Green Chemistry in Process Engineering



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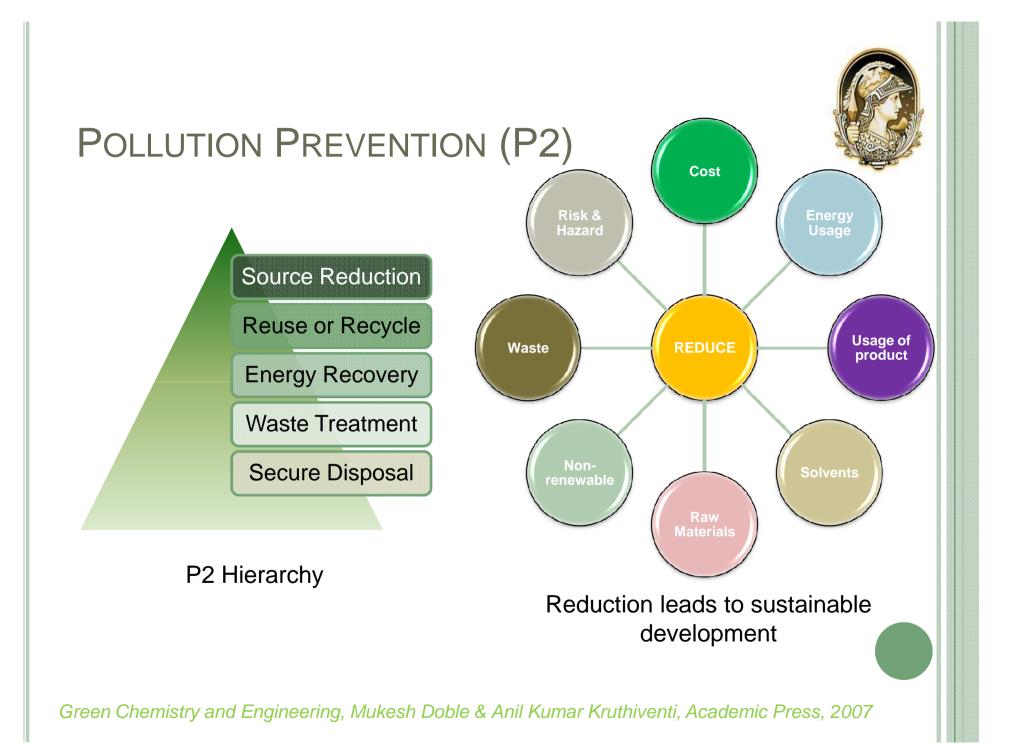
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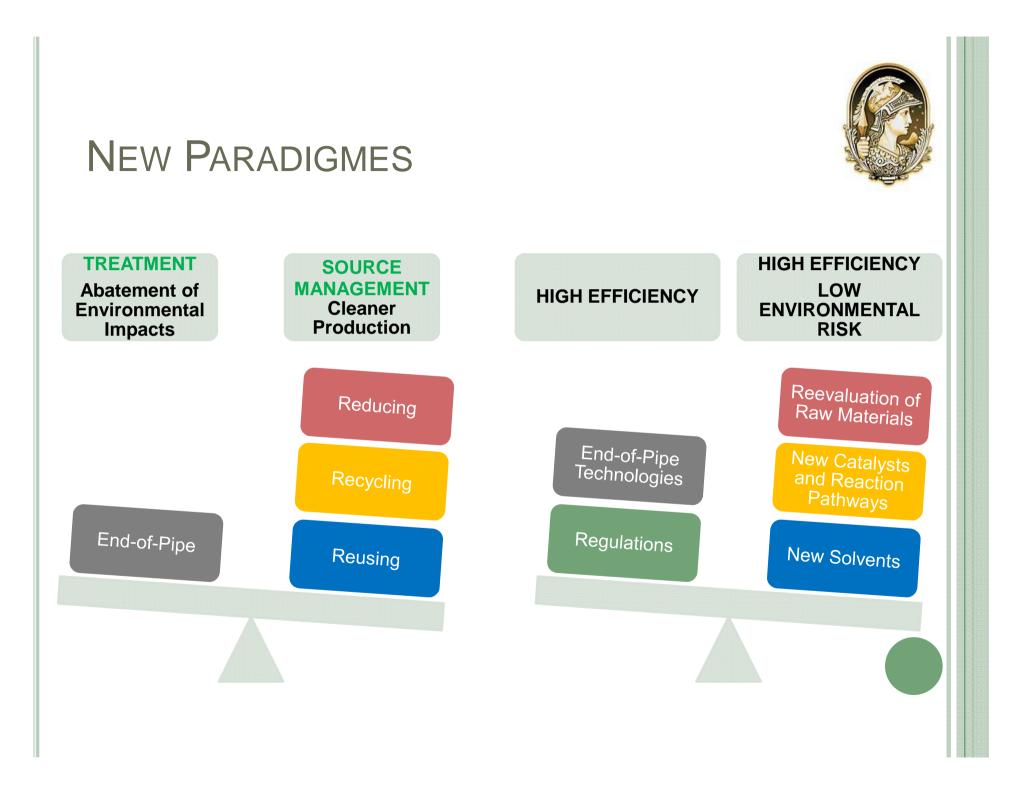
PROCESS SYNTHESIS AND SUSTAINABLE DEVELOPMENT

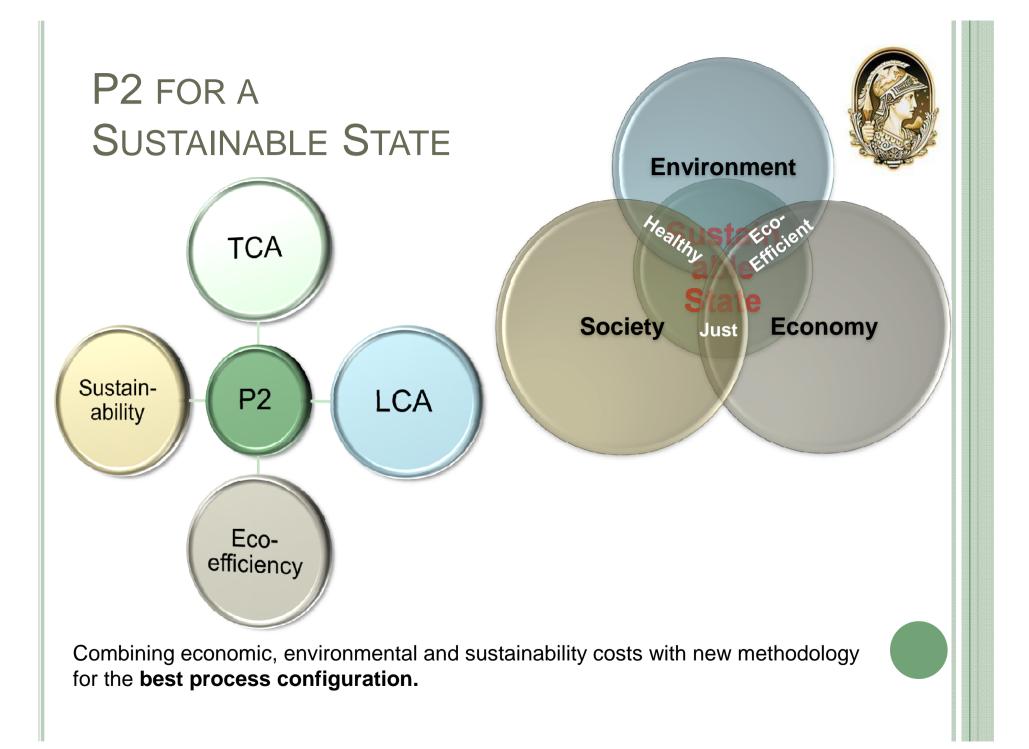


- Process synthesis is a systematic approach to the selection among potentially profitable process alternatives.
- **Process design aims for Sustainable Development**, the concept that development should meet the needs of the present without sacrificing the ability of the future to meet its needs.
- Process evaluation for process synthesis decision making:





