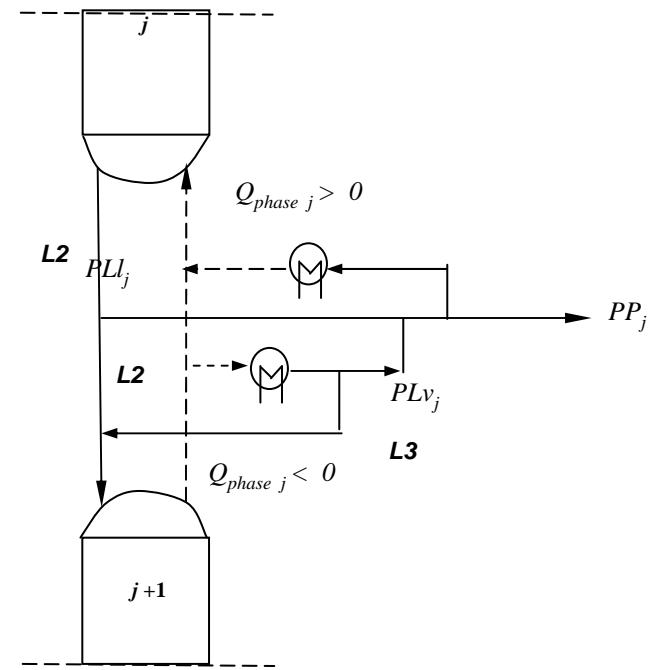
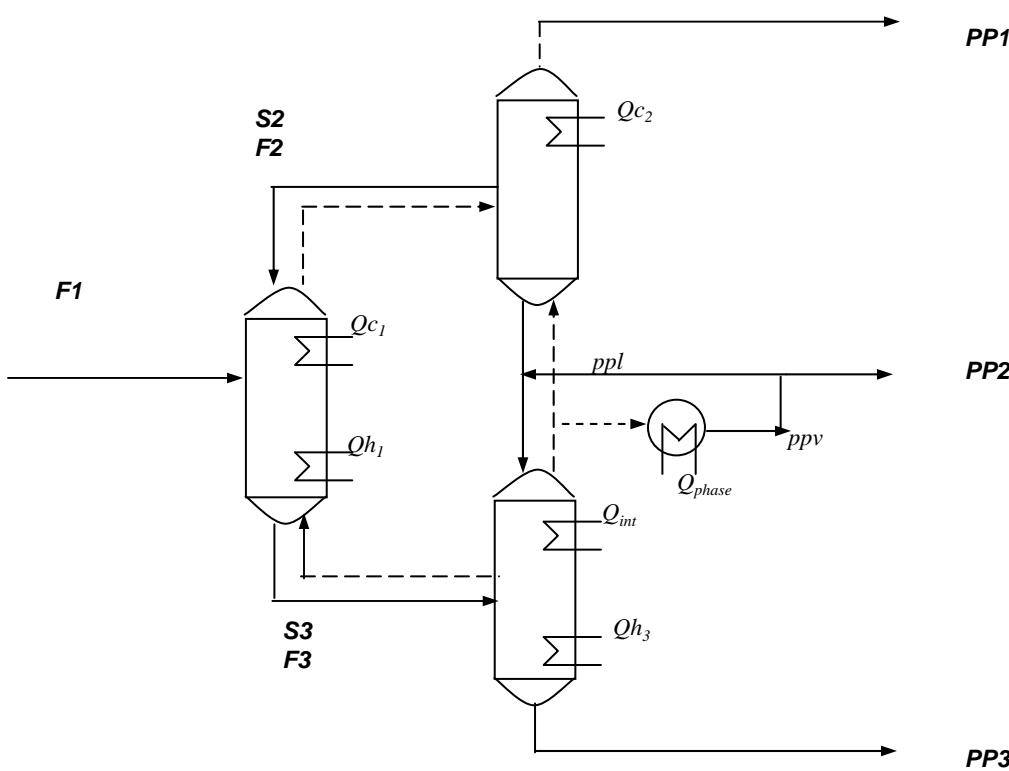


Azeotropic Mixture

3.- Reversible distillation columns and sequences.