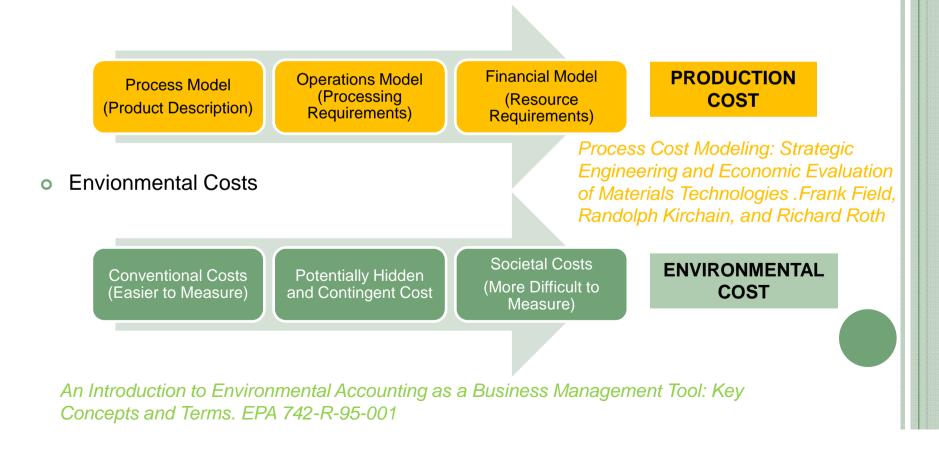


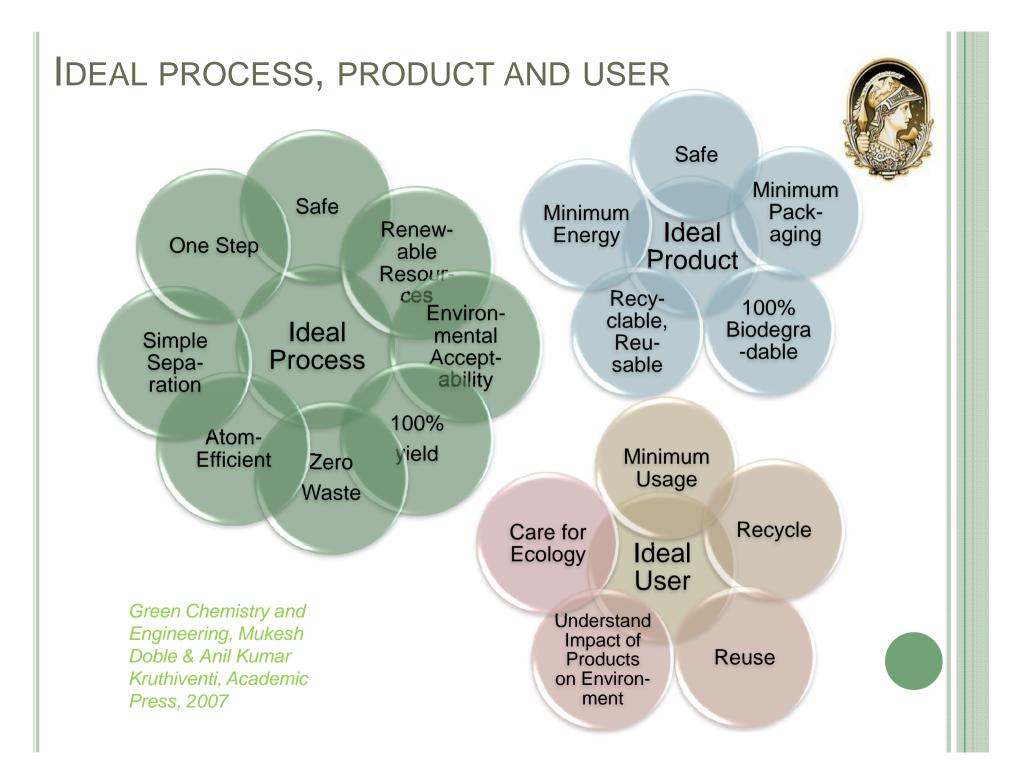


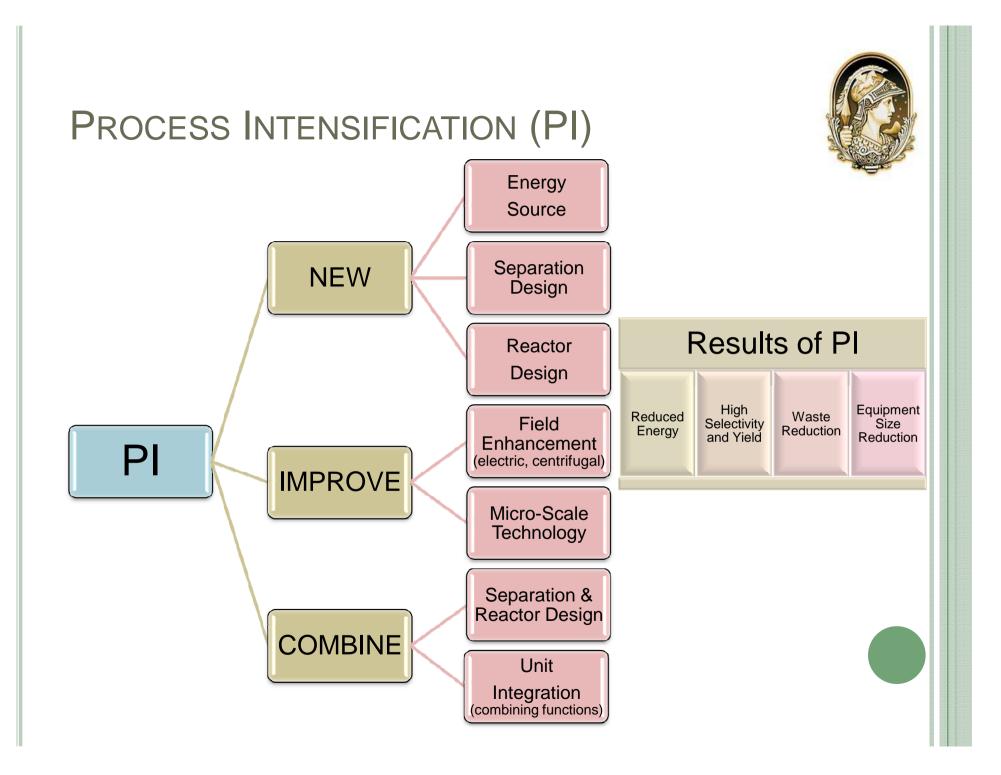


BEST PROCESS CONFIGURATION (DESIGN)

- Production cost is a central performance metric for engineering analysis, throughout the product development cycle.
- The key to good design lies in the conceptual framework that the designer employs to relate a design's properties to the design goals.







TRENDS IN PROCESS DESIGN

ECO-EFFICIENCY:

minimizing waste, pollution and natural resource depletion (concept of P2).



DESIGN FOR

ENVIRONMENT: The

systematic consideration during design of issues associated with environmental safety and health over the entire product life cycle

INDUSTRIAL

ECOLOGY: designing and operating industrial systems, where wastes or byproducts from one facility provide feedstock for other facilities.

HIERARCHY OF CONCEPTUAL CONCEPTS RELATING TO SUSTAINABILITY



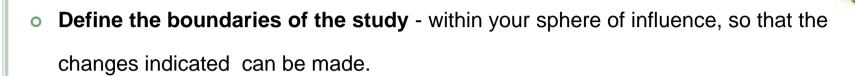
Sustainable Development/ Sustainability

Cleaner Production, Industrial Ecology, Natural Step, P2, Triple-Bottom-Line Life-Cycle Approaches for Assessing Green Chemistry Technologies, Rebecca L. Lankey, and Paul T. Anastas, Ind. Eng. Chem. Res. 2002, 41, 4498-4502

Dematerialization: Design for Disassembly, Design for Environment, Design for Recycling, etc; Eco-Industrial Parks, Full-Cost Accounting, Green Chemistry and Engineering, Life Cycle Assessment.

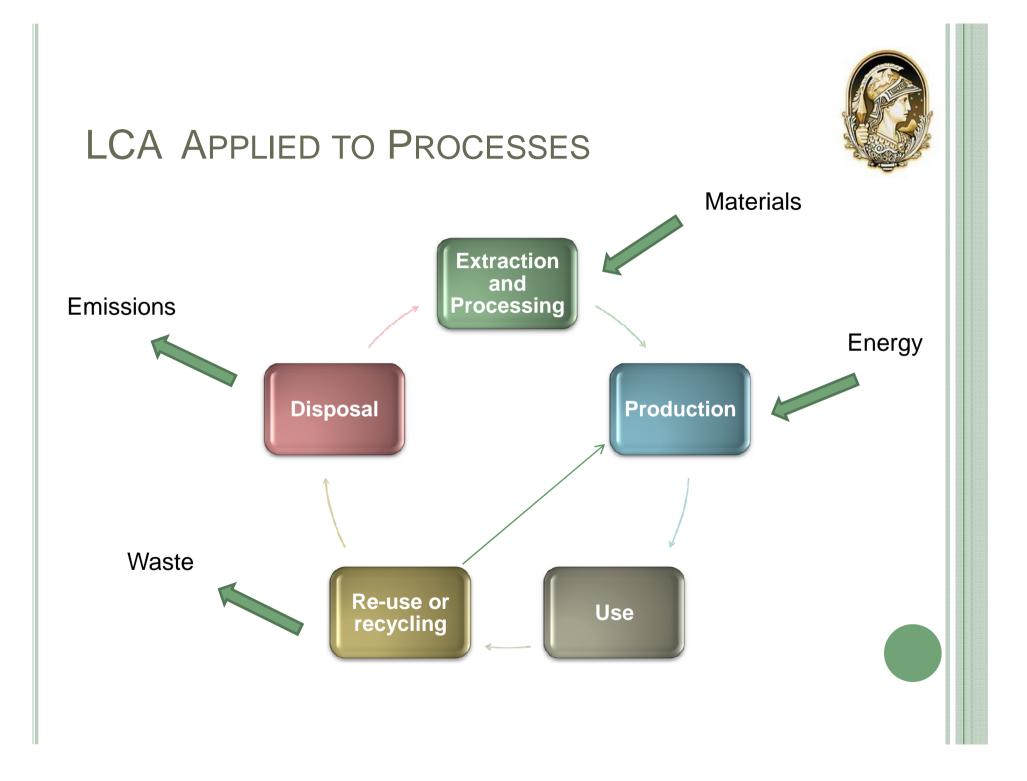
PRACTICAL

LCA FOR ASSESSING GREEN CHEMISTRY



- Metrics should be specific and detailed enough to provide useful information but simple enough to address the environmental issues within a useful time frame.
- **Desired metrics for LCA include:** (1) amounts of inputs, (2) emissions; (3) relative toxicities of materials; (4) process or product costs; (5) use of recycled material (waste or byproduct used as an input); and percentage of waste produced.
- Assessing the life-cycle impacts of a product or process and assigning metrics for the comparison of two options allow to identify where environmental vulnerabilities occur over the life cycle.