Numerical Examples

Models implemented and solved in **GAMS** employing **DICOPT**
CONOPT.



3.- Reversible distillation columns and sequences.

Stream	Flow rate (mol/sec)		Composition		Entering stage number for
	Theoretical	Our Model	Theoretical	Our Model	
Feed: n-pentane, n-hexane, n-heptane 0.3/0.5/0.2 Stages: col1 30, col2 80, col3 50					
<i>F2</i>	5.34	5.36	0.6355/0.364/0	0.634/0.3652/6e-4	40
<i>F3</i>	8.16	8.08	0/0.714/0.286	4e-3/0.71/0.2856	26
<i>S2</i>	1.05	1.15	0.375/0.625/0	0.373/0.6257/9e-4	1
<i>S3</i>	2.45	2.29	0/0.864/0.1356	1e-3/0.863/0.1356	30
<i>L2^v</i>	1.88	1.77	0/1/0	2e-3/0.997/7e-4	80
<i>L2^l</i>	2.49	2.4	0/1/0	2e-3/0.997/7e-4	1
<i>ppl</i>	0.67	0.60	0/1/0	2e-3/0.997/7e-4	-
<i>ppv</i>	4.33	4.42	0/1/0	2e-3/0.997/7e-4	-

3.- Reversible distillation columns and sequences.

Product	Flow rate (mol/sec)	Composition
<i>PP1</i>	2.974	0.99/6e-5/0
<i>PP2</i>	5.03	5e-3/0.994/3e-4
<i>PP3</i>	1.998	0/1e-4/0.999

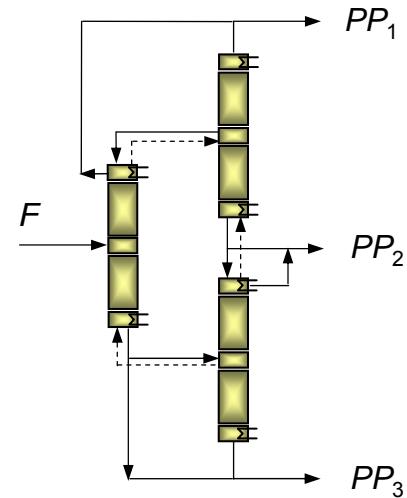
Heat Duty	Energy (KJ/sec)	
	Theoretical	Our Model
<i>Qctot1</i>	-16.63	-18.29
<i>Qhtot1</i>	107.21	114.57
<i>Qctot2</i>	-204.3	-203
<i>Qhtot2</i>	-	1
<i>Qctot3</i>	-50.27	-57.81
<i>Qhtot3</i>	293.4	313.78

4.- Optimal synthesis distillation columns sequences.