Life-Cycle Approaches for Assessing Green Chemistry Technologies, Rebecca L. Lankey, and Paul T. Anastas, Ind. Eng. Chem. Res. 2002, 41, 4498-4502





12 PRINCIPLES OF GREEN CHEMISTRY

1. Prevention

- 2. Atom economy
- 3. Less hazardous chemical synthesis
- 4. Design safer chemicals
- 5. Safety solvents and auxiliaries
- 6. Design for energy efficiency

- 7. Use renewable feedstocks
- 8. Reduce derivatives
- 9. Catalysis
- 10. Design for degradation
- 11. Real-time analysis for pollution prevention
- 12. Inherently safer chemistry for accident prevention

EPA Green Chemistry Mission: "To promote innovative chemical technologies that reduce or eliminate the use or generation of hazardous substances in the design, manufacture and use of chemical substances"

- a. Green Chemistry is the application of P2 principles to the chemistry discipline;
- b. Emphasis of Green Chemistry tends to be on synthesis routes and solvent selection, ignoring the role of equipment engineering



12 PRINCIPLES OF GREEN ENGINEERING

- 1. All material and energy inputs and outputs are as inherently non-hazardous as possible
- 2. Prevention Instead of Treatment
- 3. Design for Separation and Purification
- 4. Maximize efficiencies (Le Chatelier's Principle)
- 5. Output-Pulled Versus Input Pushed
- 6. Conserve Complexity

- 7. Durability Rather than Immortality
- 8. Meet Need, Minimize Excess
- 9. Minimize Material Diversity
- 10. Integrate Material and Energy Flows
- 11. Design for Commercial "Afterlife"
- 12. Renewable Rather than Depleting

Anastas, P.T., and Zimmerman, J.B., "Design through the Twelve Principles, Principles of Green Engineering", Env. Sci. and Tech., 37, 5, 95 -101, 2003.

PROCESS ALTERNATIVES UNDER GC AND GE PERSPECTIVES



- Increase the integration of process chemistry into the generation of design alternatives.
- Predict by-products and emissions.
- Recognize opportunities to match waste streams with feed streams.
- Link process and environmental models (environmental databanks and process simulators).
- Detail used in process models should match the accuracy needed to make decisions.
- Allocate environmental impacts to specific processes and products in plants.
- Develop environmental impact indexes.
- Define preferences needed to weight multi-objective optimization.
- Sensitivity analysis and identification of the features that drive environmental impact.

J. A. Cano-Ruiz and G. J. McRae, ENVIRONMENTALLY CONSCIOUS CHEMICAL PROCESS DESIGN, Annu. Rev. Energy Environ. 1998. 23:499–536

ENVIRONMENTAL IMPACT ASSESSMENT BASED ON RISKS



• **Risk** is a combination of the probability that an adverse event will occur and the consequences of the adverse event. Process designer should identify, evaluate, select and implement actions to reduce risk to human health and to ecosystems.

Risk = *f*(*hazard*, *exposure*)

- **Hazard** is the potential for a substance or situation to cause harm or to create adverse impacts on persons or the environment. The magnitude of the hazard reflects the potential adverse consequences.
- **Exposure** denotes the magnitude and the length of time the organism is in contact with an environmental contaminant.

QUALITIES OF SUCCESSFUL METRICS

Efficient

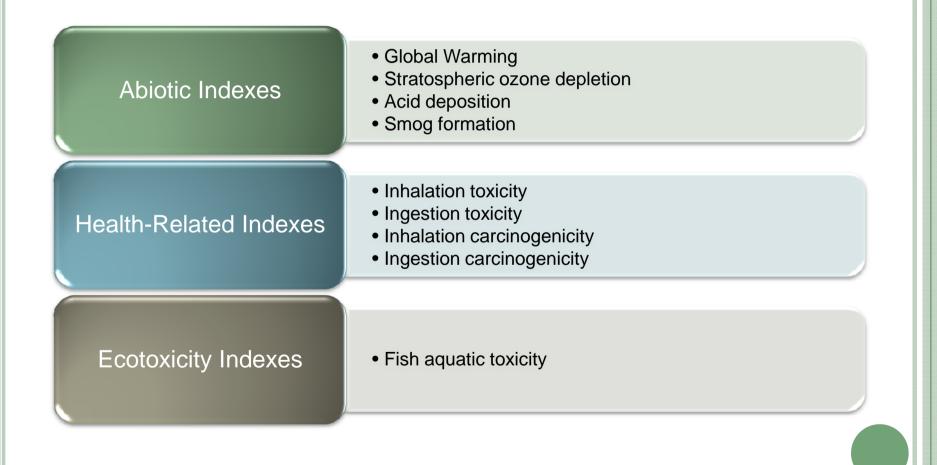
(Few, simple, robust, easy to collect, calculate and understand)

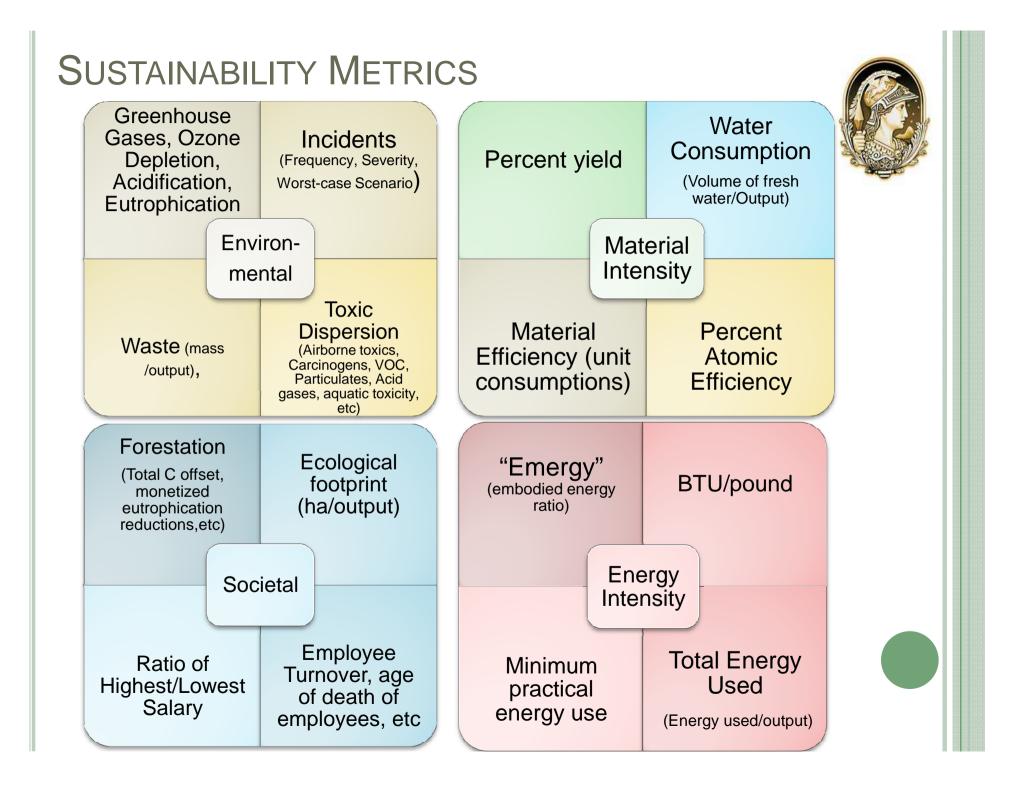
Normalizable

(for priorization and comparison)

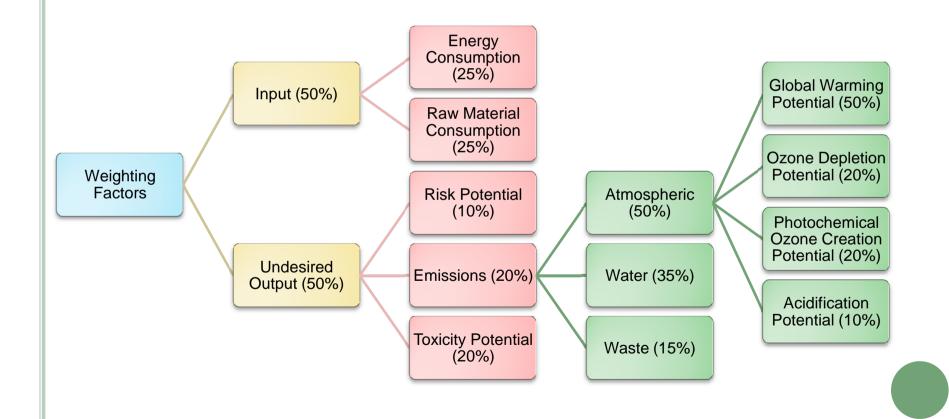
Business and Environmental Value (Growth of business and environmental quality)

METRICS OF ENVIRONMENTAL RISKS FOR FLOWSHEET EVALUATION





SUSTAINABILITY METRICS – WEIGHTING FACTORS



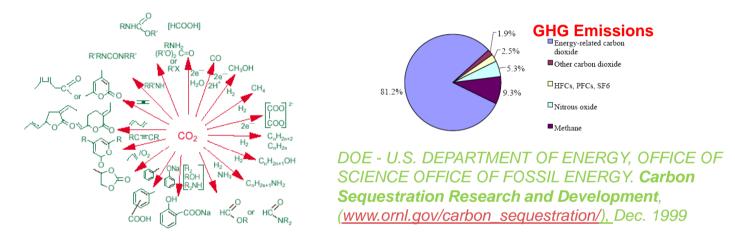


CASE STUDY CHEMICAL SEQUESTRATION OF CO₂



CO_2 SEQUESTRATION

• Carbon dioxide use as a raw material (production of urea, methanol, DMC, plastics, etc).

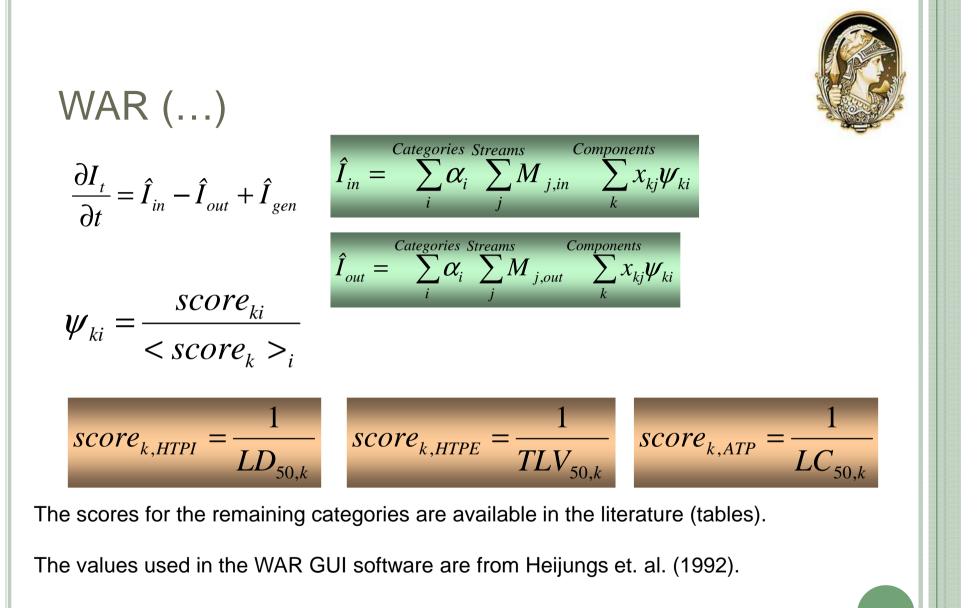


- Processes should overcome challenges of economics, performance, and associated environmental impacts;
- Most commercial plants capturing CO₂ from power plant flue gas use is based on chemical absorption with monoethanolamine (MEA) solvent (\$41/t CO₂).



WASTE REDUCTION ALGORITHM (WAR)

- WAR was selected to calculate the EI of DMC production in each route.
- It characterizes sustainability with an index that measures Potential Environmental Impacts (PEI), meaning it works with hazards rather than risks.
- Although the algorithm defines seven hazard categories, only three of them were taken into consideration in the present work: human toxicity potential by ingestion (HTPI), human toxicity potential by inhalation or dermal exposure (HTPE), and aquatic toxicity potential (ATP).
- The scores for these categories measured using easily obtainable data (LD₅₀, TLV₅₀ e LC₅₀).
- Terrestrial toxicity potential is also measured using LD₅₀ being, therefore, proportional to HTPI.
- Global warming potential (GWP) of both routes would be negative, since both routes sequestrate CO₂ (and produce no carbon-equivalent substances).
- Additionally, none of the chemicals involved appear in Ozone Depletion Potential (ODP) or Acidification Potential (AP).
- The only chemical which has Photochemical Oxidation Potential (PCOP) in both routes is methanol. But methanol consumption of the routes can be directly compared.



Heijungs, R. et al; Life Cycle Assessment; United Nations Environment Program - UNEP, Paris, France (1996)

HUMAN HEALTH AND ENVIRONMENTAL IMPACT INDEXES



Dimensionless Risk Index	Eqn. #	Equation	Parameter / Software Source(s)
Ingestion Route Toxicity Potential	1	$INGTP_i = \frac{C_{i,a}/RfD_i}{C_{Toluene,a}/RfD_{Toluene}}$	C _{i,a} & C _{Toluene,a} - Mackay Model, 1992-4; RfD i & RfD _{Toluene} – EPA 1994, 1997
Inhalation Route Toxicity Potential	2	$\text{INHTP}_{i} = \frac{\text{C}_{i,a}/\text{RfC}_{i}}{\text{C}_{\text{Toluene,a}}/\text{RfC}_{\text{Toluene}}}$	<i>C_{i,a} & C_{Toluene,a} –</i> Mackay Model, 1992-4; RfC <i>i</i> & RfC <i>Toluene</i> – EPA 1994, 1997
Ingestion Route Carcinogenicity Potential	3	$INGCP_{i} = \frac{C_{i,w} \times (SF_{i})_{ING}}{C_{Benzene,w} \times (SF_{Benzene})_{ING}}$	C _{i,w} & C _{Benzene,w} - Mackay Model, 1992-4; SF- EPA 1994, 1997
Inhalation Route Carcinogenicity Potential	4	$INHCP_{i} = \frac{C_{i,a} \times (SF_{i})_{INH}}{C_{Benzene,a} \times (SF_{Benzene})_{INH}}$	C _{i,w} & C _{Benzene,w} - Mackay Model, 1992-4; SF- EPA 1994, 1997
Fish Toxicity Potential	5	$FTP_{i} = \frac{C_{i,w} \times LC_{50f,PCP}}{C_{PCP,w} \times LC_{50f,i}}$	<i>C_{i,w} & C_{PCP,w}</i> – Mackay Model, 1992-4; <i>LC</i> 50 <i>f</i> - Verschueren, 1996; Davis, 1994

C_i is the concentration if species "i".

RfD is the reference dose; LD50 may be substituted for RfD.

RfC is the reference concentration, LC50 may be substituted for RfC, and Hazard Value (HV, Davis 1994) may be substituted for SF (slope factor).

HUMAN HEALTH AND ENVIRONMENTAL IMPACT INDEXES



Dimensionless	Eqn.	Equation	Parameter / Software
Risk Index	- #		Source(s)
Global Warming	6	GWP_i	<i>GWP</i> - Fisher, 1990a; WMO, 1992a; IPCC, 1991, 1996
	ба	$GWP_i = N_C \times \frac{MW_{CO_2}}{MW_i}$	N _C -
Ozone Depletion	7	ODP _i	<i>ODP</i> - Fisher, 1990b; WMO, 1990a; WMO 1992b
Smog Formation	8	$SFP_i = \frac{MIR_i}{MIR_{ROG}}$	MIR- Carter, 1994; Heijungs, 1992
Acid Rain	9	ARP _i	ARP- Heijungs, 1992; Goedkoop, 1995

GWP is global warming potential.

ODP is the ozone depletion potential.

MIR is the maximum incremental reactivity for forming ozone in the lower atmosphere. ARP is the acid rain potential.