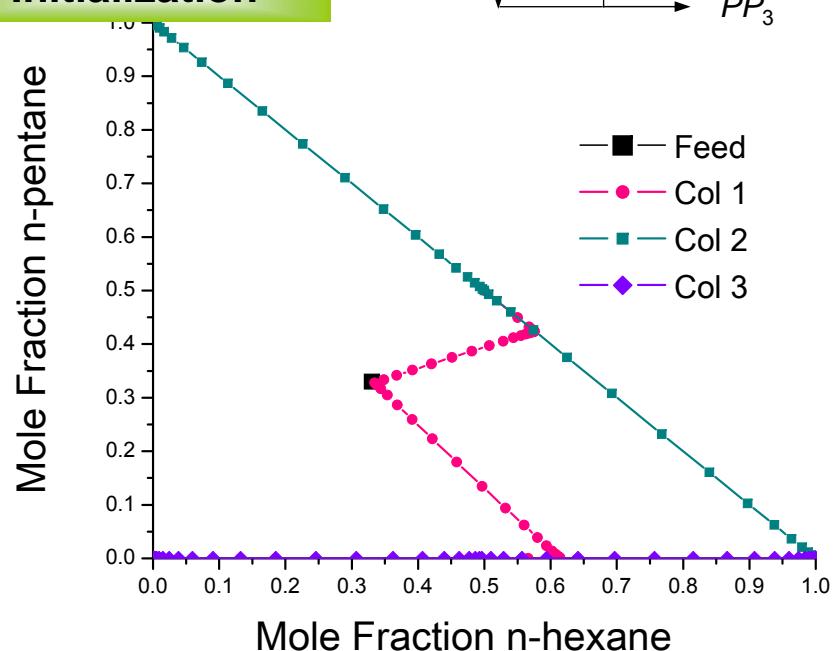
Problem specs

Mixture: N-pentane/ N-hexane/ N-heptane
Feed composition: 0.33/ 0.33/ 0.34
Feed: 10 moles/s
Pressure: 1 atm
Max no trays: 15 (each section)
Min purity: 98%
Ideal thermodynamics

Superstructure



Initialization



NLP Model

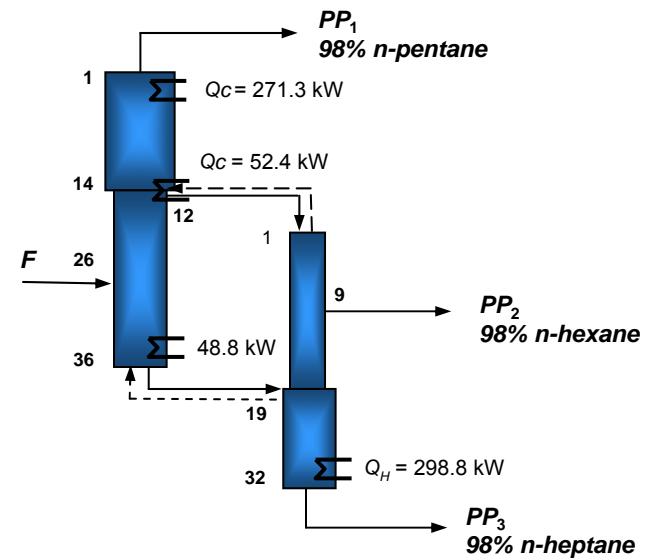
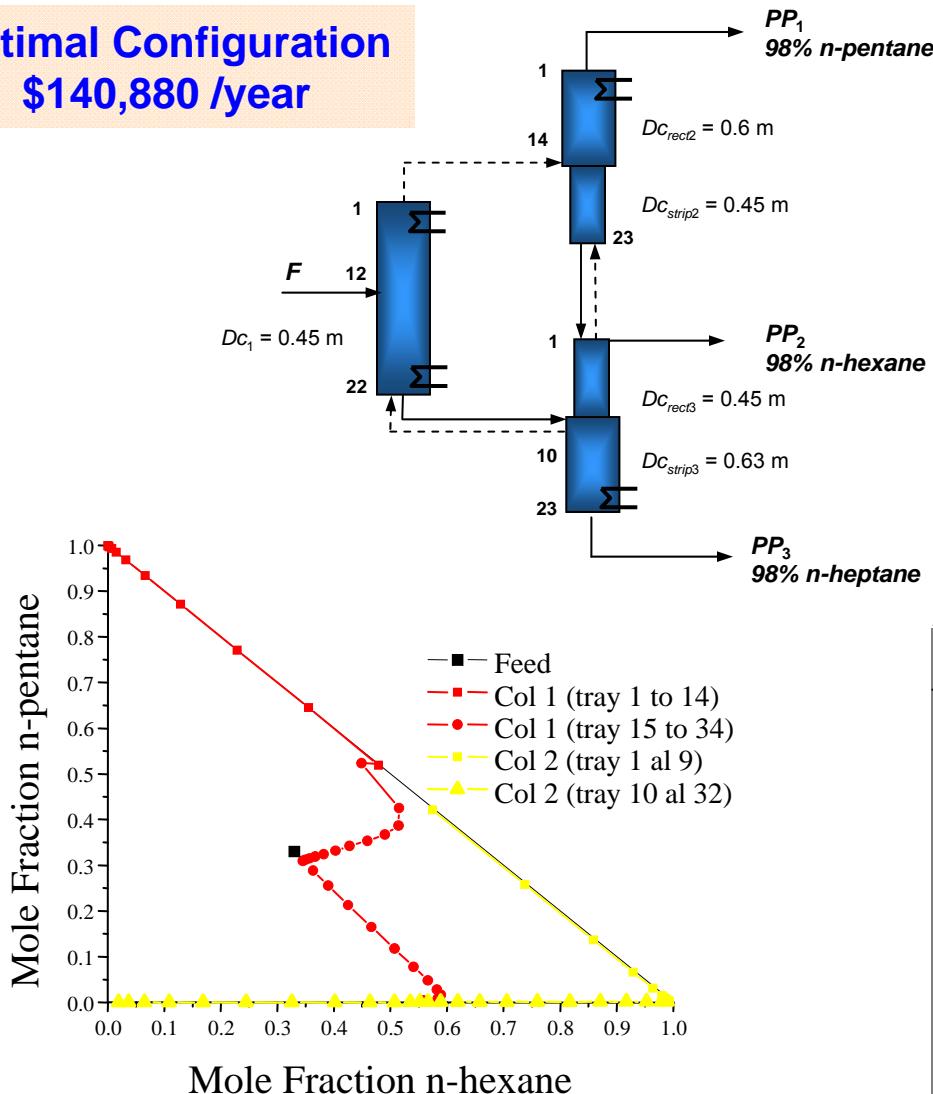
Continuous Variables	3301
Constraints	3230

MILP Model

Continuous Variables	15000
Discrete Variables	96
Constraints	8000

4.- Optimal synthesis distillation columns sequences.

**Optimal Configuration
\$140,880 /year**

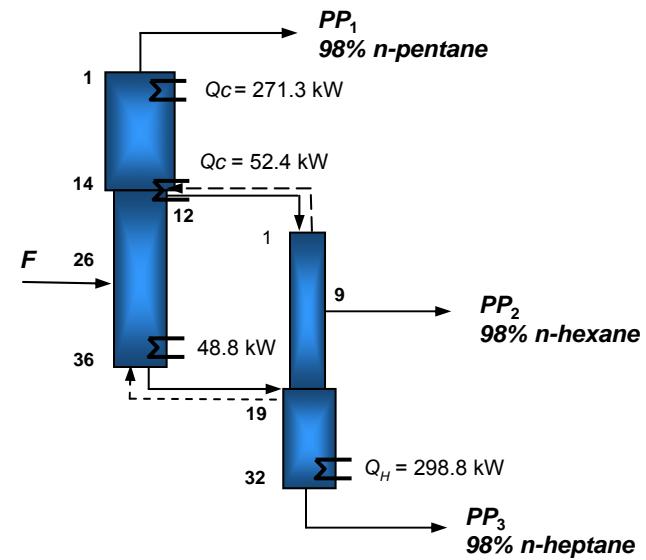
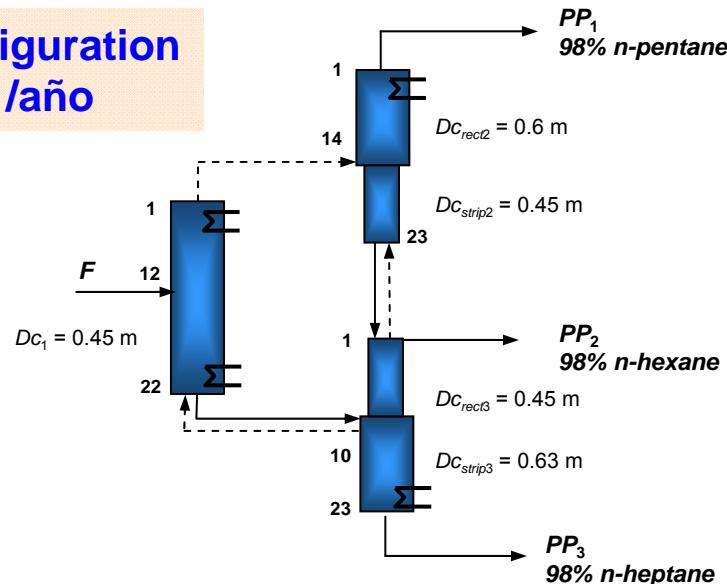