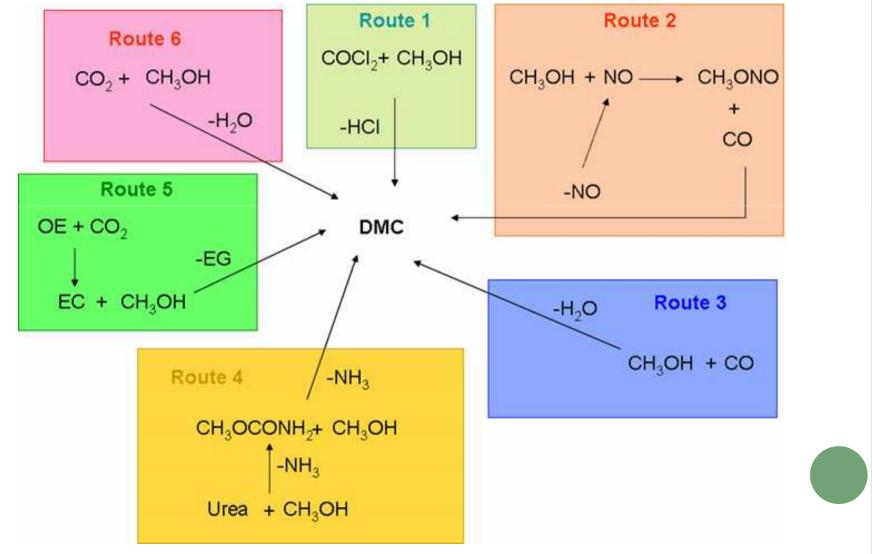


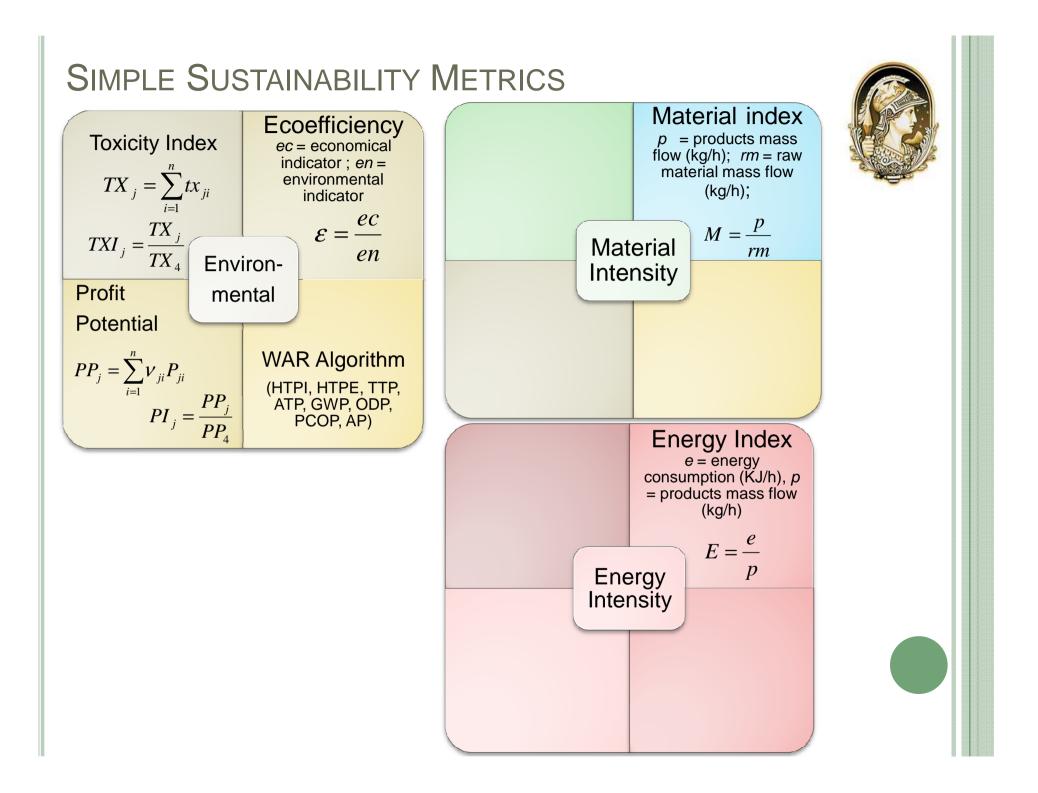
DMC PRODUCTION

- DMC market is broadening and it is moving to the category of chemical commodity: DMC can be used, for example, as alkylation agent, gas or diesel additive and as a monomer in polycarbonate synthesis
- Exploratory Analysis of six routes for DMC production (three of which have sequestration potential), briefly described as:
 - <u>ROUTE 1</u>: production of DMC and co-production of HCl from methanol and phosgene
 - <u>ROUTE 2</u>: production of methyl nitrite from methanol and NO, followed by production of DMC from methyl nitrite and CO, recovering NO
 - **<u>ROUTE 3</u>**: production of DMC and water from CO and methanol
 - <u>ROUTE 4</u>: production of DMC and NH₃ from urea and methanol (urea production involves CO₂ sequestration)
 - **<u>ROUTE 5</u>**: production of DMC and ethylene glycol from ethylene oxide and CO₂
 - **<u>ROUTE 6</u>**: production of DMC and water from CO₂ and methanol
- ROUTES 4, 5 and 6 show CO₂ sequestration potential



ROUTES FOR DMC PRODUCTION





PROFIT POTENTIAL FOR RANKING ROUTES

Chemical	Route 1	Route 2	Route 3	Route 4	Route 5	Route 6
Stoichiometric coefficients (gate-to-gate domain) v_{ii}						
Hydrochloric acid	2					Л
Water		1	1			1
Ammonium				1		
Ethylene carbonate					0	
Dimethyl carbonate	1	1	1	1	1	1
Carbon dioxide					-1	-1
Ethylene glycol					1	
Phosgene	-1					
Methanol	-2	-2	-2	-3	-2	-2
Methyl carbamate				0		
Carbon monoxide		-1	-1			
Methyl nitrite		0				
Ethylene oxide					-1	
Nitric oxide		1				
Oxygen		- 1/2	- 1/2			
Urea				-1		
	Cra	dle-to-ga	te domai	in		
Carbon	•	•	•	•	•	•
Sodium chloride	•					
Chlorine	•					
Carbon dioxide				•		
Ethane					•	
Ethylene					•	
Hydrogen	•	•	•	•	•	•
Methane	•	•	•	•	•	•
Carbon monoxide	•	•	•	•	•	•
Nitrogen				•		
Oxygen		•	•		•	

The symbol (•) was used in the cradle-to-gate domain to indicate the presence of the chemical in the route.

	Calore Calore
Chemical	P (US\$/mol)
Hydrochloric acid	0.00342
Ammonium	0.00496
Carbon credits	0.00084
Dimethyl carbonate	0.10810
Ethylene glycol	0.06238
Phosgene	0.16571
Methanol	0.01047
Carbon monoxide	0.00140
Ethylene oxide	0.05487
Nitric oxide	0.00150
Oxygen	0.00477
Urea	0.02019

Profit Potential

 v_{ji} = stoichiometric coefficient of chemical i on route j; P_{ji} = price in US\$/mol of chemical i on route j

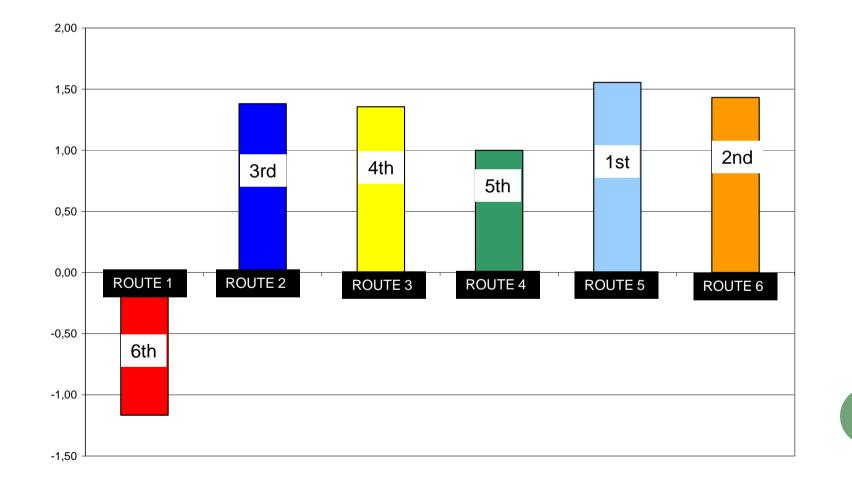
$$PP_{j} = \sum_{i=1}^{n} \mathcal{V}_{ji} P_{ji} \qquad PI_{j} = \frac{PP_{j}}{PP_{4}}$$

ANATAS, P. T. e ALLEN, D. Green Chemistry. In: ALLEN, D. T. Green Engineering: Environmentally Conscious Design of Chemical Processes. Prentice Hall PTR: New Jersey, 2002. pp 177-196.





RESULT OF EXPLORATORY ANALYSIS



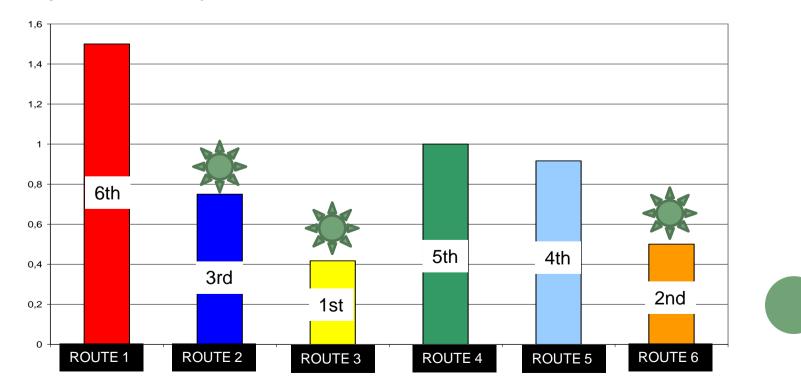


TOXICITY RANKING

• The toxicity index of each route was calculated, using ROUTE 4 as basis

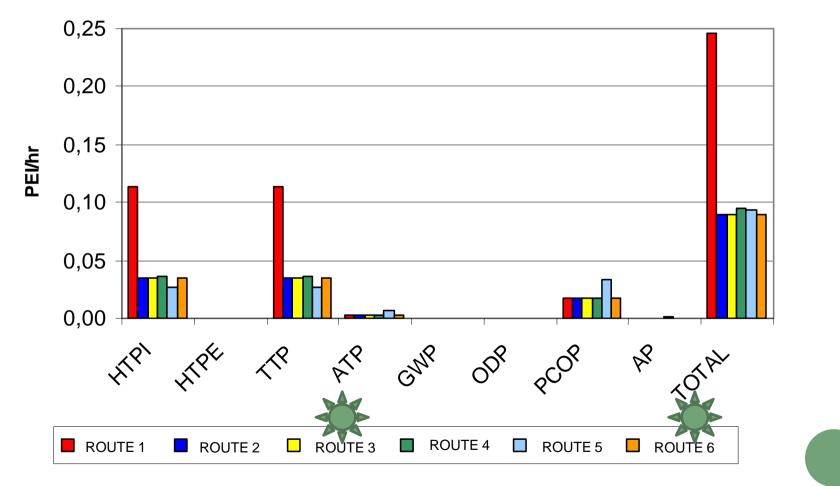
$$TX_{j} = \sum_{i=1}^{n} tx_{ji} \qquad TXI_{j} = \frac{TX_{j}}{TX_{4}}$$

where: TX_j = toxicity of route j; tx_{ji} = toxicity of chemical i on route j; TXI_j = toxicity index of route j.





WAR RANKING



The toxicity index results are in general agreement with the WAR (PEI) results



OVERALL RANKING

• Total score = sum of economical ranking position + the average of environmental ranking positions (toxicity and PEI criteria). The lower the score, the greener the route.