Optimal Design

Annual cost (\$/year)	140,880
Preprocessing(min)	2.20
Subproblems NLP (min)	6.97
Subproblems MILP (min)	2.29
Iterations	5
Total solution time (min)	11.46

4.- Optimal synthesis distillation columns sequences.

**Optimal Configuration
\$140,880 /año**



Side-Rectifier \$143,440 /year

Direct Sequence \$145,040 /year

Optimal Design

Annual cost (\$/year)	140,880
Preprocessing(min)	2.20
Subproblems NLP (min)	6.97
Subproblems MILP (min)	2.29
Iterations	5
Total solution time (min)	11.46

4.- Optimal synthesis distillation columns sequences.

Problem specifications
(Yeomans y Grossmann, 2000)

Mixture: **Methanol/ Ethanol/ Water**

Feed composition: 0.5 / 0.3 / 0.2

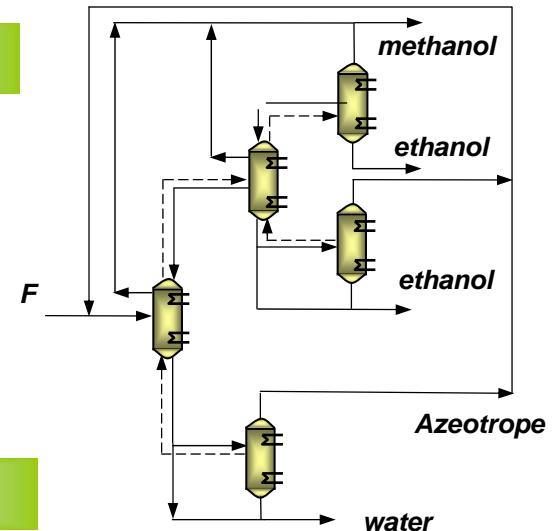
Feed flowrate: 10 moles/s

Pressure: 1 atm

Initial trays: 20 (per section)