Route	Economical ranking	Environmental ranking		Total score
)	Toxicity	PEI	
1	6	6	6	12
2	3	3	1	5
3	4	1	1	5
4	5	5	4	9.5
5	1	4	4	5
6	2	2	1	3.5

- ROUTE 6 might be the greener route, but ROUTE 5 has a better profit potential and its total score is close to ROUTE 6's
- ROUTE 6 is eliminated as there's no indication that it is feasible in industrial scale
- ROUTES 2 and 3 have the same total score as ROUTE 5, but only intermediate profit potential.
- For CO₂ sequestration, ROUTES 2 and 3 must be abandoned.
- ROUTES 4 and 5 should be further investigated, as they combine intermediate total score, sequestration potential, and industrial feasibility.

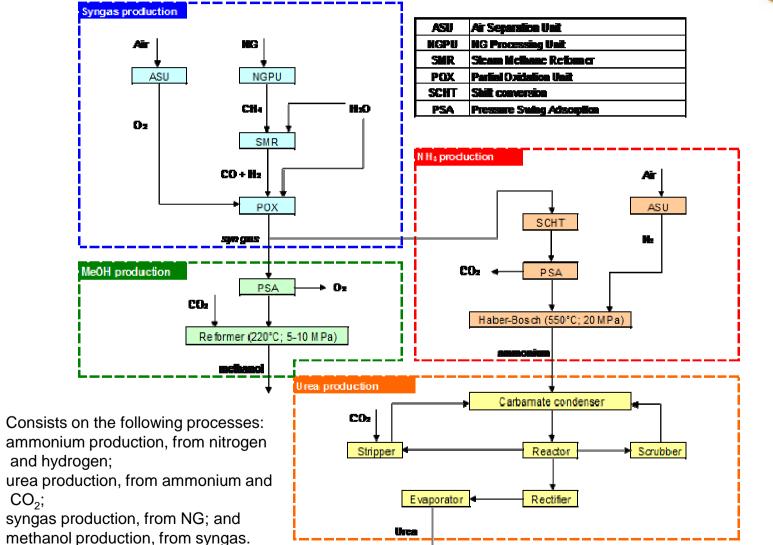
LIFE CYCLE ASSESSMENT

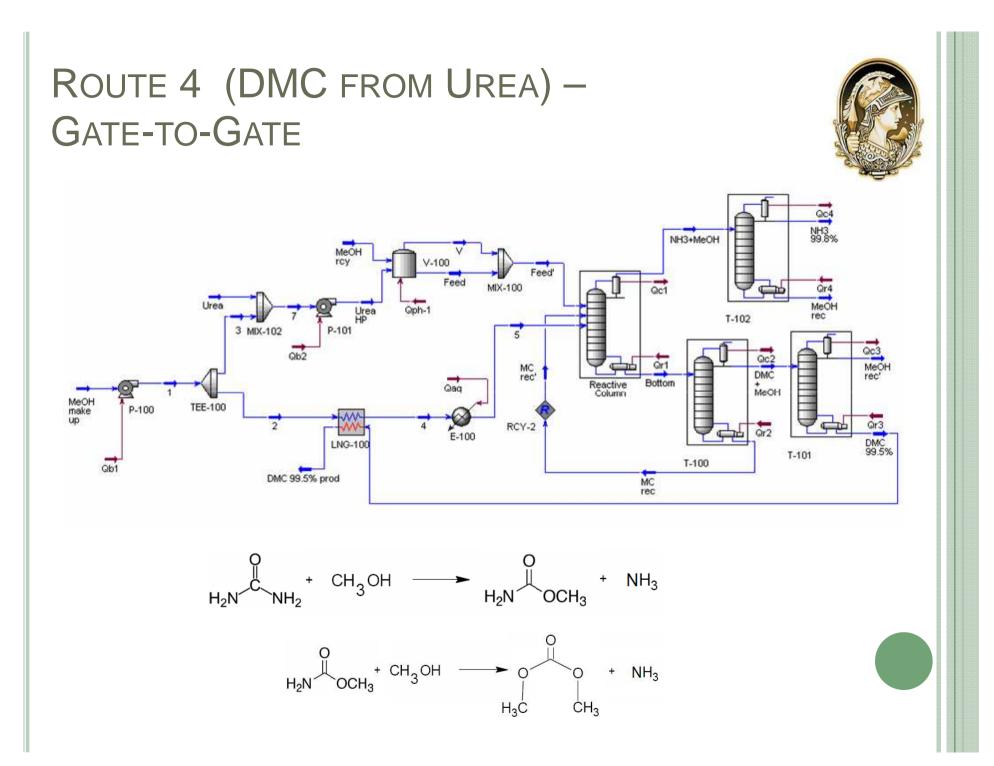


- DMC production via ROUTES 4 and 5 were conceived in two domains: **cradle-to-gate** (raw material production processes) and **gate-to-gate** (DMC production processes).
- Domain gate-to-gate is the actual industrial venture in focus, which receives raw materials produced in cradle-to-gate domain processes.
- Processes of **cradle-to-gate domain are herein seen as auxiliary**, and were addressed exclusively to allow LCA. In this sense, there are other possible processes that were not taken into consideration and could equally be used.
- For the gate-to-gate domain analysis, DMC production processes were **simulated and optimized using HYSYS** (Aspentech).
- To assess the **environmental impact of the considered routes**, the WAR algorithm, which requires streams' information, was implemented in the simulation environment.
- The cradle-to-gate domain was analyzed exclusively based on the WAR algorithm using information estimated from stoichiometric data available in literature

ROUTE 4 (DMC FROM UREA) – CRADLE-TO-GATE







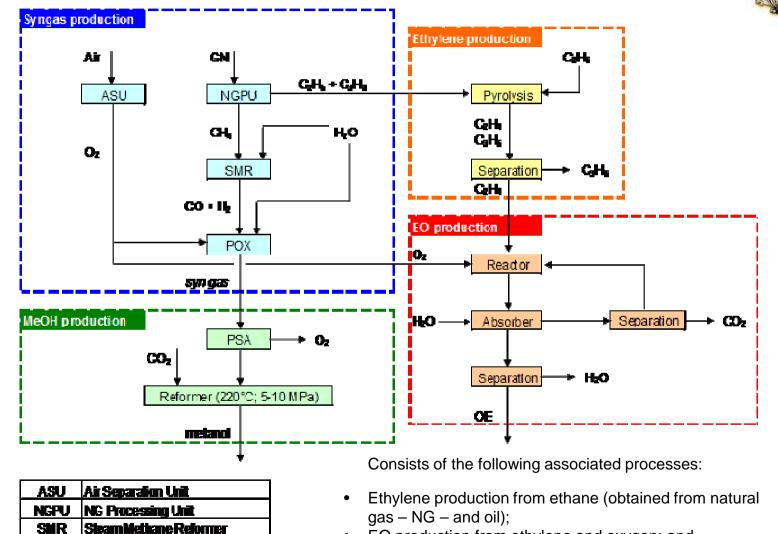
ROUTE 5 (DMC FROM EO) – CRADLE-TO-GATE

POX

PSA

Parilal Coddation Unit

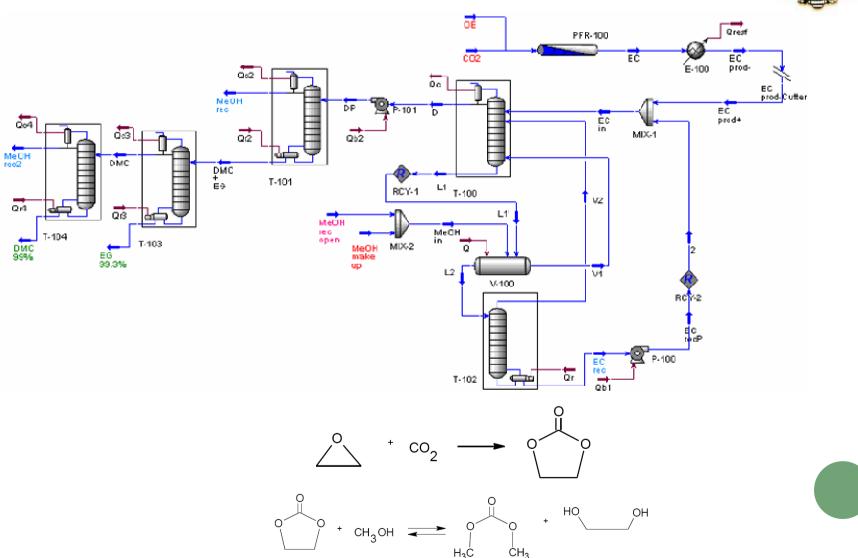
Proceure Swing Adsorption



- EO production from ethylene and oxygen; and
- Syngas production (steam methane reform SMR); methanol production from syngas

ROUTE 5 (DMC FROM EO) – GATE-TO-GATE





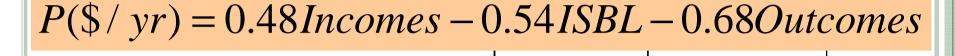


ENVIRONMENTAL + ECONOMIC FACTORS

- Objective Function for searching Optimal Design must incorporate environmental and economic factors together.
- The economic assessment uses **economic indexes** that include Total Revenue, Capital Costs and Operational Costs.
- The environmental indexes used to **quantify environmental impact** (global warming, ozone depletion, acid rain, smog formation, human-ingestion-route toxicity, human-inhalation-route toxicity, human-ingestion-route-carcinogenicity toxicity, human-inhalation-route carcinogenicity toxicity and ecotoxicity). An environmental process composite index is found (**EI**).
- A Sustainability Function (S)

 $S = \omega_P P - \omega_{EI} EI$

ECONOMIC OBJECTIVE



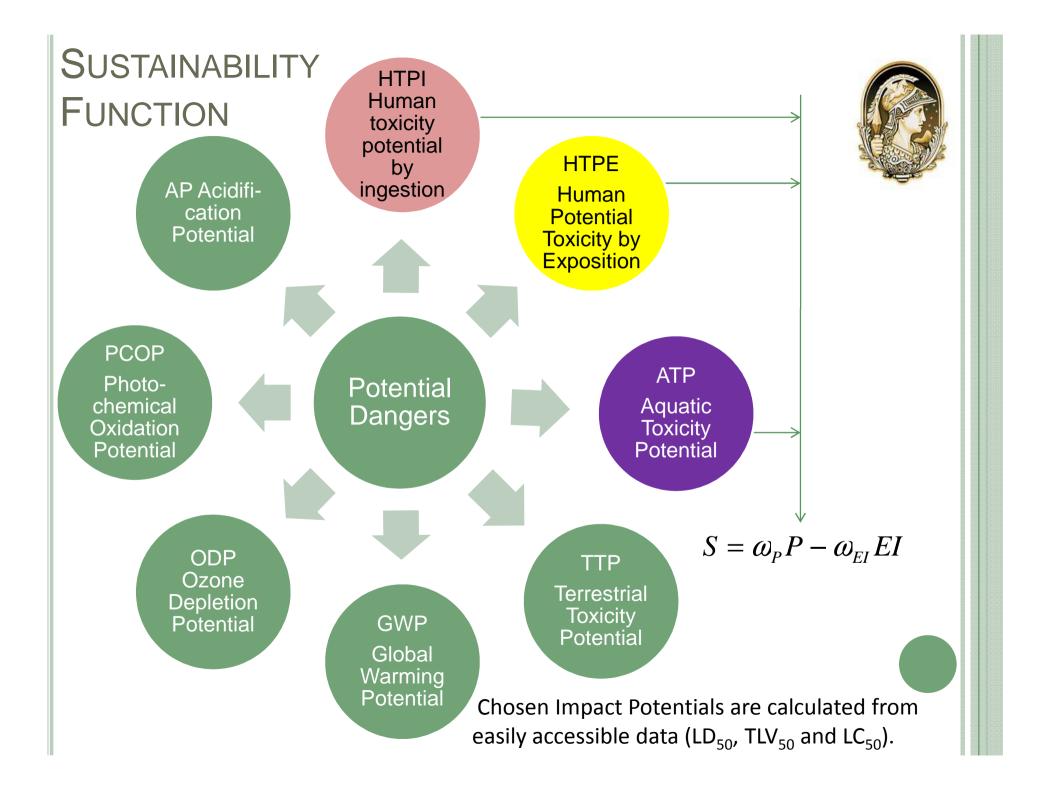
Price_{DMC}(US\$/kg)*Production_{DMC}(kg/yr) + Price_{EG}(US\$/kg)*Production_{EG}(kg/yr) + Price_{MeOH}(US\$/kg)*Recovery_{MeOH}(kg/yr) Heat exchangers, columns, vessels, reactors and pumps

 $\begin{array}{l} {\operatorname{Price}_{{\operatorname{EO}}}({\operatorname{US}}{\operatorname{/kg}})^{*}{\operatorname{Production}_{{\operatorname{EO}}}(\operatorname{kg}/\operatorname{yr})} \\ {\operatorname{+}\operatorname{Price}_{{\operatorname{MeOH}}}({\operatorname{US}}{\operatorname{/kg}})^{*}{\operatorname{Feed}_{{\operatorname{MeOH}}}(\operatorname{kg}/\operatorname{yr})} \\ {\operatorname{+}\operatorname{Price}_{{\operatorname{Water}}}({\operatorname{US}}{\operatorname{/kg}})^{*}{\operatorname{Consumption}_{{\operatorname{Water}}}(\operatorname{kg}/\operatorname{yr})} \\ {\operatorname{+}\operatorname{Cost}_{{\operatorname{Vapour}}}({\operatorname{US}}{\operatorname{/kg}})^{*}{\operatorname{Consumption}_{{\operatorname{Vapour}}}(\operatorname{kg}/\operatorname{yr})} \\ {\operatorname{+}\operatorname{Cost}_{{\operatorname{EE}}}({\operatorname{US}}{\operatorname{/kg}})^{*}{\operatorname{Consumption}_{{\operatorname{EE}}}(\operatorname{kg}/\operatorname{yr})} \end{array} \end{array}$

ISBL

Cost of Equipments (US\$)		
Heat Exchanger	$C = \frac{(M \& S)}{280} 101,3A^{0,65} (2,29 + F_c)$ F _c =0,85 A = surface of heat exchange; em ft ²	
Vessesl, colums and reactors	$C = \frac{(M \& S)}{280} 101,9D^{1,066}H^{0,802}(2,18 + F_c)$ F _c =1,00 H = height; ft ² D = diameter; ft ²	
Internals of distillation columns	$C = \frac{(M \& S)}{280} 4,7D^{1,55}H \cdot F_c$ F _c =1,00 H = height; ft ² D = diameter; ft ²	
Pumps	30.000,00* *estimated cost	



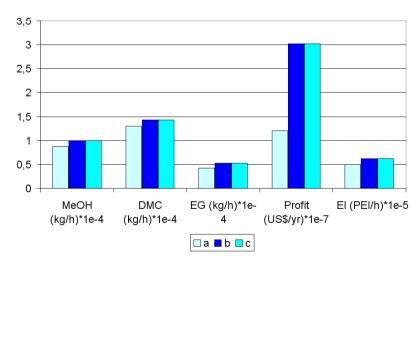




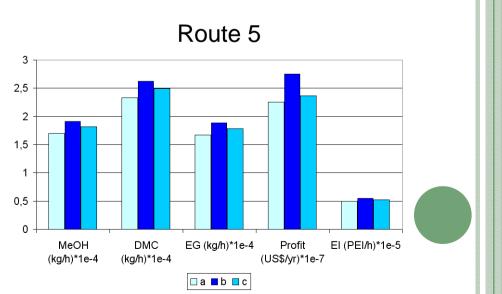
PROCESS OPTIMIZATION – GATE-TO-GATE

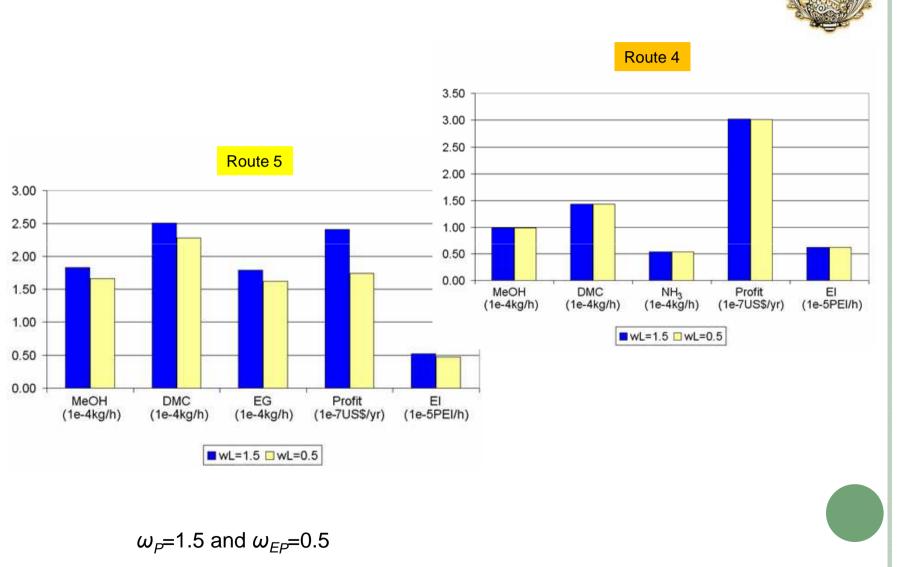
Problem Formulations

- sustainability maximization constrained by maximum EI of 50.000 PEI/h;
- sustainability maximization with $\omega_I=0$ (i.e., profit maximization);
- sustainability unconstrained maximization.