Purity specification: 90%

Superstructure



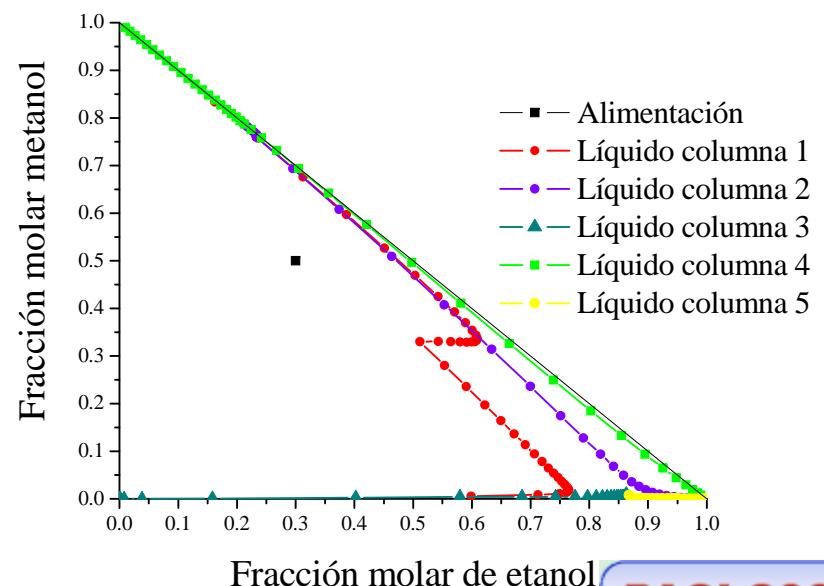
NLP Model

Continuous Variables	9025
Constraints	8996
Nonlinear non-zeroes	18230

MILP Model

Continuous Variables	12850
Discrete Variables	96
Constraints	18000

Initialization

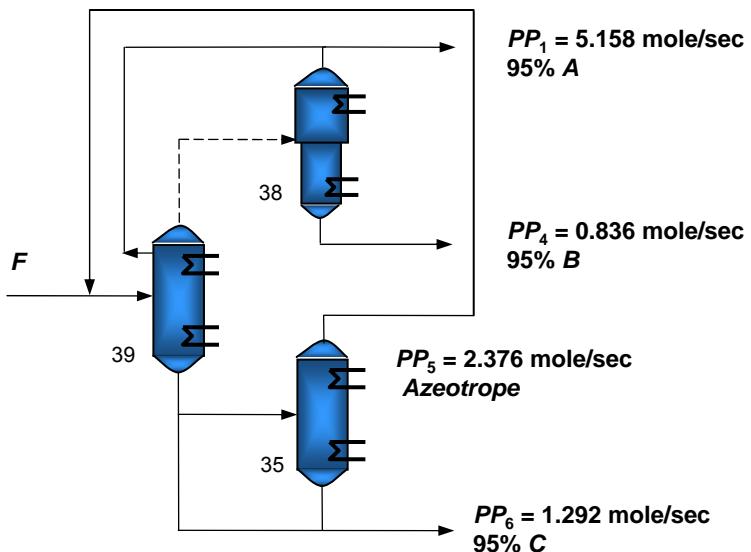
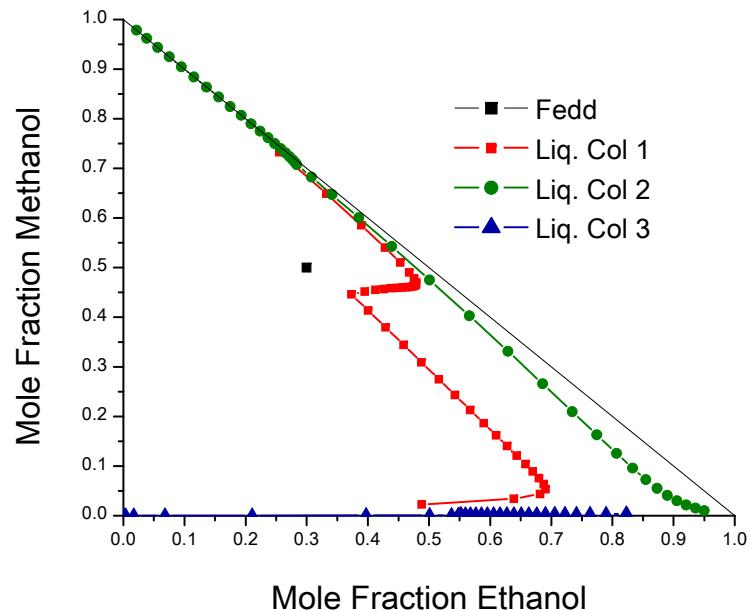


4.- Optimal synthesis distillation columns sequences.

Product Specifications 95%

Optimal Configuration
\$318,400 /año

Profiles Optimal Configuration



Optimal Solution

Annual Cost (\$/year)	318,400
Preprocessing (min)	6.05
Subproblems NLP (min)	36.3
Subproblems MILP (min)	3.70
Iteraciones	3
Total Solution Time (min)	46.01

Conclusions

Distillation optimization with rigorous models remains major computational challenge

Optimal feed tray and number of trays problems are solvable

Keys: Initialization, MINLP/GDP models

Synthesis of complex columns remains non-trivial
Progress with initialization, GDP, decomposition

Improvements potential in distillation processes