Route 4



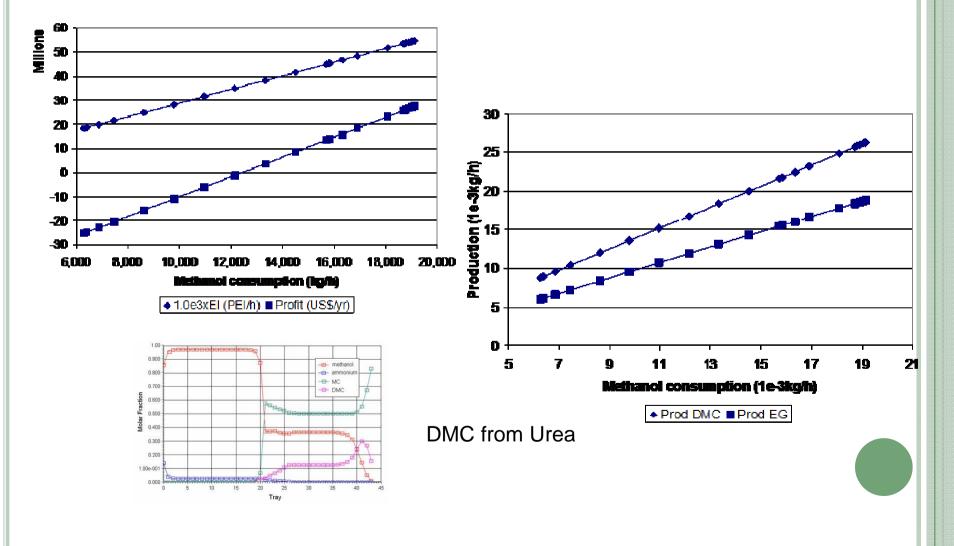


$Optimization \ of \ Sustainability$



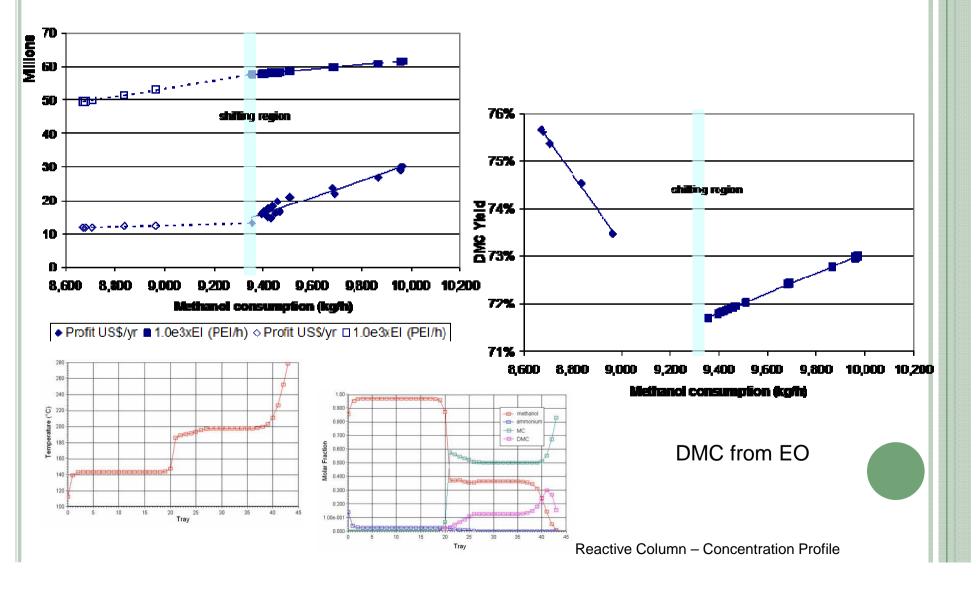


Route 4 – Gate-to-Gate





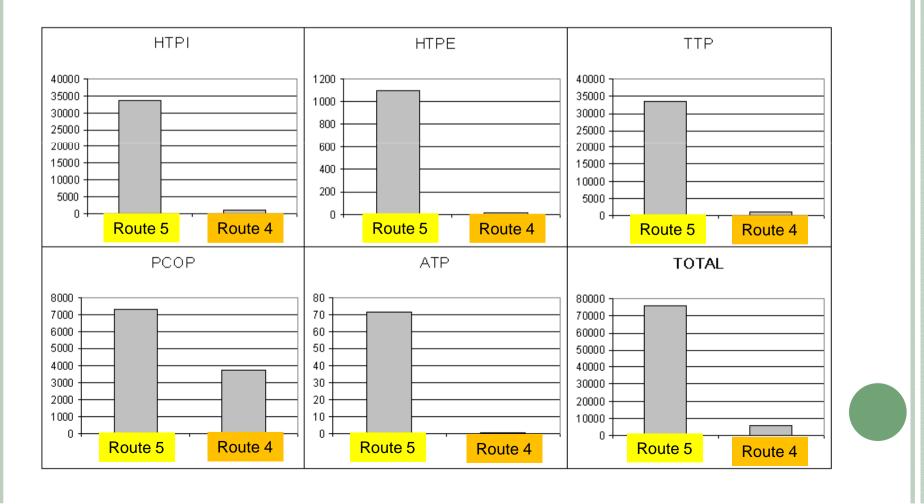
Route 5 – Gate-to-Gate





CRADLE-TO-GATE DOMAIN

Problem Formulation: WAR Algorithm





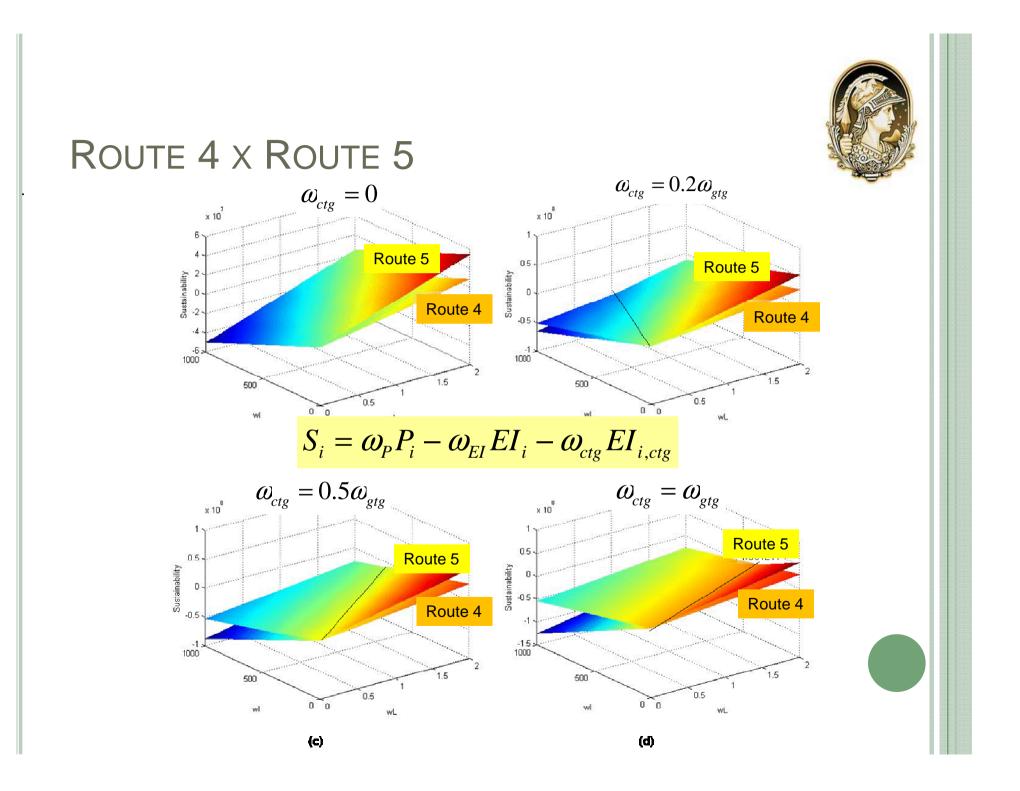
Route 4 x Route 5

	ROUTE 5	ROUTE 4
El cradle-to gate	75,600	5,740
El gate-to gate	50,000	50,000
Profit (M\$)	27.26	14.5
ω_{P}	1	
ω _{El}	100	
Sustainability (M)	14.7	8.9

	ROUTE 5		ROUTE A	
Sensitivity	ω _P =1.5	ω _P =0.5	ω _P =1.5	ω _P =0.5
EI	0.66%	-8.48%	0.006%	0.003%
				-
Profit	2.18%	-26.30%	0.013%	0.031%
		,	```	

$$\sigma = \frac{f(\omega_P = 1)}{f(\omega_P = k) - f(\omega_P = 1)}$$

Function	Importance grade			
Function	Low	Normal	High	
EI	$0 \le \omega_{_{EI}} < 500$	$500 \le \omega_{_{EI}} < 1000$	$\omega_{_{EI}} \ge 1000$	
Р	$0 \le \omega_P < 0.75$	$0.75 \le \omega_p < 1.25$	$\omega_p \ge 1.25$	





Route 4 x Route 5

CO ₂ Mass Balance in
CRADLE-TO-GATE DOMAIN

	CO ₂ mass flow (kg/h)		
Route			
	Inlet	Outlet	Sequestration
5	16,161	4,165	11,996
4	8,468	1,952	6,517

Metric	El _{max} =50,000		
	ROUTE 5	ROUTE 4	Additional Sustainability Metrics
M (kg/kg)	1.30	0.97	
E (kJ/kg)	225,114	104,128	
ec (US\$/h)	22.198	9,136	_
<i>en</i> (PEl _{out} /h)	50,000	50,000	
ε (US\$/PEI _{out})	0.44	0.18	

CONCLUDING REMARKS

- P2 was introduced as the basis for GC and GE
- Sustainability Metrics were presented for screening Process Alternatives
- A CASE STUDY was used to illustrate the Selection of the best ROUTE among process alternatives.
- An Exploratory Metric was first used to reduce the candidate processes.
- For the 2 most promising alternatives (ROUTES 4 and 5), process simulation and optimization in HYSYS was employed for SUSTAINABILITY ANALYSIS.
- CRADLE-TO-GATE and GATE-TO-GATE domains for each routes were defined.
- A sensitivity analysis was used to stress the impact of the "degree of relevance" attributed to a given metric and/or domain.
- Sustainability metrics show that ROUTE 5 has better Ecoefficiency and Material index. In contrast, ROUTE 4 has better Energy index. The EI of cradle-to-gate domain (global impact) of ROUTE 5 is around 12
- The simultaneous consideration of the two domains reveals that the choice of the better route depends on the aspects that are being prioritized. In general, profit and local impact should be given priority. If this rule is applied, then ROUTE 5 is the most sustainable.

